Supplementary information

Factors on the separation of photogenerated charges and the charge

dynamics in oxide/ZnFe₂O₄ composites

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Scanning electron microscopy and energy dispersive spectrometry

Morphologies and chemical compositions were observed using scanning electron

microscopy (SEM, JSM-6700F, JEOL) combining with energy dispersive

spectrometry (EDS, INCA-ENERGY). The SEM images of four composites

(WO₃/ZFO, α-Bi₂O₃/ZFO, TiO₂/ZFO and SnO₂/ZFO with the molar ratio 4:1) are

shown in Fig. SI1 (a)-(d), respectively. From the SEM images, four composites all

take on the phenomenon of aggregation and agglomerate to bulks. It is also shown that

the particle sizes of composites from SEM are similar to TEM.

Fig. SI1 (e)-(h) illustrate the energy spectra of four composites (WO₃/ZFO,

α-Bi₂O₃/ZFO, TiO₂/ZFO and SnO₂/ZFO with the molar ratio 4:1), respectively. Each

energy spectrum includes the elements of their compositions and the atomic percents

are almost consistent with the nominal molar ratios. The Table SI1 shows the atomic

percents of four composites from their energy spectra and the corresponding

composition ratios are about W:Zn:Fe:O=4:1:2:16, Bi:Zn:Fe:O=4:1:2:10,

Ti:Zn:Fe:O=4:1:2:12 and Sn:Zn:Fe:O=4:1:2:12, respectively.

Supplementary Tables

Table SI1 The atomic percents of four composites (WO₃/ZFO, α -Bi₂O₃/ZFO, TiO₂/ZFO and SnO₂/ZFO) with the molar ratio 4:1 of (W, Bi, Ti and Sn)/Zn from their energy dispersive spectra.

material	W	O_{3}	/ZI	FO (α-Β :	i ₂ O	₃ /Z	FO	Т	iO ₂	/ZI	O	Sr	O_2	/ZF	O
element	W	Zn	Fe	О	Bi	Zn	Fe	О	Ti	Zn	Fe	О	Sn	Zn	Fe	О
atomic percent	17.41	4.36	8.66	69.56	23.57	7 5.84	11.72	2 58.86	21.08	3 5.23	10.51	63.18	21.1	1 5.18	10.48	3 63.22

Table SI2 The average particle sizes of WO₃/ZFO, α -Bi₂O₃/ZFO, TiO₂/ZFO and SnO₂/ZFO with four molar ratios.

Material	WO ₃ /ZFO	α-Bi ₂ O ₃ /ZFO	TiO ₂ /ZFO	SnO ₂ /ZFO	
molar ratio	a	verage particle	e sizes (nm)		
2:1	198	395	201	306	
4:1	21	202	83	218	
6:1	165	343	172	372	
8:1	200	238	186	524	

Table SI3 The surface areas of WO₃/ZFO, α -Bi₂O₃/ZFO, TiO₂/ZFO and SnO₂/ZFO with four molar ratios.

Material	WO ₃ /ZFO	α-Bi ₂ O ₃ /ZFO	TiO ₂ /ZFO	SnO ₂ /ZFO
molar ratio		Surface area (n	n^2/g)	
2:1	2.58	39.9	5.72	2.97
4:1	23.6	78.2	13.8	4.19
6:1	3.01	46.5	6.74	2.42
8:1	2.41	66.2	6.22	1.76

Table SI4 The zeta potentials of WO₃/ZFO, α -Bi₂O₃/ZFO, TiO₂/ZFO and SnO₂/ZFO with four molar ratios.

Material	WO ₃ /ZFO	α-Bi ₂ O ₃ /ZFO	TiO ₂ /ZFO	SnO ₂ /ZFO		
molar ratio	zeta potentials (mV)					
2:1	-38	-21.4	-30.6	10.7		
4:1	-23.8	22.4	-21.4	33.3		
6:1	-36.7	-10.2	-24.8	-16.6		
8:1	-38.1	-21.6	-26.9	8.61		

Supplementary Figures

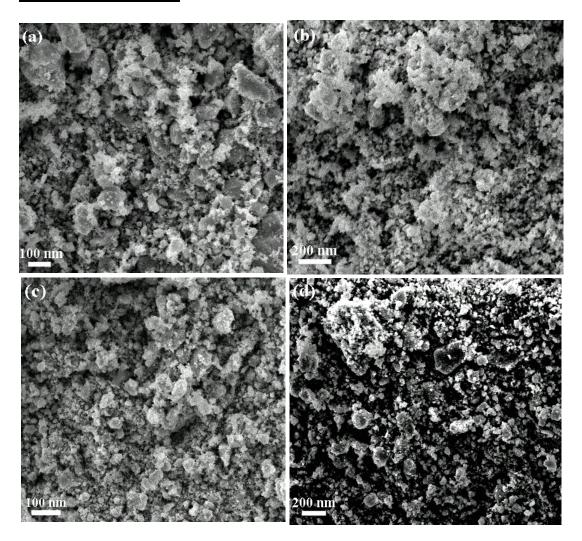


Fig. SI1 ((a), (b), (c) and (d)) SEM images of WO₃/ZFO, α -Bi₂O₃/ZFO, TiO₂/ZFO and SnO₂/ZFO with the molar ratio 4:1, respectively.

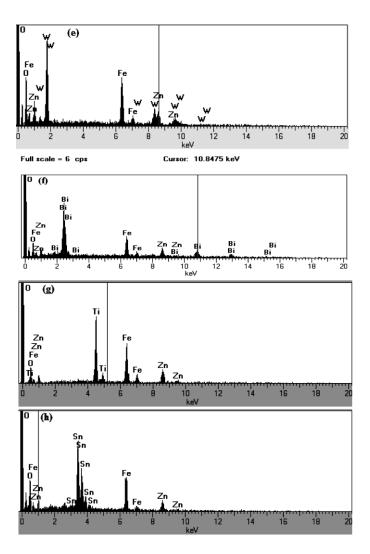


Fig. SI1 ((e), (f), (g) and (h)) Energy dispersive spectrometry (EDS) of WO₃/ZFO, α -Bi₂O₃/ZFO, TiO₂/ZFO and SnO₂/ZFO with the molar ratio 4:1, respectively.

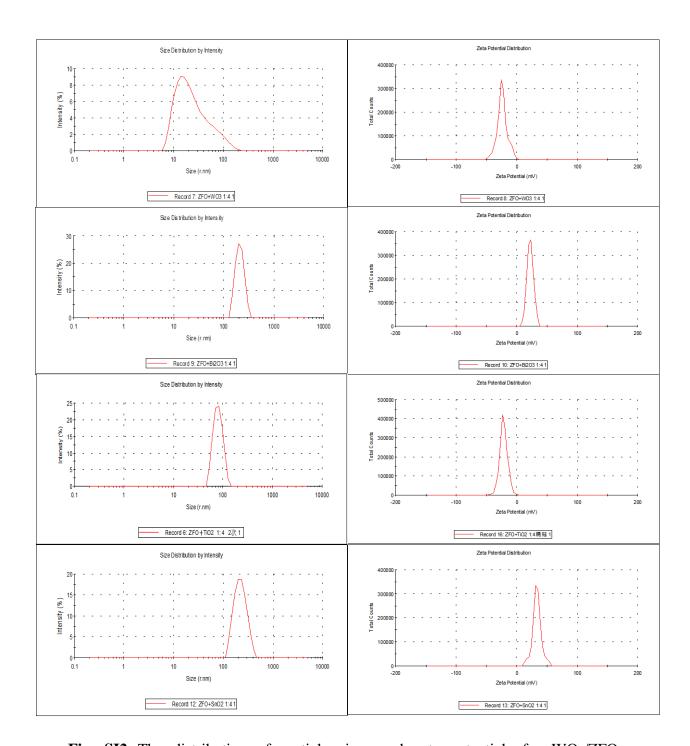


Fig. SI2 The distribution of particle sizes and zeta potentials for WO₃/ZFO, $\alpha\text{-Bi}_2\text{O}_3\text{/ZFO}, \text{TiO}_2\text{/ZFO} \text{ and } \text{SnO}_2\text{/ZFO} \text{ with molar ratio 4:1.}$

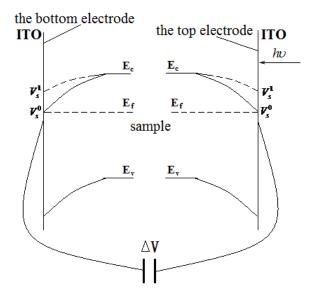


Fig. SI3 The schematic diagram of sandwich structure consisting of ITO (indium tin oxide) and material in the steady-state SPS, TPV and EFISPS. (E_c : the bottom of conduction band; E_v : the top of valence band; E_f : the Fermi energy level; ΔV : the difference of different surface potential; V_s^0 : the surface potential before illumination; V_s^1 : the surface potential after illumination; V_s^0 , V_s^1 <0; hv: the incident photon energy).