

Electronic supplementary information

Aqueous Synthesis of Glutathione-capped Cu^+ and Ag^+ -doped $\text{Zn}_x\text{Cd}_{1-x}\text{S}$ Quantum Dots with Full Color Emission†

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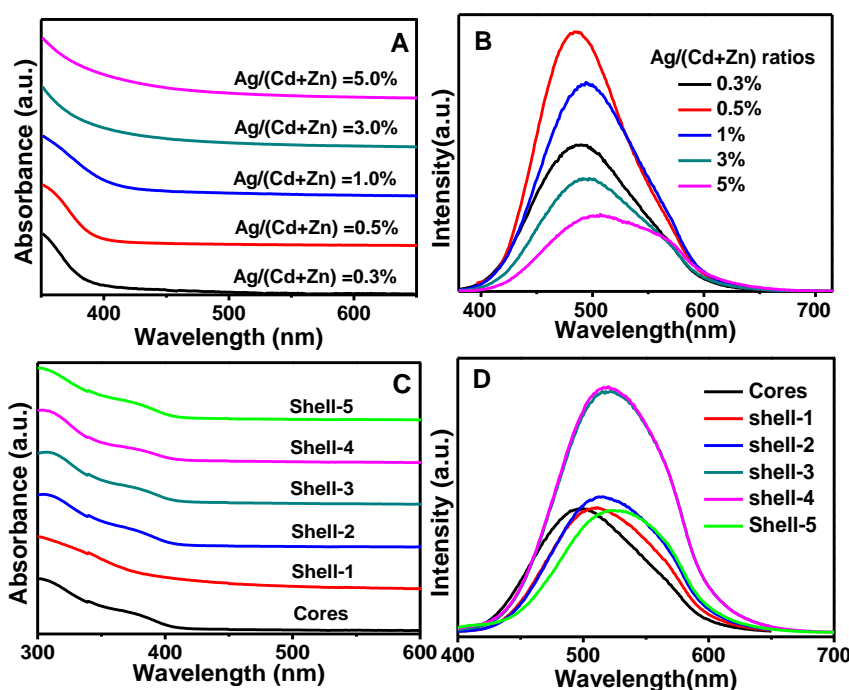


Figure S1. UV-vis absorption (A) and PL emission (B) spectra of Ag-doped $\text{Zn}_x\text{Cd}_{1-x}\text{S}$ core QDs synthesized under different Ag/(Cd+Zn) ratios (0.3-5%) with fixation of growth time of 30 min; Evolution of UV-vis absorption (C) and PL emission (D) spectra of the Ag-doped $\text{Zn}_x\text{Cd}_{1-x}\text{S}/\text{ZnS}$ core/shell QDs with deposition of different amount of ZnS shell around the core NCs.

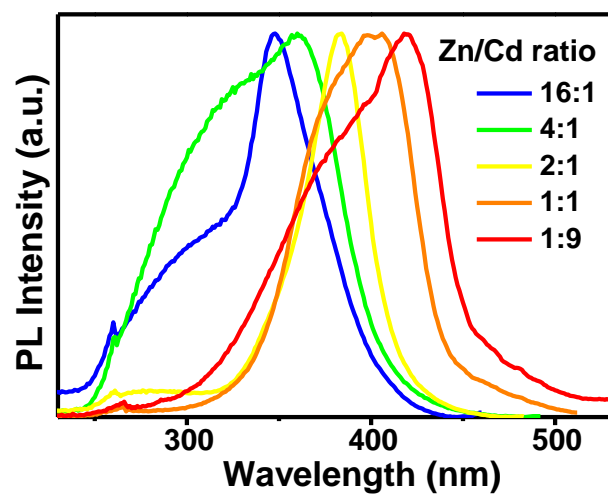


Figure S2. The excitation spectra of the Cu-doped Zn_xCd_{1-x}S QDs with different Zn/Cd ratios.

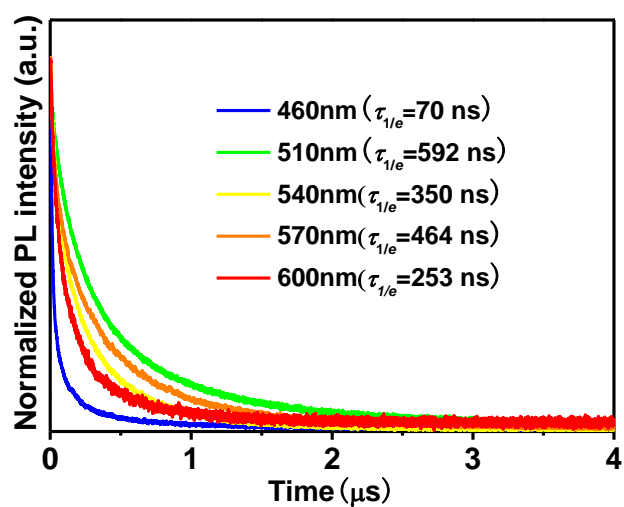


Figure S3. PL decay curves of Cu-doped $Zn_xCd_{1-x}S$ d-dots with representative emission wavelengths.

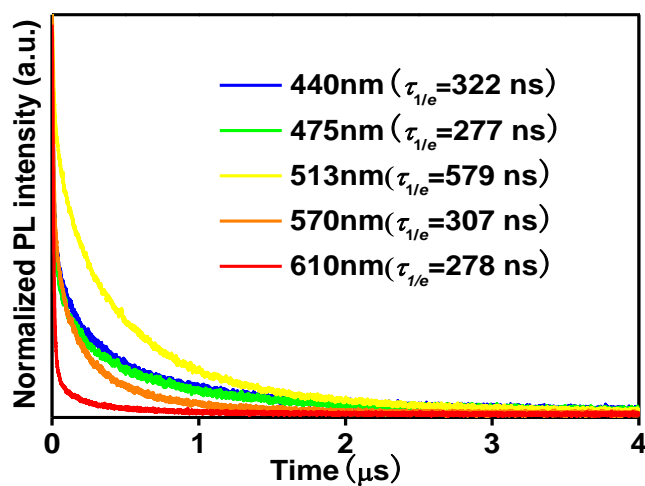


Figure S4. PL decay curves of Ag-doped $Zn_xCd_{1-x}S$ core QDs with representative emission wavelengths.

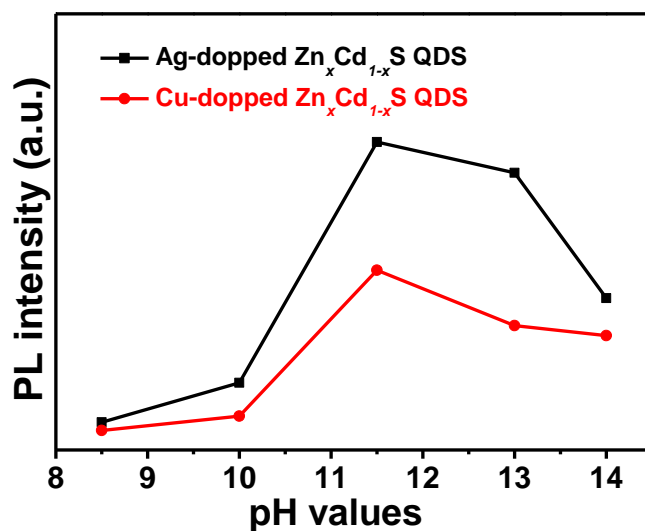


Figure S5. The PL intensity curves of Cu-doped and Ag-doped Zn_xCd_{1-x}S/ZnS core/shell QDs synthesized under different pH values.

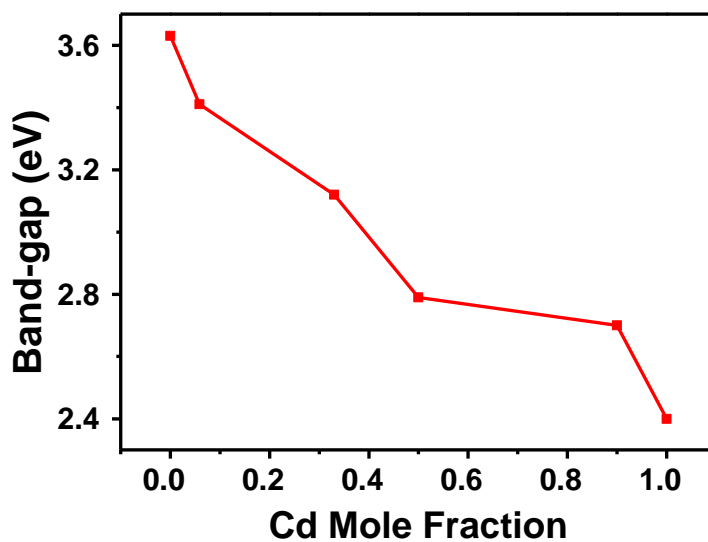


Figure S6. The band gap of alloyed Zn_xCd_{1-x}S QDs as a function of Cd mole fraction.