

Supporting Information

White-Light Electrofluorescence Switching from Electrochemically Convertible Yellow and Blue Fluorescent Conjugated Polymers

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Figure S1. Fluorescence intensity changes ($\lambda_{\text{exc}} = 360 \text{ nm}$) of the EFD of Blend-1.5 at different applied potentials. Each spectrum was obtained after applying the target potential for 30 s to obtain the fluorescence spectra at saturated state. Inset: Photographs are the image of the fluorescence changes of the EFDs at indicated potentials under UV excitation (365 nm).

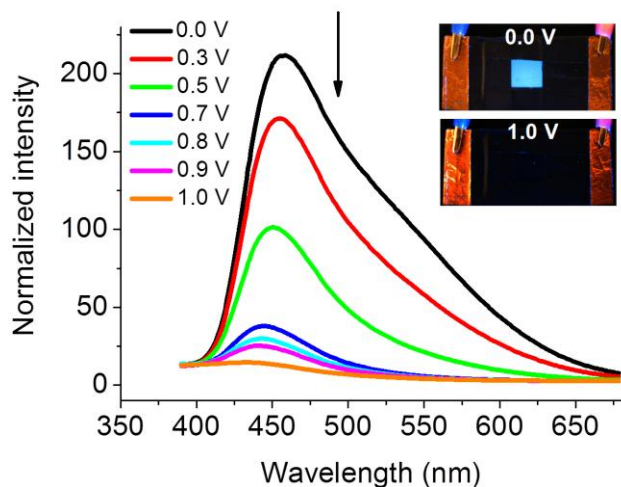


Figure S2. GPC of P0 and P1.

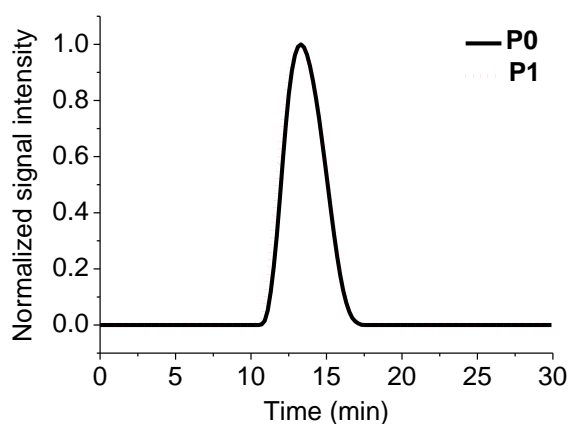


Figure S3. (a) TGA and (b) DSC thermograms of P0 and P1.

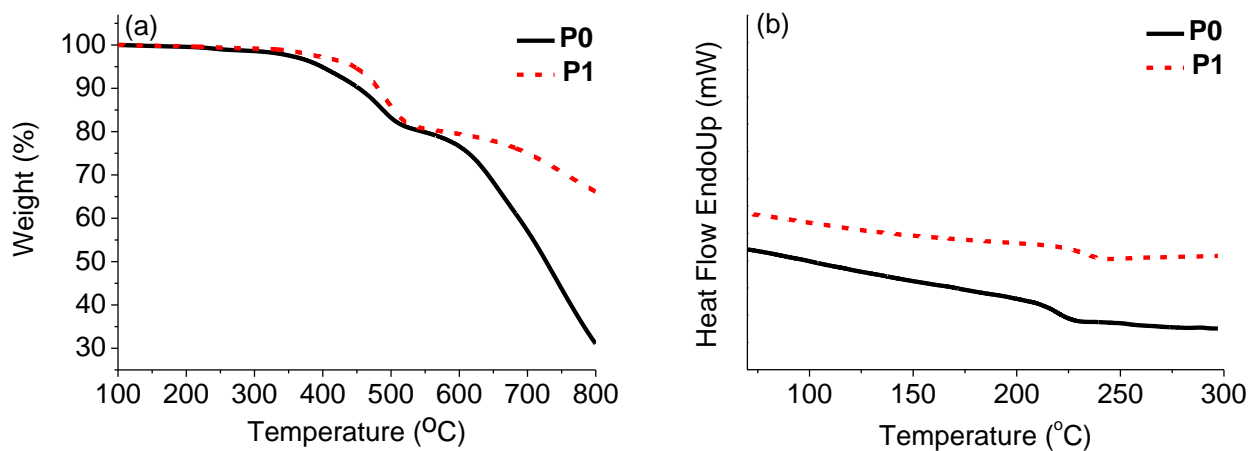


Figure S4. ^1H NMR Spectrum of **P0**

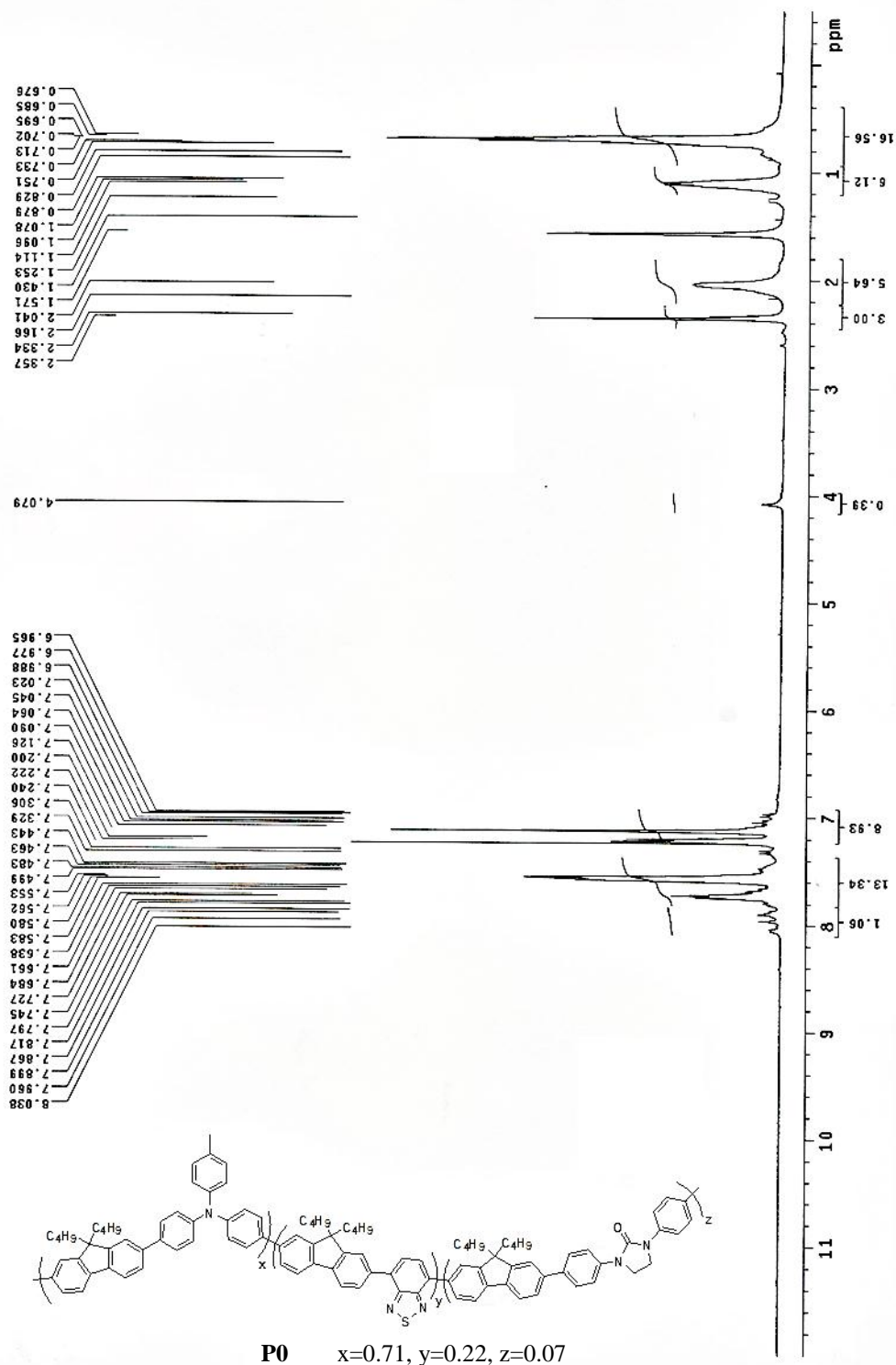


Figure S5. ^{13}C NMR Spectrum of **P0**

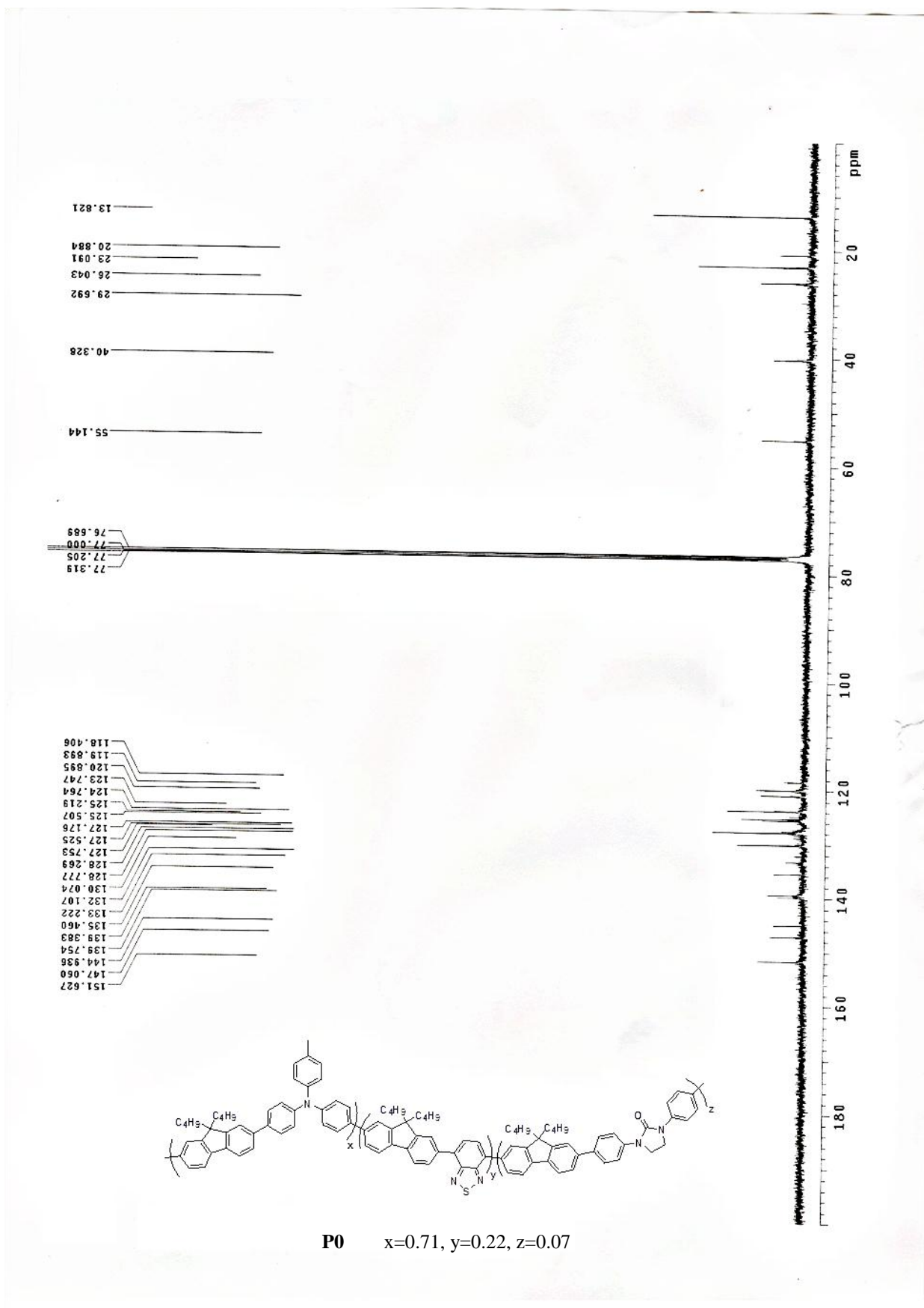


Figure S6. ^1H NMR Spectrum of **P1**

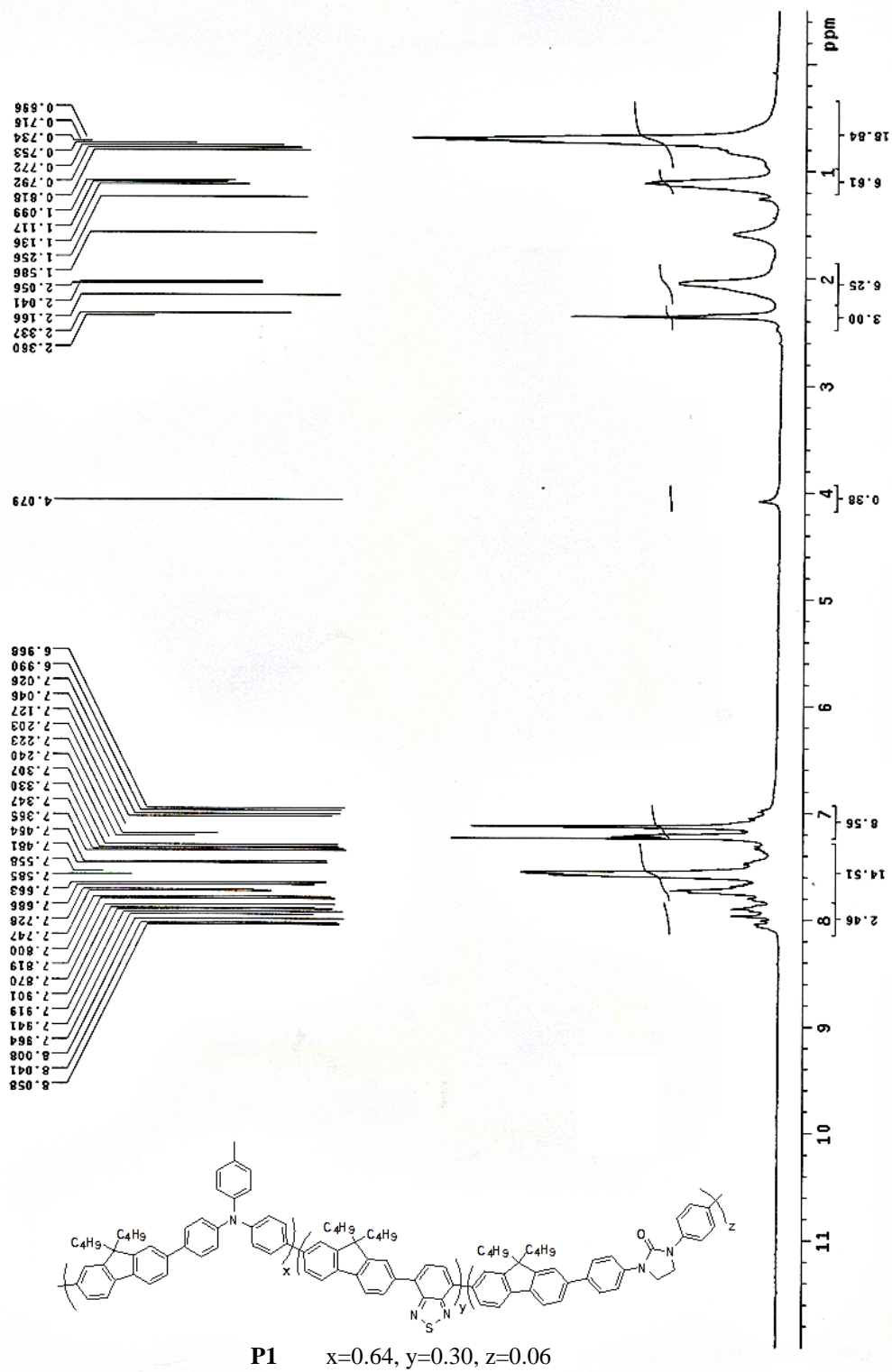
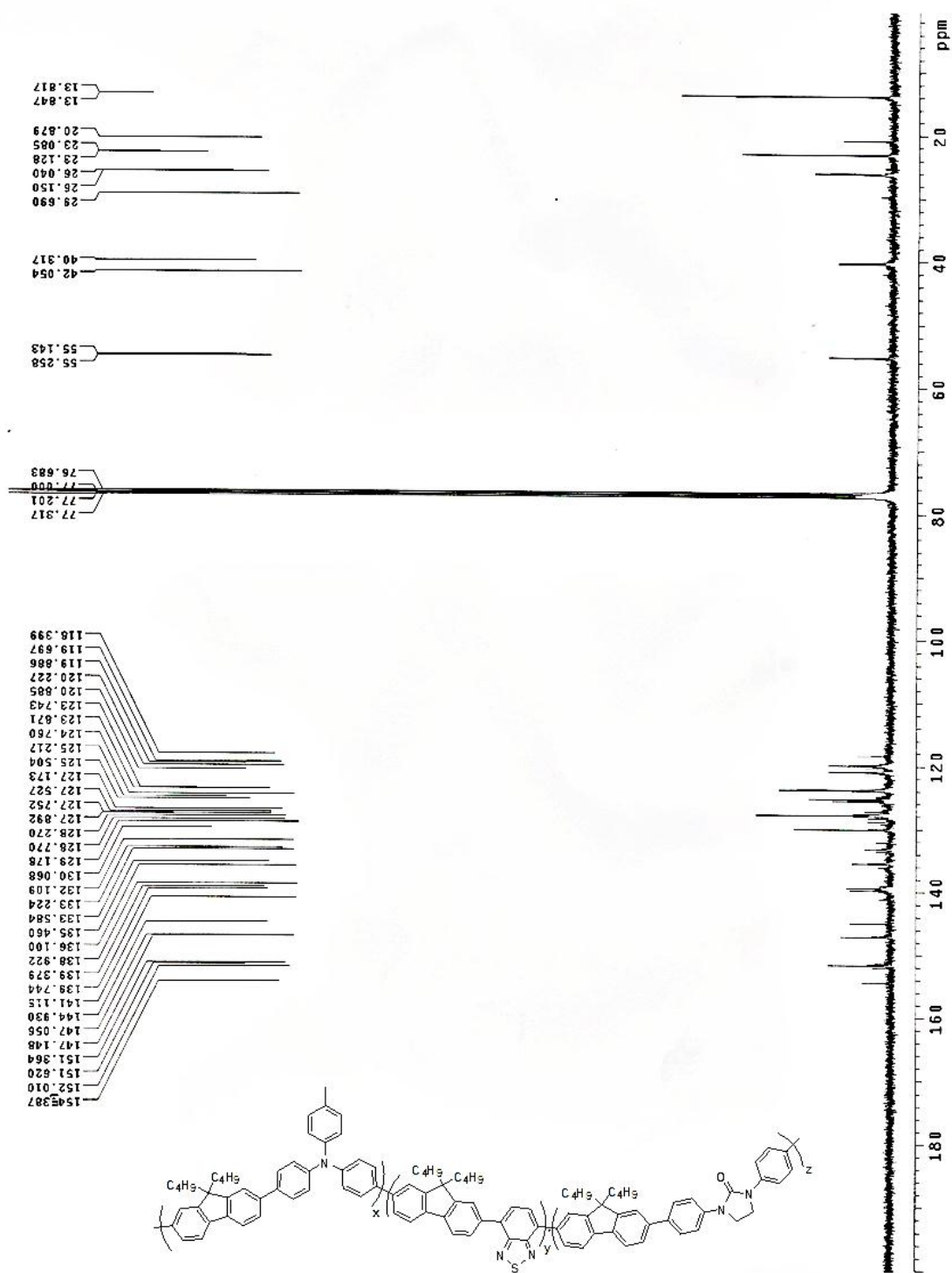


Figure S7. ^{13}C NMR Spectrum of **P1**



P1 $x=0.64, y=0.30, z=0.06$

Figure S8. ^1H NMR Spectrum of **P2**

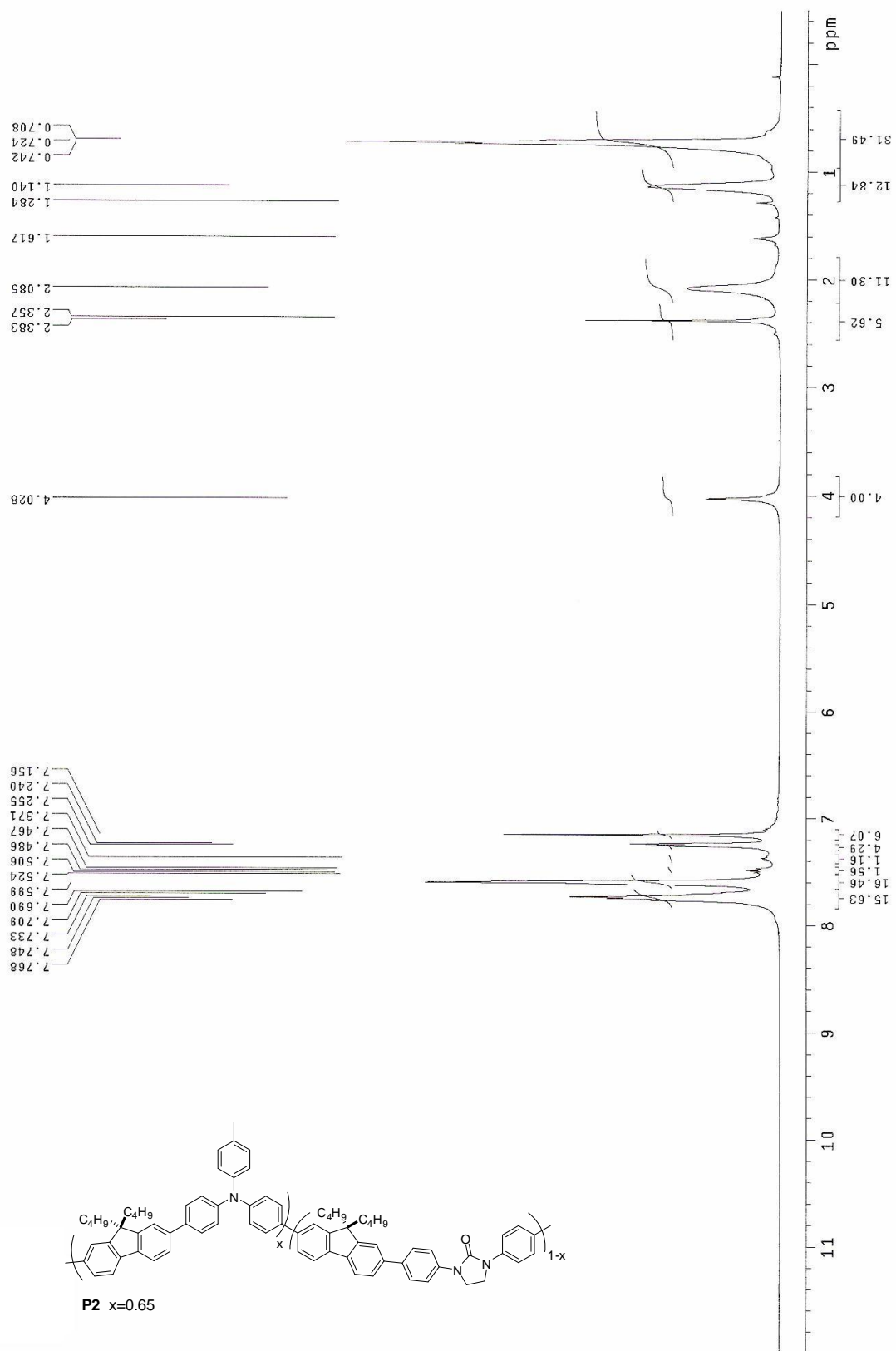


Figure S9. ^{13}C NMR Spectrum of **P2**

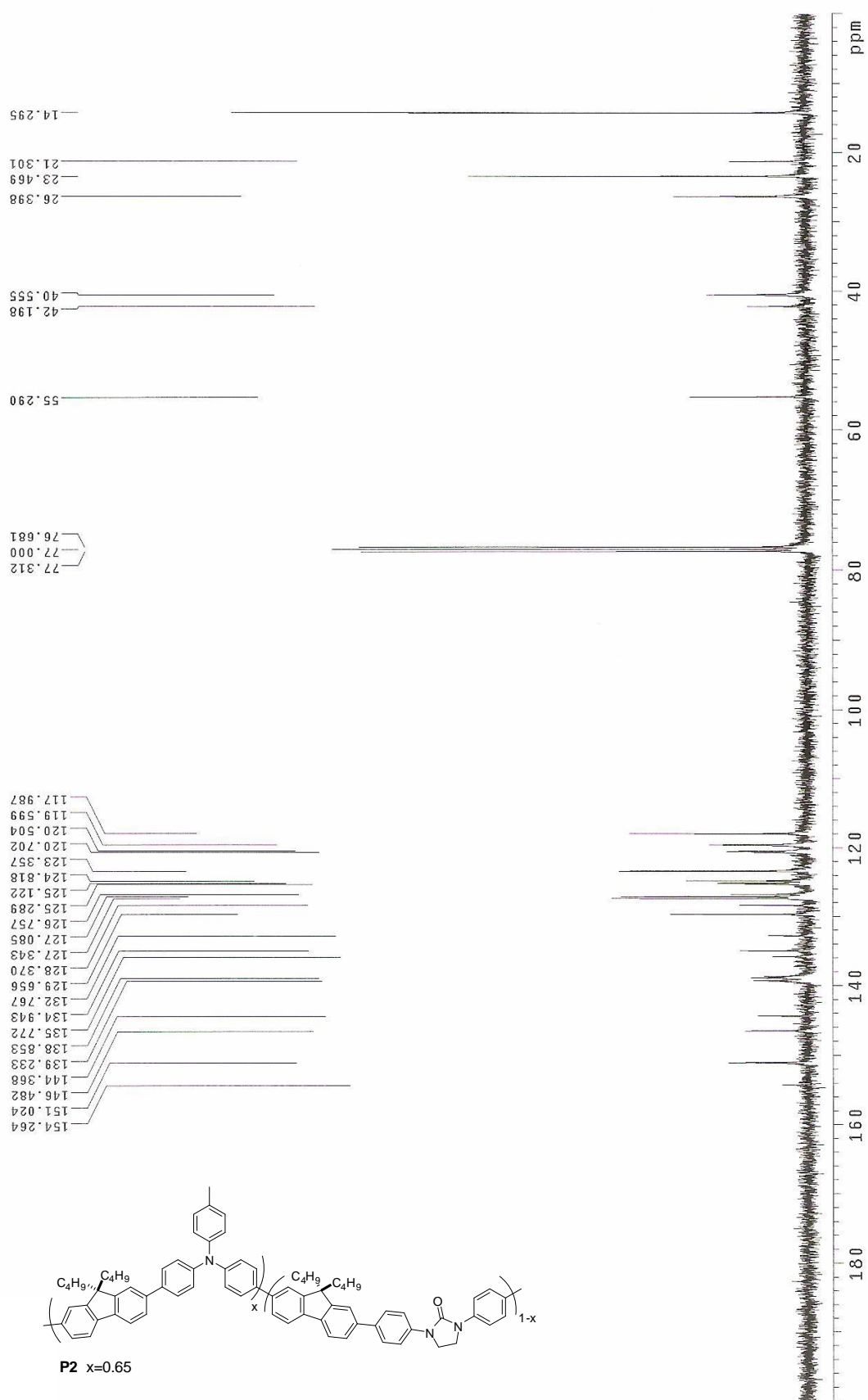


Figure S10. ^1H NMR spectra of (a) **P0** and (b) **P1**.

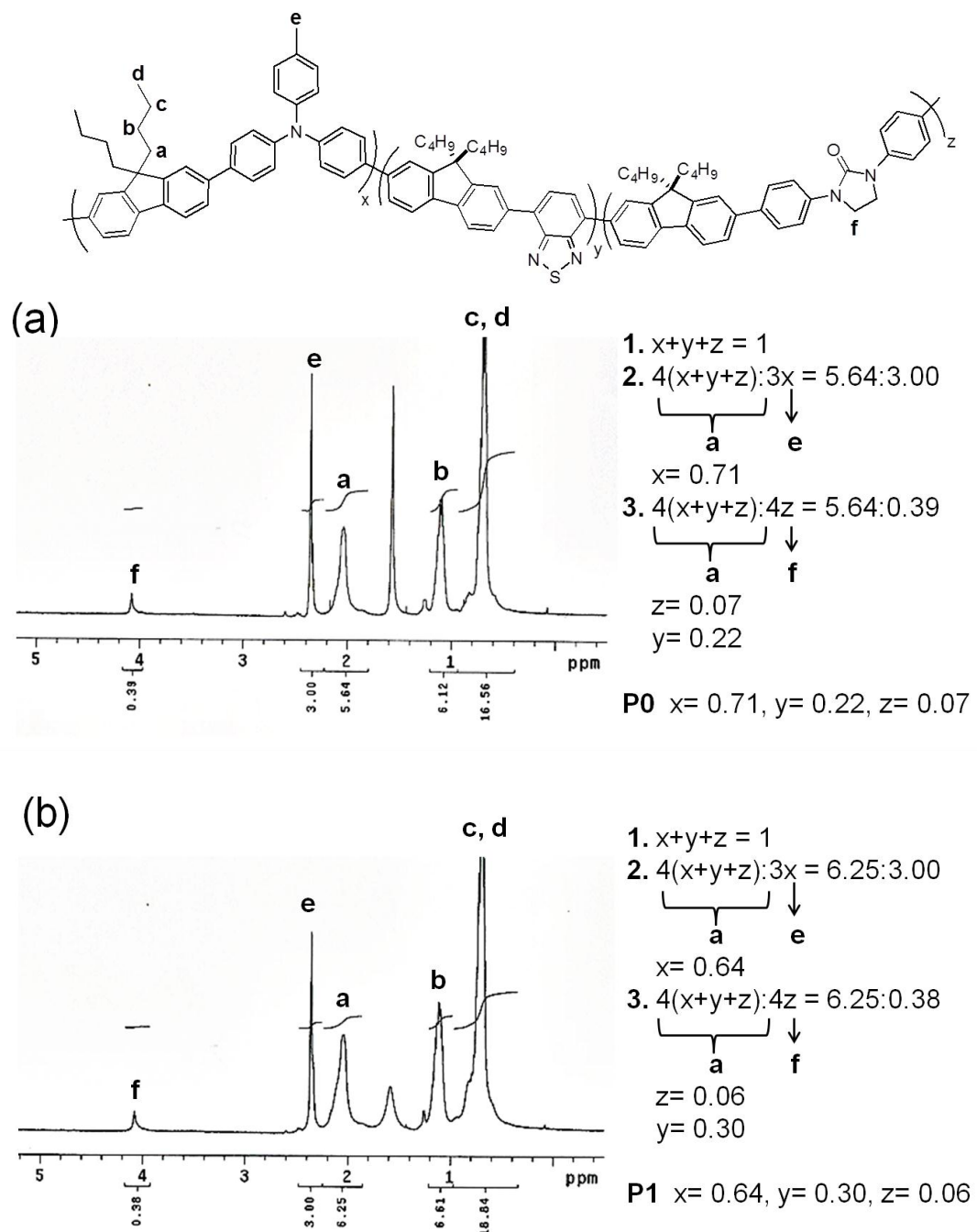


Figure S11. ^1H NMR spectra of **P2**.

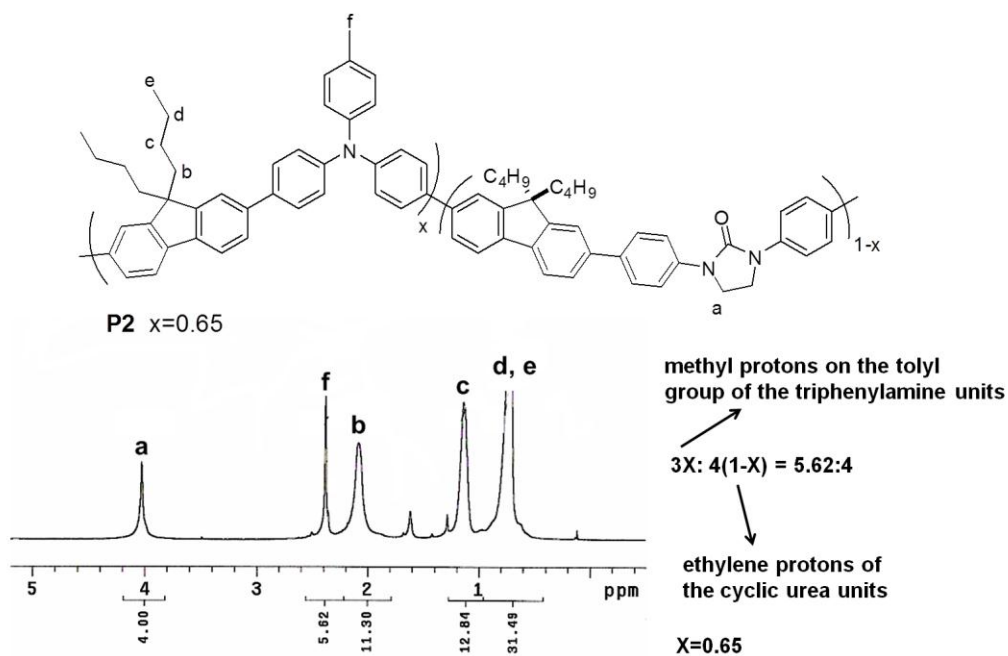


Figure S12. Fluorescence intensity changes ($\lambda_{\text{exc}} = 360$ nm) of the EFD of **P2** at different applied potentials. Each spectrum was obtained after applying the target potential for 30 s to obtain the fluorescence spectra at saturated state.

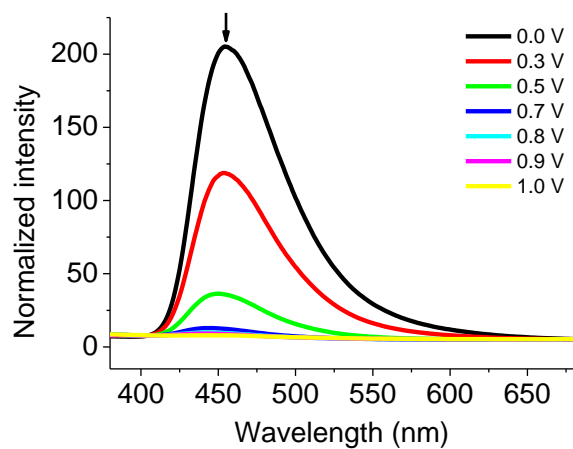


Figure S13. DSC of **P1** and Blend.

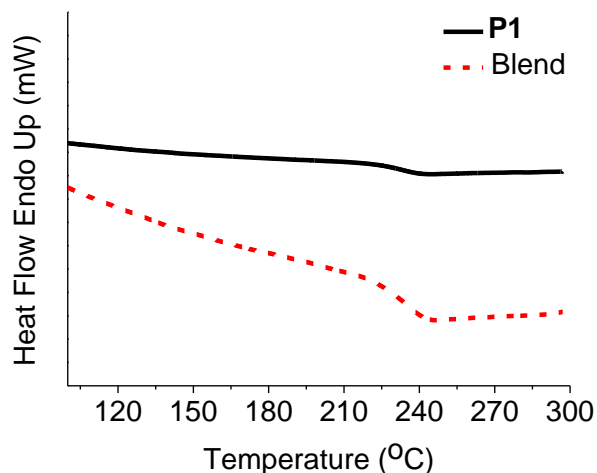


Figure S14. Fluorescence switching responses of the EFD of (a) **P1** monitored at 560 nm ($\lambda_{\text{exc}}=370$ nm) and (b) Blend-2 monitored at 445 nm ($\lambda_{\text{exc}}=360$ nm). Fixed potentials at 1.0 V (for oxidation) and -1.0 V (for reduction) were used with a step duration time for 5 second at each potential.

