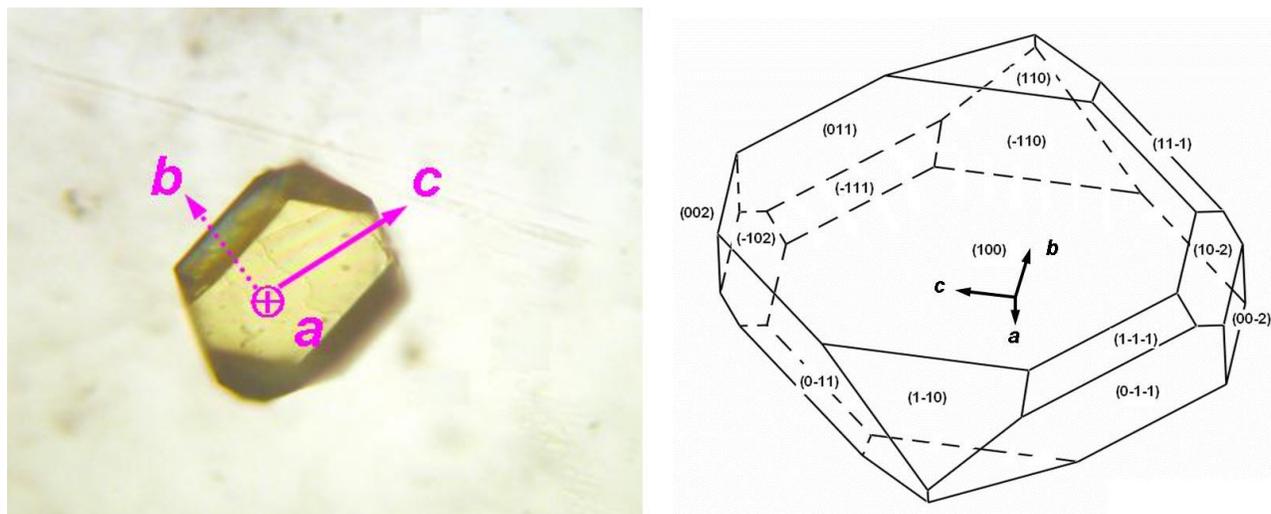
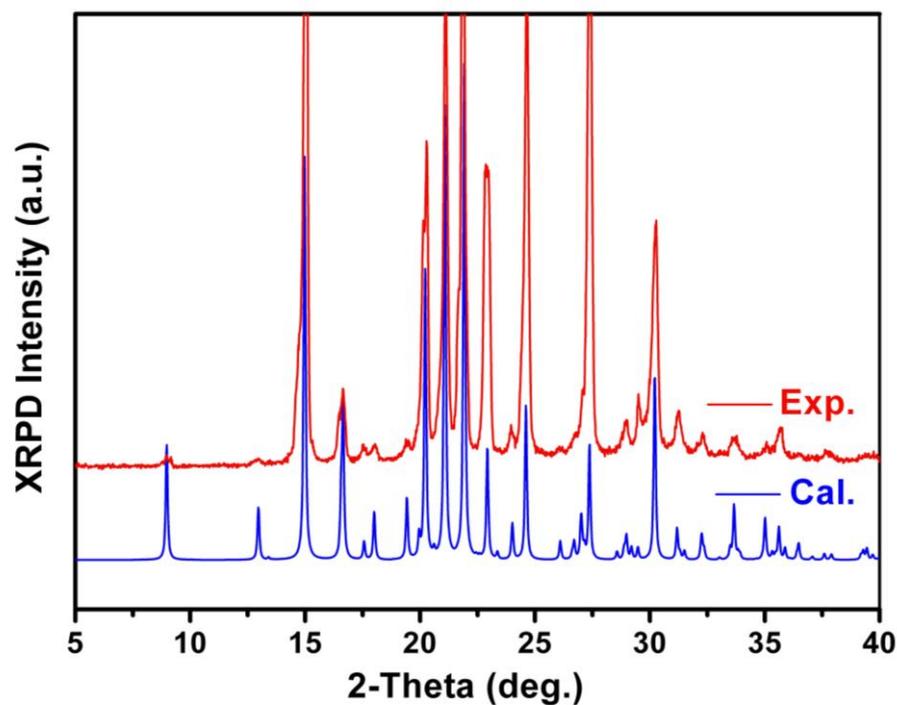


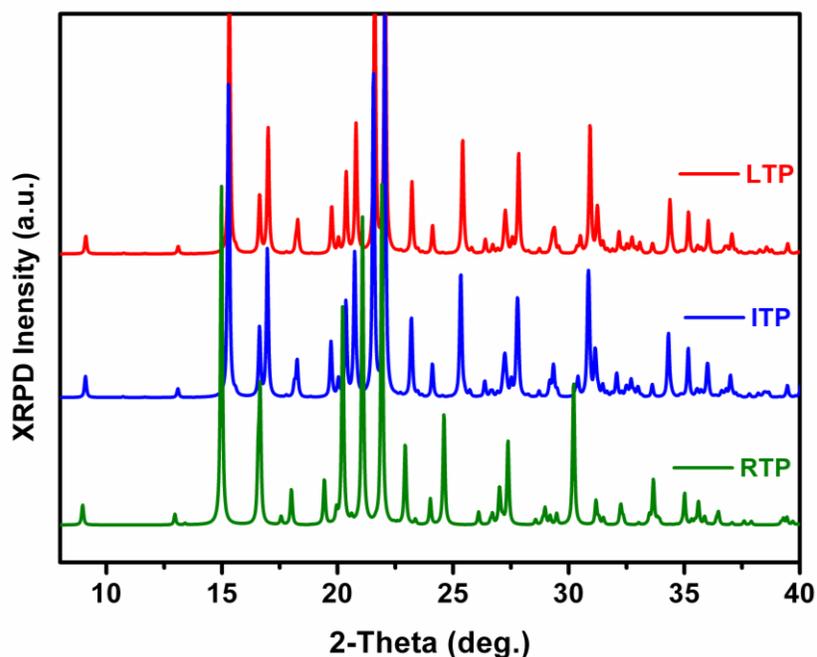
## Supporting information



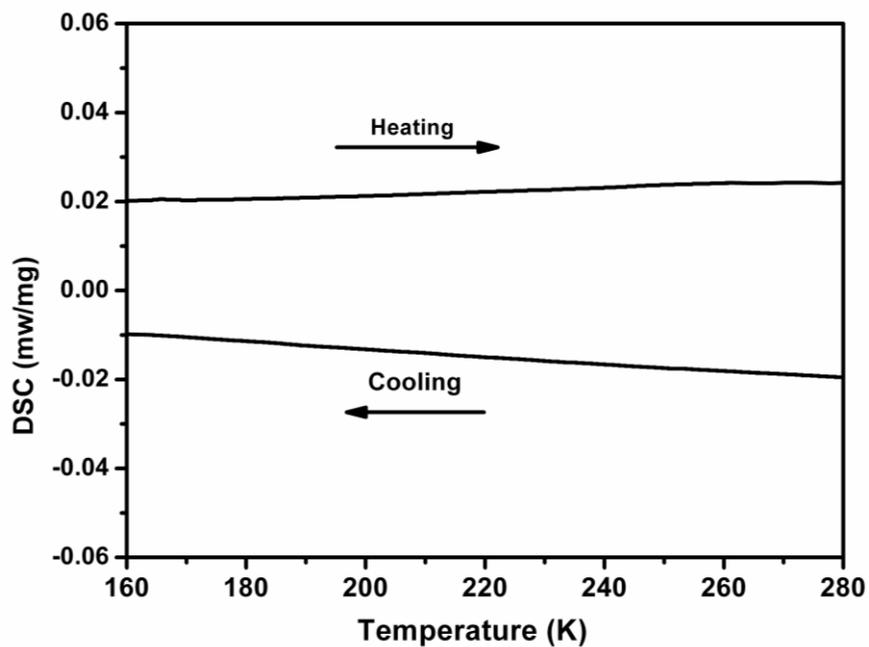
**Figure S1.** The obtained yellow crystal of **1** (left) and its growth habit determined by the X-ray powder diffraction (right).



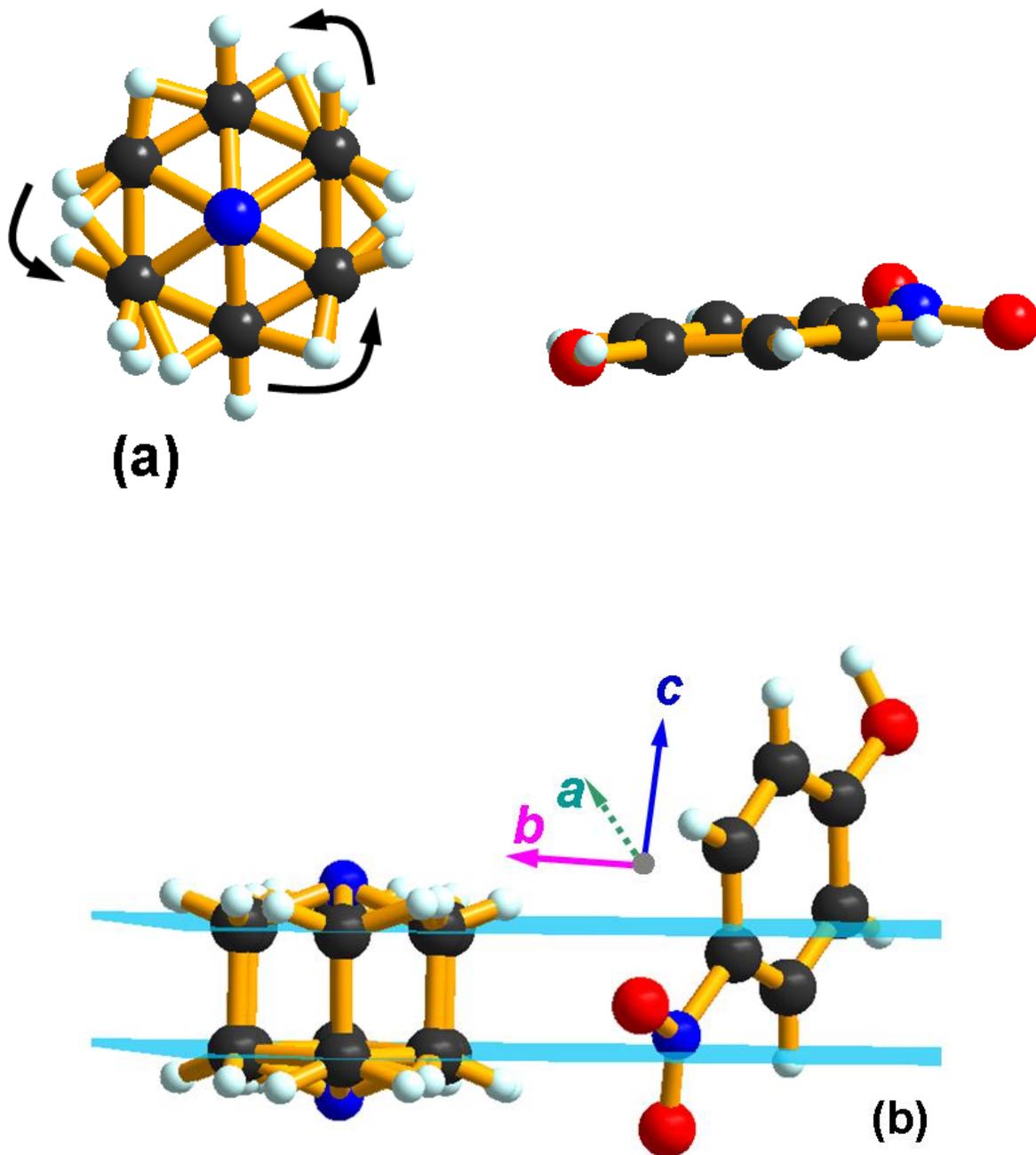
**Figure S2.** Calculated and experimental XRPD patterns on the polycrystalline samples of **1** at the room temperature, which fairly well confirms the purities of the obtained sample.



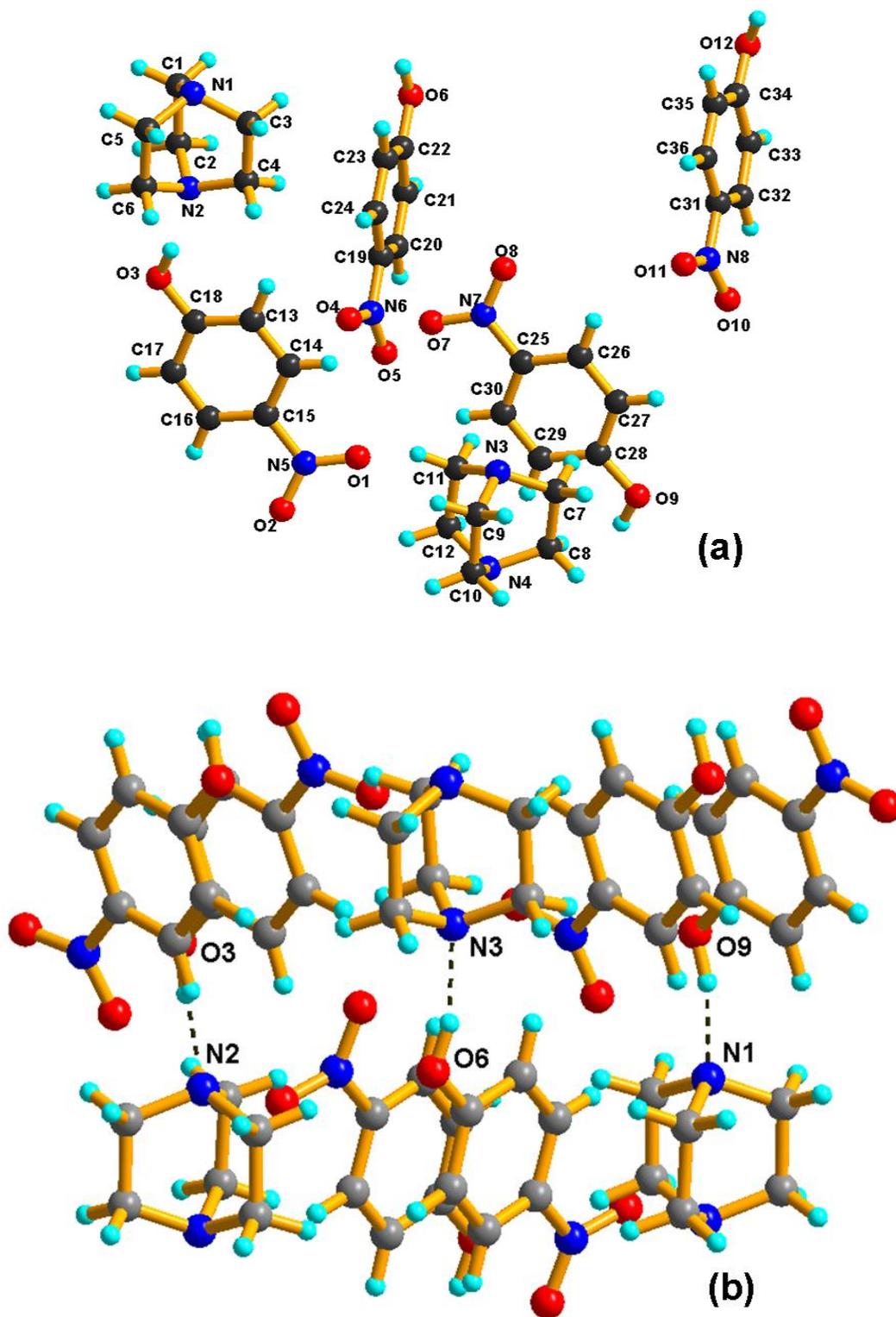
**Figure S3.** Comparison of the variable-temperature X-ray powder diffraction patterns of **1** on the basis of structure determination. The results indicate that spectra in LTP and ITP are well coincident while the patterns in RTP change obviously, explicitly confirming the phase transition of **1**.



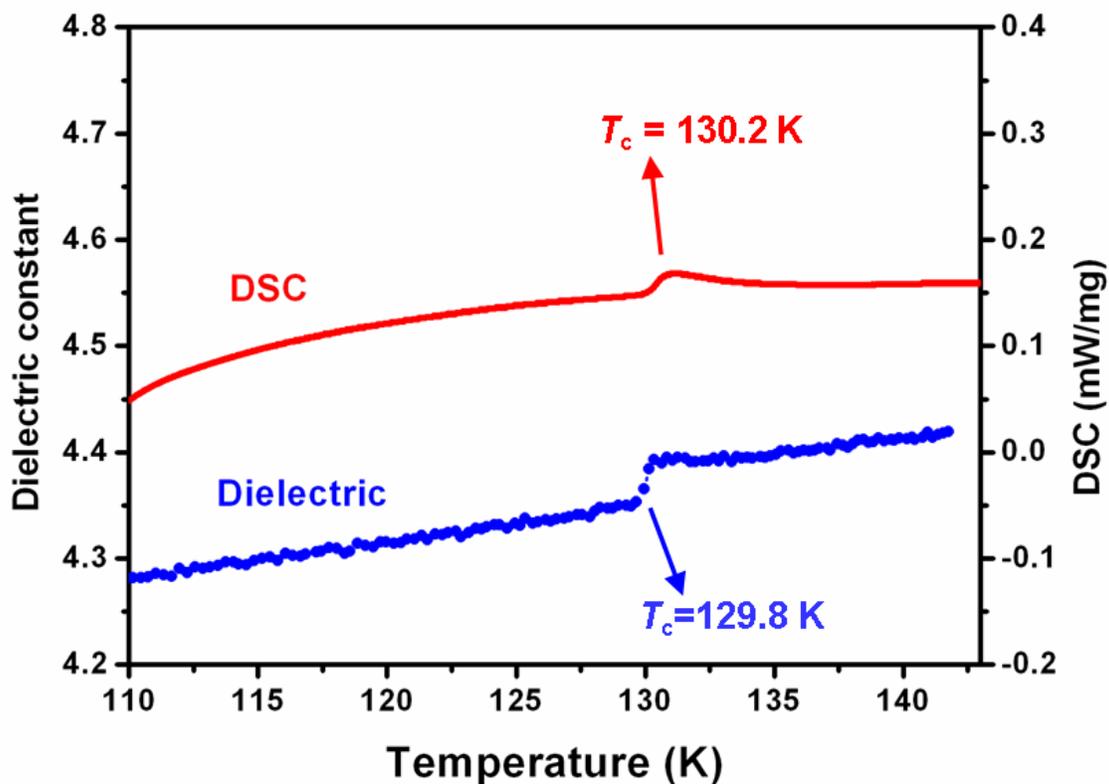
**Figure S4.** DSC traces of **1** measured in the temperature range of 160-290 K.



**Figure S5.** (a) Schematic illustration for the rotational motions of room-temperature dabco molecule in **1** viewed along the N-N axis direction. All the bonded-hydrogens are split owing to the disordings of rotational motions of dabco; (b) Corresponding plane through carbon atoms in dabco. The rotational N-N axis lies along the direction of the normal for the current coplane, which is almost parallel to the crystallographic *c*-axis.



**Figure S6.** (a) Molecule structure of **1** with atomic labellings at the LT phase, in which all the dabco molecules have become much more ordered; (b) Several strong N-H...O H-bonds are shown as the black dashed lines between N atoms of the dabco and O atoms of *p*-nitrophenol moieties.



**Figure S7.** DSC and dielectric traces measured on the polycrystalline powder of the deuterated analog of **1** in the heating mode. The observed anomalies confirm the existence of phase transition. However, phase-transition temperature is still located around 130 K and the absence of deuterium isotope effect exclude the possible motions of proton.

**Table S1.** Crystal data collection and structure refinement for **1**.

	RTP, 293 K	ITP, 120 K	LTP, 100 K
Empirical formula	C <sub>18</sub> H <sub>22</sub> N <sub>4</sub> O <sub>6</sub>	C <sub>18</sub> H <sub>22</sub> N <sub>4</sub> O <sub>6</sub>	C <sub>18</sub> H <sub>22</sub> N <sub>4</sub> O <sub>6</sub>
Formula weight	390.40	390.40	390.40
Crystal system, space group	Monoclinic, <i>P2<sub>1</sub>/c</i>	Triclinic, <i>P</i> $\bar{1}$	Triclinic, <i>P</i> $\bar{1}$
Unit cell dimensions	<i>a</i> = 9.984(6) Å <i>b</i> = 8.890(5) Å <i>c</i> = 10.791(7) Å $\beta$ = 99.593(9)°	<i>a</i> = 9.805(5) Å <i>b</i> = 13.762(6) Å <i>c</i> = 13.774(7) Å $\alpha$ = 80.039(13)° $\beta$ = 83.930(9)° $\gamma$ = 84.024(11)°	<i>a</i> = 9.785(2) Å <i>b</i> = 13.7474(18) Å <i>c</i> = 13.7474(18) Å $\alpha$ = 80.157(14)° $\beta$ = 84.13(2)° $\gamma$ = 84.13(2)°
Volume	944.4(10) Å <sup>3</sup>	1813.3(15) Å <sup>3</sup>	1805.7(5) Å <sup>3</sup>
Z, Calculated density	2, 1.373 Mg/m <sup>3</sup>	4, 1.430 Mg/m <sup>3</sup>	4, 1.436 Mg/m <sup>3</sup>
Absorption coefficient	0.105 mm <sup>-1</sup>	0.109 mm <sup>-1</sup>	0.109 mm <sup>-1</sup>
<i>F</i> (000)	412	824	824
Theta range for data collection	4.04° - 25.00°	2.30° - 27.39°	1.51° - 23.99°
Limiting indices	-11 ≤ <i>h</i> ≤ 11, -10 ≤ <i>k</i> ≤ 10, -12 ≤ <i>l</i> ≤ 12	-12 ≤ <i>h</i> ≤ 12, -16 ≤ <i>k</i> ≤ 17, -17 ≤ <i>l</i> ≤ 17	-11 ≤ <i>h</i> ≤ 11, -15 ≤ <i>k</i> ≤ 13, -15 ≤ <i>l</i> ≤ 12
Reflections collected / unique	5754 / 1604 <i>R</i> <sub>int</sub> = 0.0341	15084 / 7628 <i>R</i> <sub>int</sub> = 0.0917	7724 / 5263 <i>R</i> <sub>int</sub> = 0.0189
Max. and min. transmission	0.9703, 0.9574	0.9628, 0.9455	0.9710, 0.9555
Data / restraints / parameters	1604 / 0 / 154	7628 / 0 / 506	5263 / 0 / 505
GOF	1.068	1.089	1.184
Final <i>R</i> indices [ <i>I</i> > 2σ( <i>I</i> )]	<i>R</i> <sub>1</sub> = 0.0461 <i>wR</i> <sub>2</sub> = 0.1303	<i>R</i> <sub>1</sub> = 0.1768	<i>R</i> <sub>1</sub> = 0.1050 <i>wR</i> <sub>2</sub> = 0.2386
<i>R</i> indices (all data)	<i>R</i> <sub>1</sub> = 0.0569 <i>wR</i> <sub>2</sub> = 0.1390	<i>R</i> <sub>1</sub> = 0.1829	<i>R</i> <sub>1</sub> = 0.1193 <i>wR</i> <sub>2</sub> = 0.449

**Table S2.** Selected  $U_{\text{eq}}$  values for the atoms of dabco molecules at different temperatures, which is defined as one third of the trace of the orthogonalized  $U_{ij}$  tensor. The detailed  $U_{ij}$  tensor ratio for dabco molecule for **1** at RT phase is  $U_{\text{eq}}(\text{N})/U_{\text{eq}}(\text{mean of carbon atoms}) = 73.5/49$ ; the value at IT phase is 27.5/19; the value at LT phase is 24.7/18. These findings reveal the successive order-disordered feature of the phase transition.

Atoms	$U_{\text{eq}}$ , RTP	$U_{\text{eq}}$ , ITP	$U_{\text{eq}}$ , LTP
C7	68(1)	28(1)	22(1)
C8	69(1)	26(2)	24(1)
C9	80(1)	29(1)	25(2)
C10	80(1)	28(1)	27(1)
C11	69(1)	27(2)	25(2)
C12	75(1)	27(1)	25(2)
N1	-	18(1)	17(1)
N2	49(1)	20(1)	19(1)