

## Supporting Information

### High Performance Flexible Solid-State Asymmetric Supercapacitors of MnO<sub>2</sub>/ZnO Core/Shell Nanorods // Specially Reduced Graphene Oxide

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## Electrochemistry data analysis

The areal power density (P) and energy density (E) were calculated by using the following equations:

$$P = E/t \quad (1)$$

$$E = 0.5CV^2 \quad (2)$$

In these equations V stands for the applied voltage, t means the discharge time, and C means the areal capacitance from equation (3):

$$C = (\int IdV) / (SvV) \quad (3)$$

Here  $I$  means the applied current,  $V$  means potential,  $v$  is the potential scan speed,  $S$  means the area of active electrode materials.

As for the calculation of ASC device, the area (S) should be replaced by volume ( $V$ ).

For ASC, the charge balance will follow the relationship  $q^+ = q^-$ :

$$q = C \times \Delta E \times S \quad (5)$$

Here  $C$  stands for areal capacitance, and  $\Delta E$  means the potential range for the charge/discharge process of each electrode. So that in order to get  $q^+ = q^-$ , the area balancing will follow the Equation (6):

$$S^+/S^- = C^- \Delta E^- / C^+ \Delta E^+ \quad (6)$$

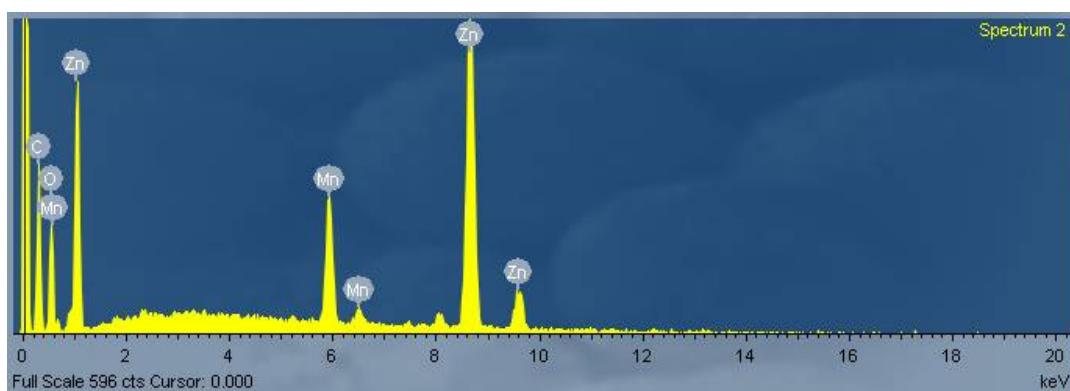


Figure S1. EDS profile of the  $\text{MnO}_2/\text{ZnO}$  core/shell nanorod array.

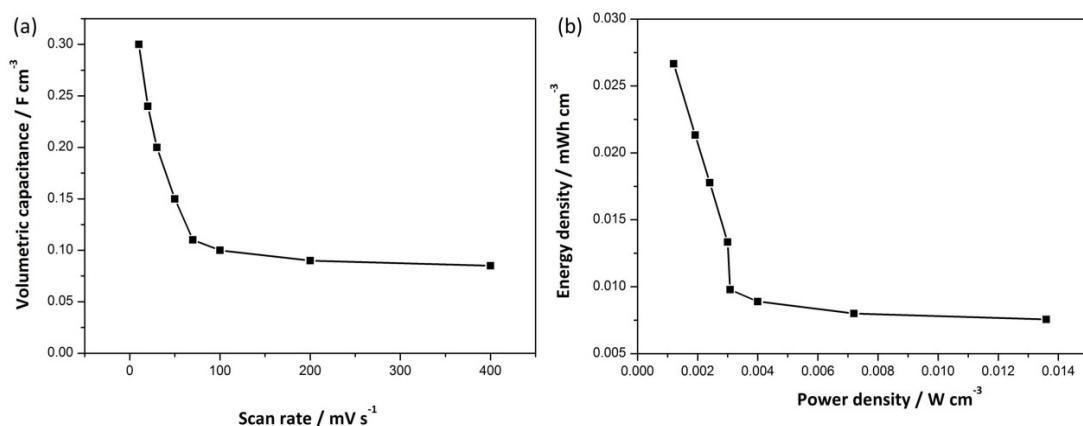


Figure S2. a) Volumetric capacitance of SSC as a function of scan rate in LiCl/PVA gel electrolyte.

b) Energy density vs. power density for SSC in LiCl/PVA solid electrolyte.

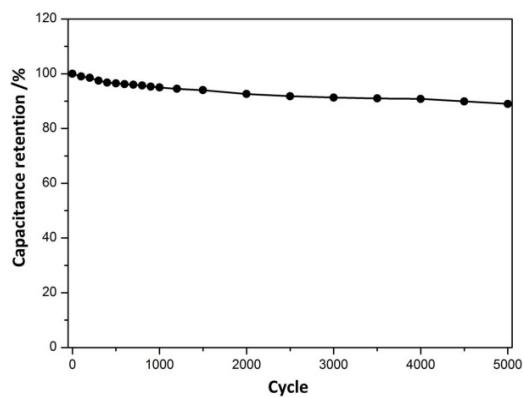


Figure S3. Cycling performance of SSC device measured at a scan rate of  $50 \text{ mV s}^{-1}$  for 5000 cycles in gel (LiCl/PVA) electrolytes.

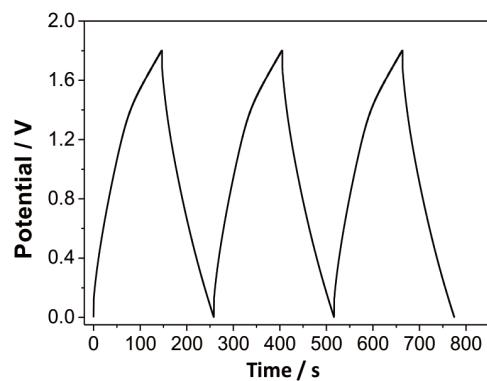


Figure S4. Galvanostatic charge and discharge curves collected at a current density of 0.5 mA/cm<sup>2</sup> for ASC device operated within a voltage window of 1.8 V.

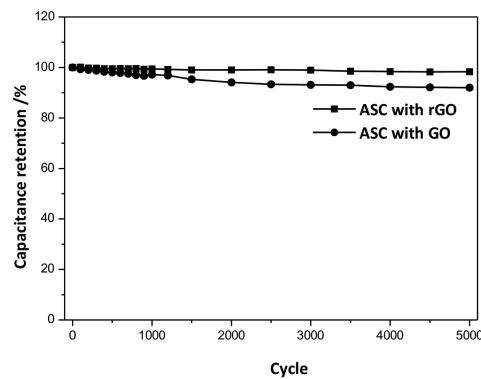


Figure S5. Capacity retention of the ASC devices at a constant charge/discharge cycling rate of 0.5 mA/cm<sup>2</sup>.