

## Supplementary Information

### **Structurally simple thienodipyrandione-containing reversible fluorescent switching piezo- and acido-chromic materials**

*Rajeswara Rao M, Chia-Wei Liao and Shih-Sheng Sun\**

#### **Content**

1. Copies of MALDI-TOF mass,  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectra
2. Absorption and emission spectra of all compounds in solution state
3. Absorption and emission spectra of all compounds in frozen state
4. Absorption and emission spectra of all compounds in solid state
5. Piezo-chromic PL spectra of **TDAn** and **TDAnPh**
6. Table 1: Photophysical data of all compounds
7. Table 2: Crystallographic data of **TDAn** and **TDAnPh** compounds

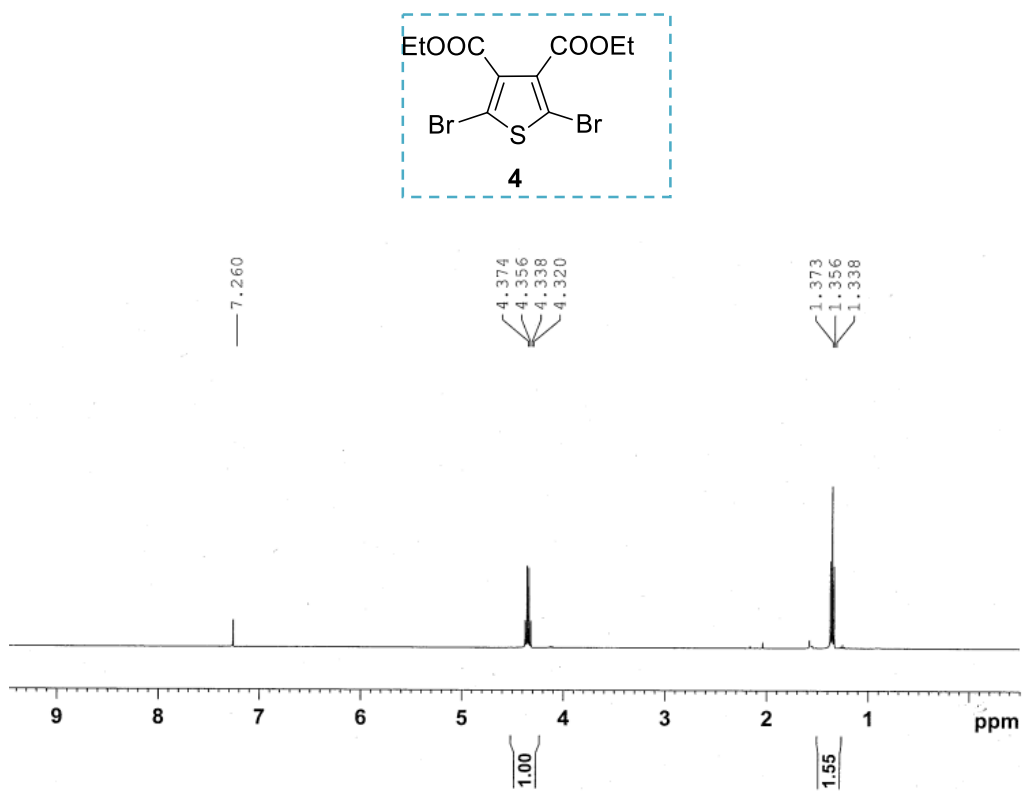


Figure S1:  $^1\text{H}$  NMR spectrum of compound **4** in  $\text{CDCl}_3$ .

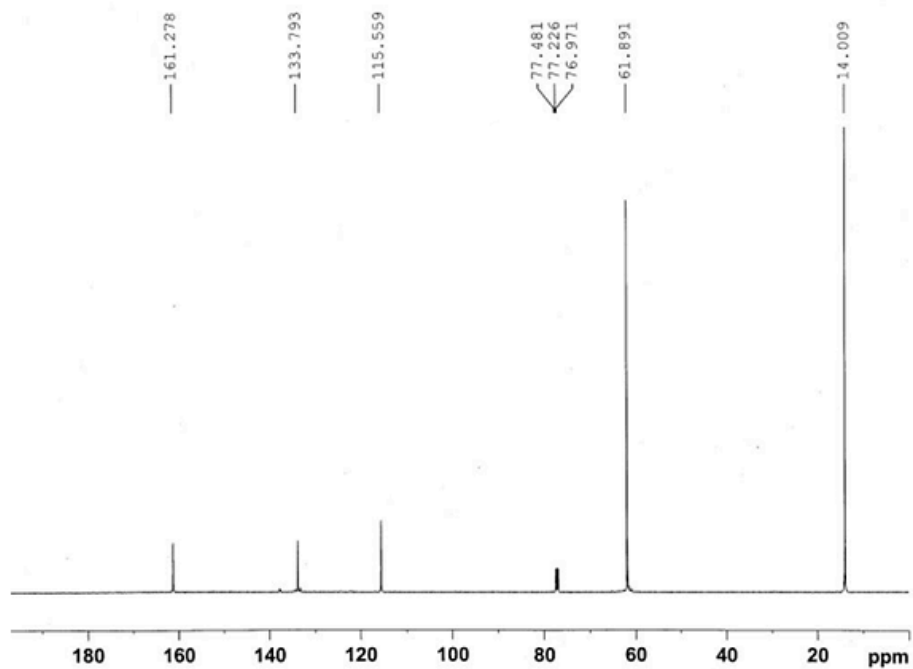
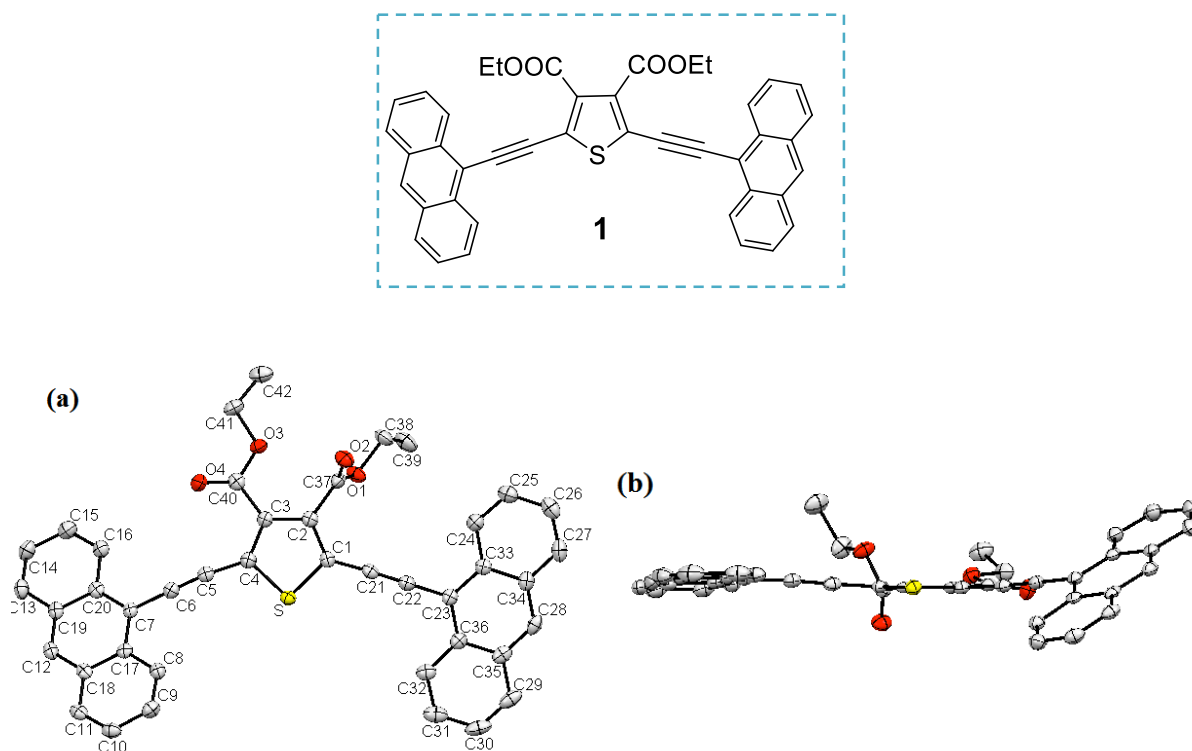
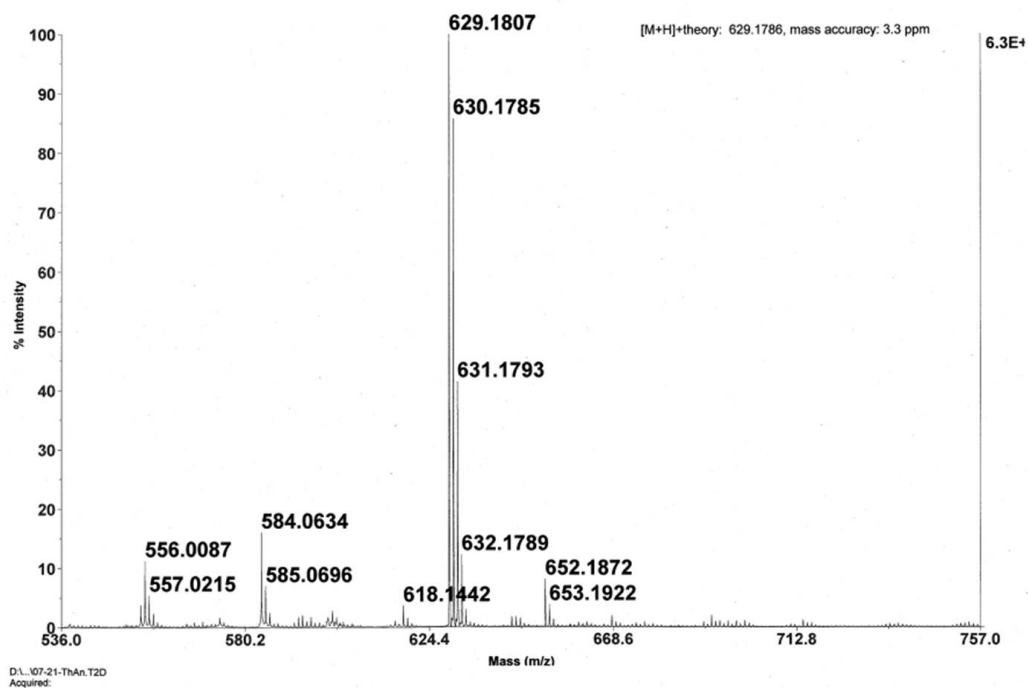


Figure S2:  $^{13}\text{C}$  NMR spectrum of compound **4** in  $\text{CDCl}_3$ .



**Figure S3:** Crystal structure of compound 1 (a) front view and (b) side view.



**Figure S4:** HR-MS mass spectrum of compound 1.

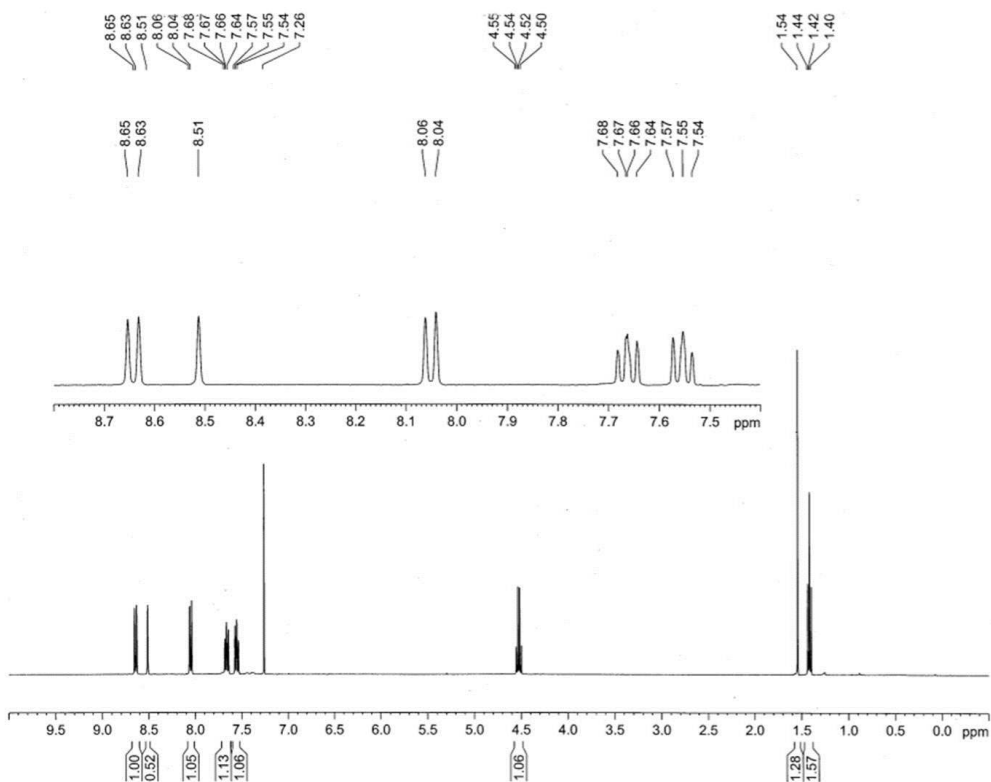


Figure S5: <sup>1</sup>H NMR spectrum of compound 1 in CDCl<sub>3</sub>.

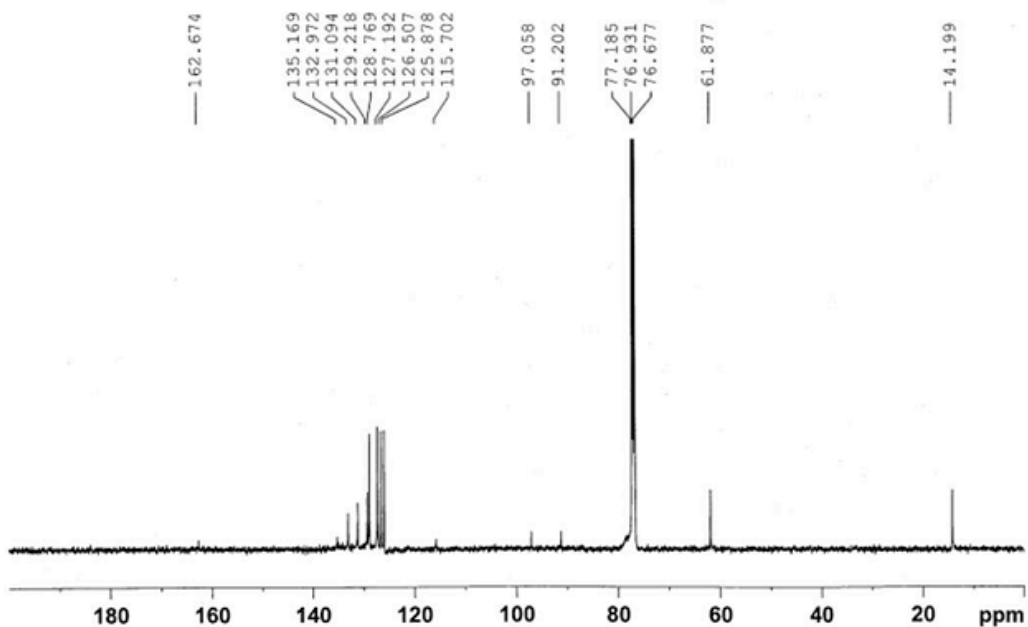


Figure S6: <sup>13</sup>C NMR spectrum of compound 1 in CDCl<sub>3</sub>.

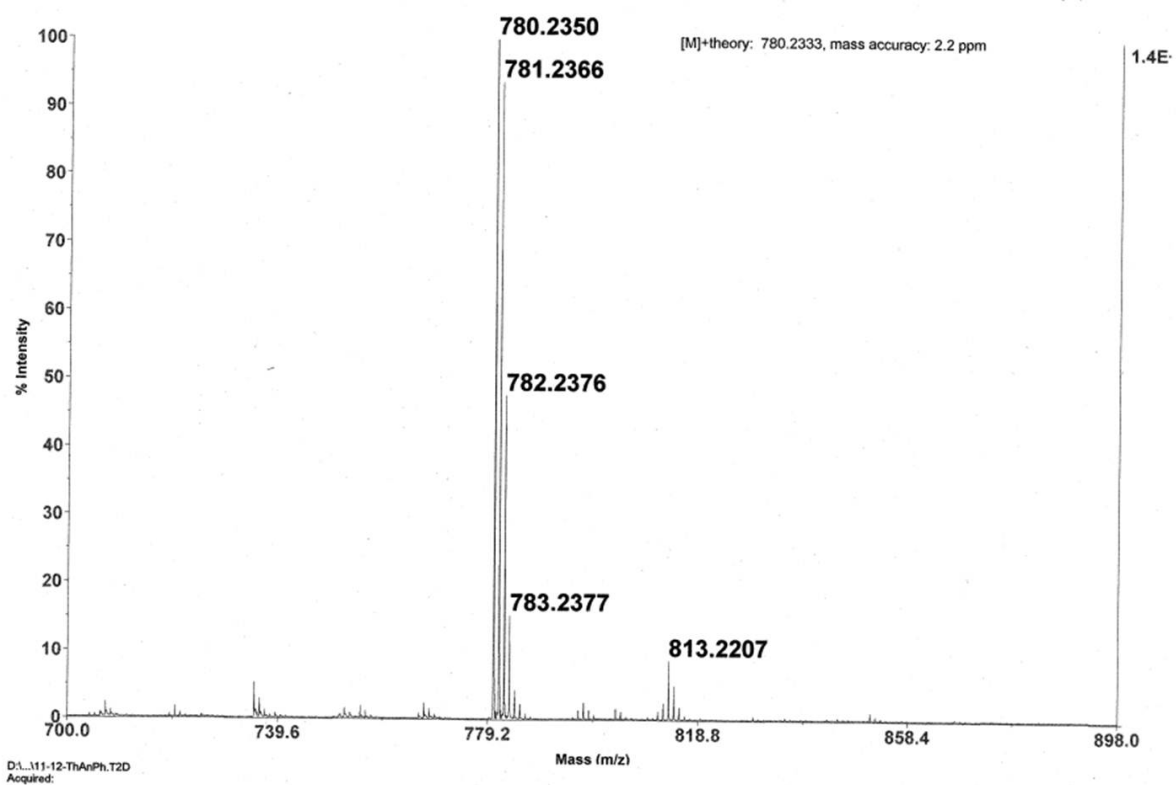
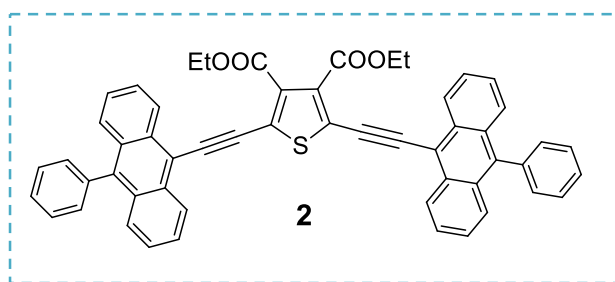


Figure S7: HR-MS mass spectrum of compound 2.

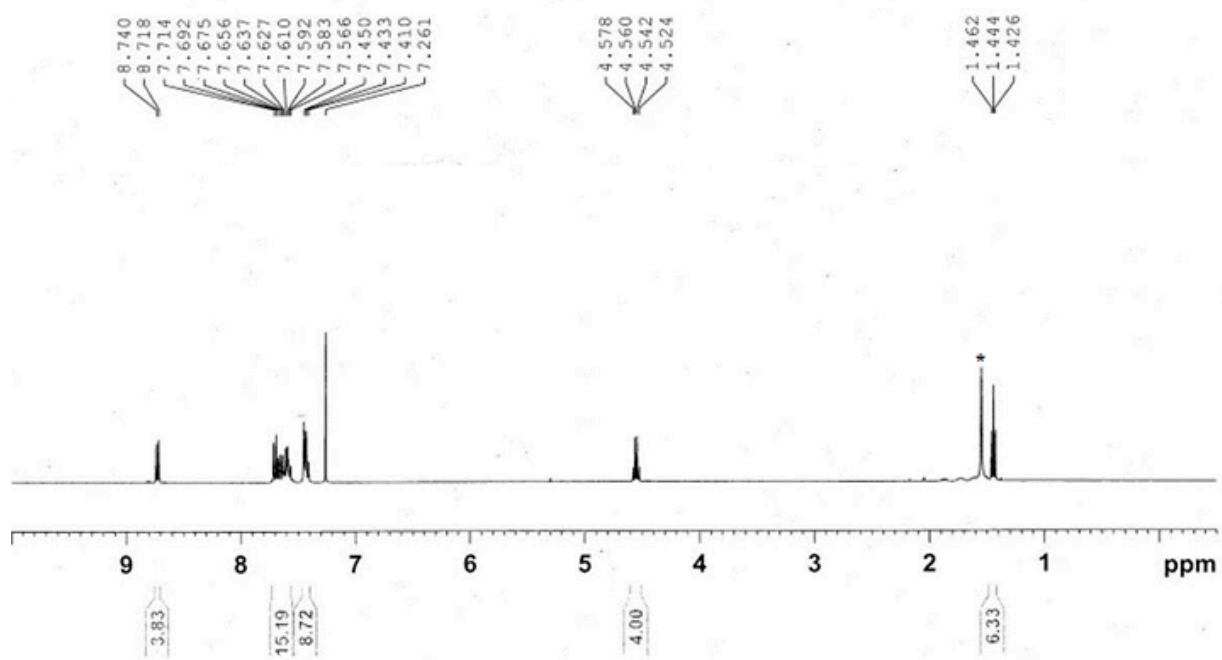


Figure S8: <sup>1</sup>H NMR spectrum of compound 2 in CDCl<sub>3</sub>.

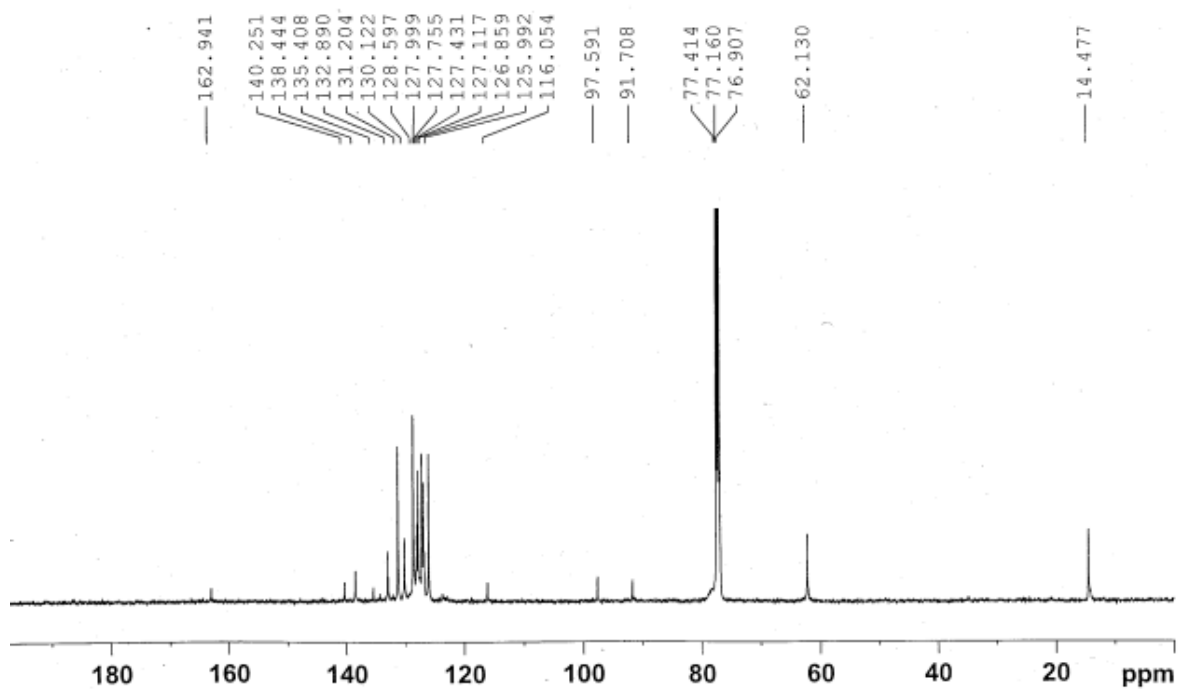


Figure S9: <sup>13</sup>C NMR spectrum of compound 2 in CDCl<sub>3</sub>.

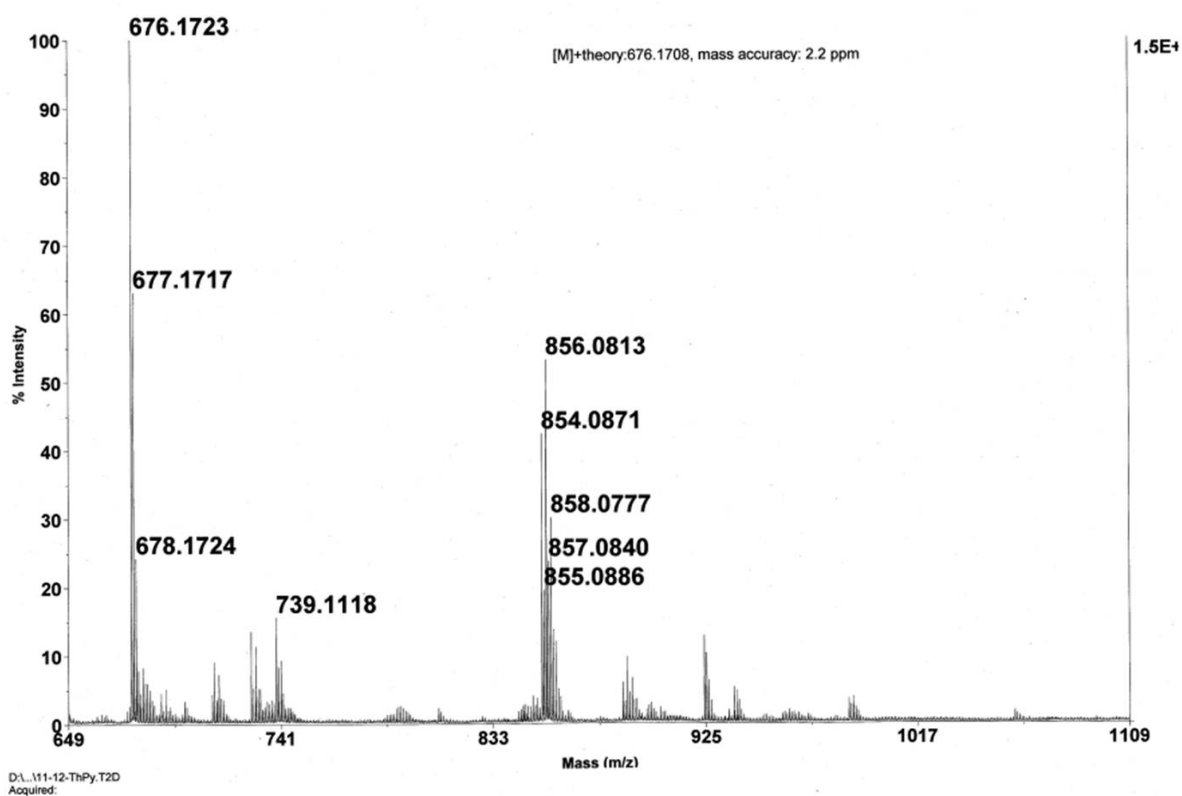
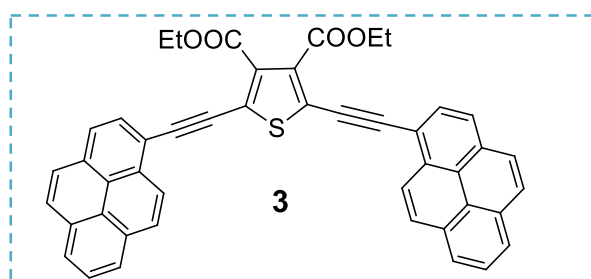


Figure S10: HR-MS mass spectrum of compound 3.

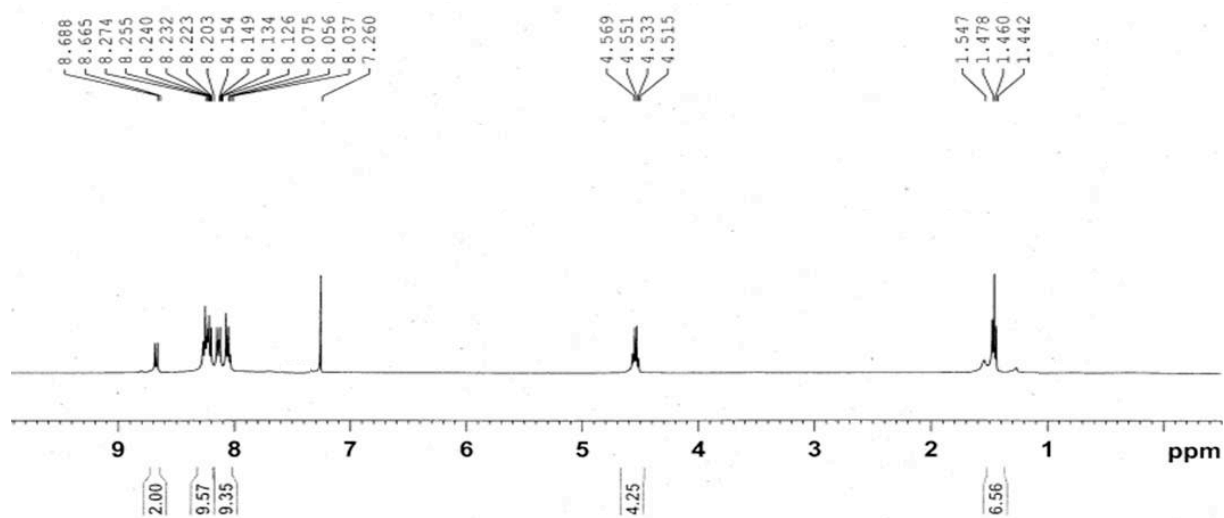


Figure S11:  $^1\text{H}$  NMR spectrum of compound **3** in  $\text{CDCl}_3$ .

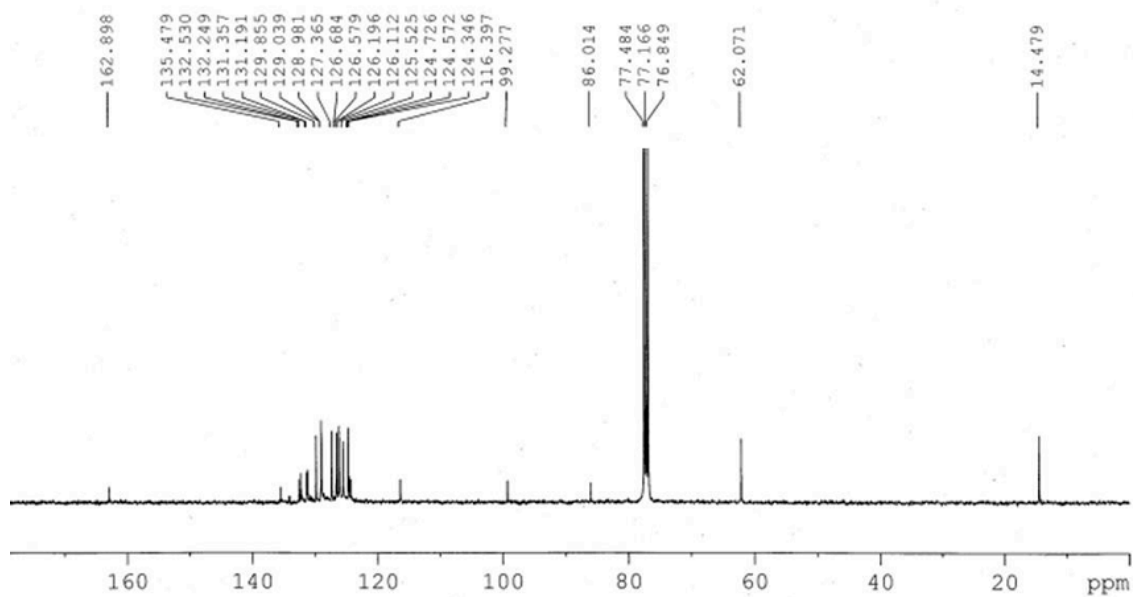


Figure S12:  $^{13}\text{C}$  NMR spectrum of compound **3** in  $\text{CDCl}_3$ .



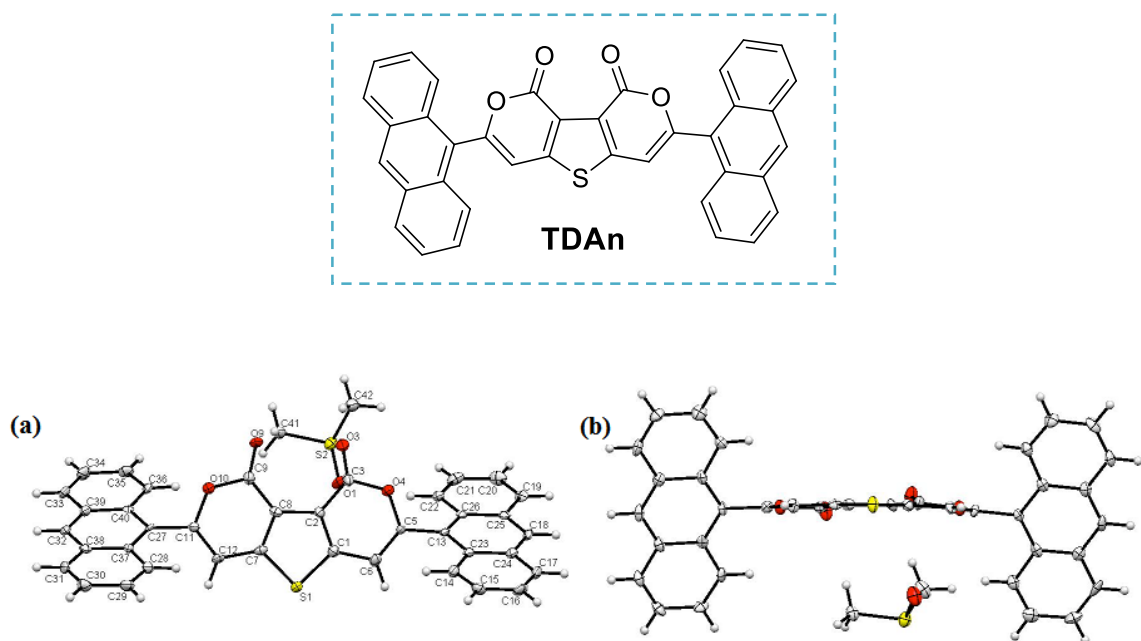


Figure S13: Crystal structure of compound TDAn in different views.

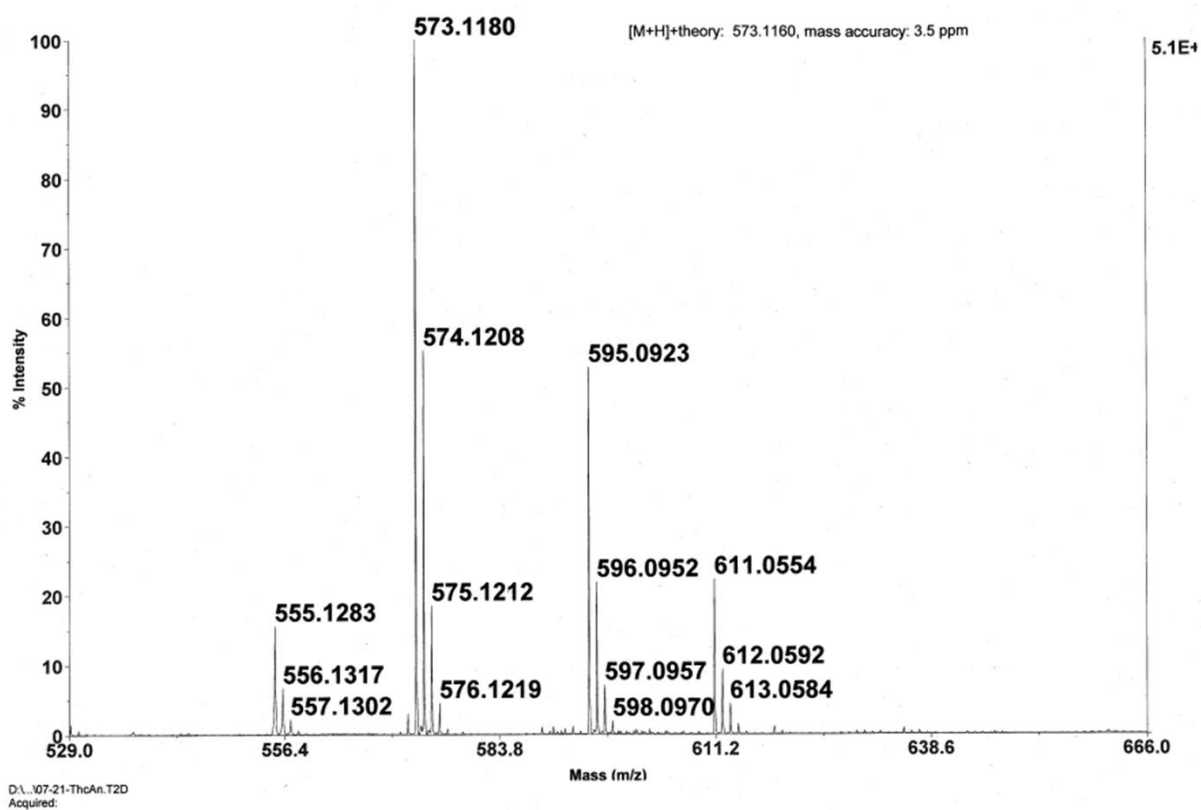
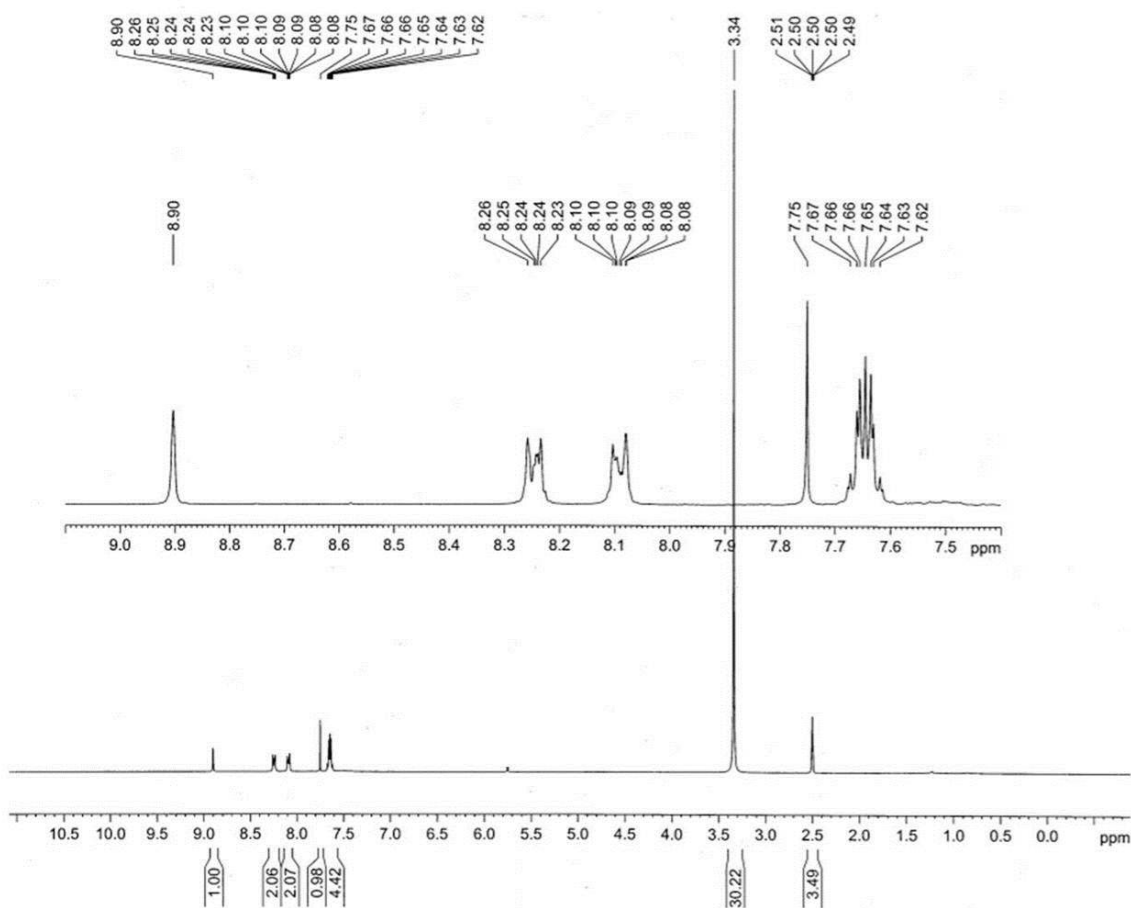
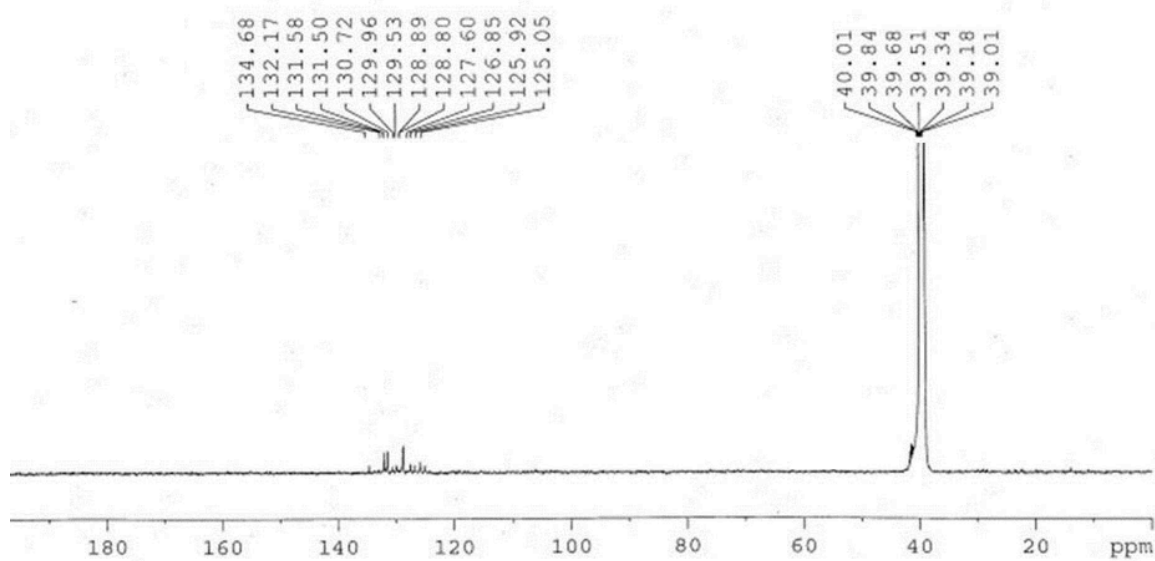


Figure S14: HR-MS mass spectrum of compound TDAn.



**Figure S15:**  $^1\text{H}$  NMR spectrum of compound **TDAn** in  $[\text{D}_6]$ -DMSO.



**Figure S16:**  $^{13}\text{C}$  NMR spectrum of compound **TDAn** in  $[\text{D}_6]$ -DMSO.

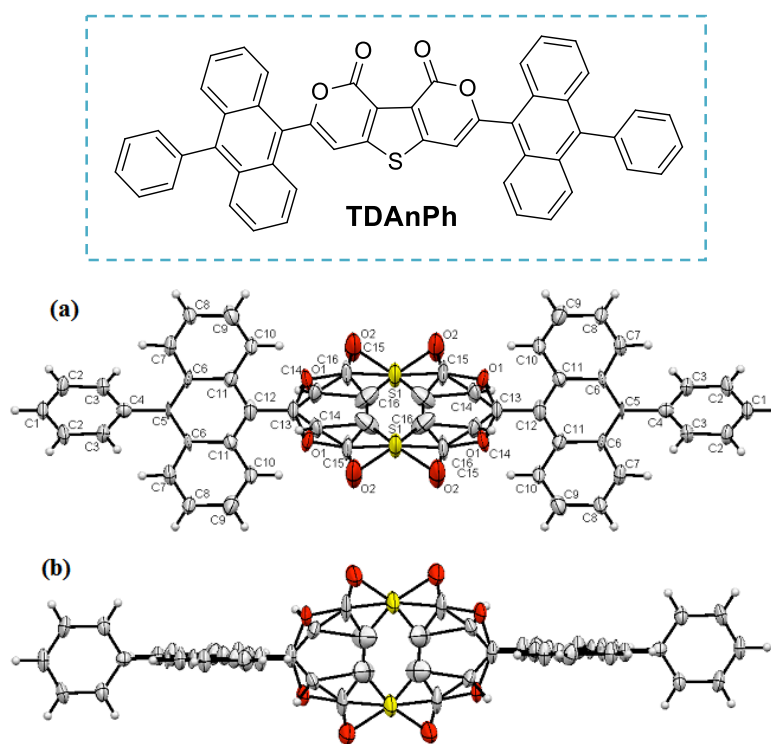


Figure S17: Crystal structure of compound **TDAnPh** in different views.

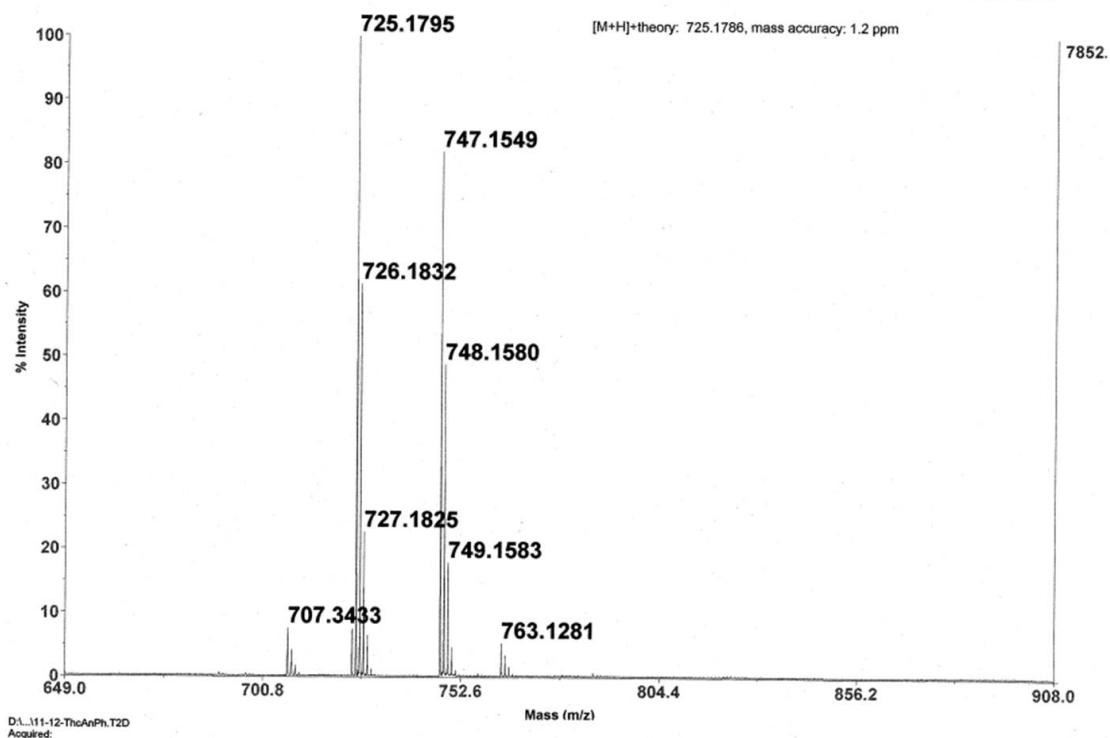
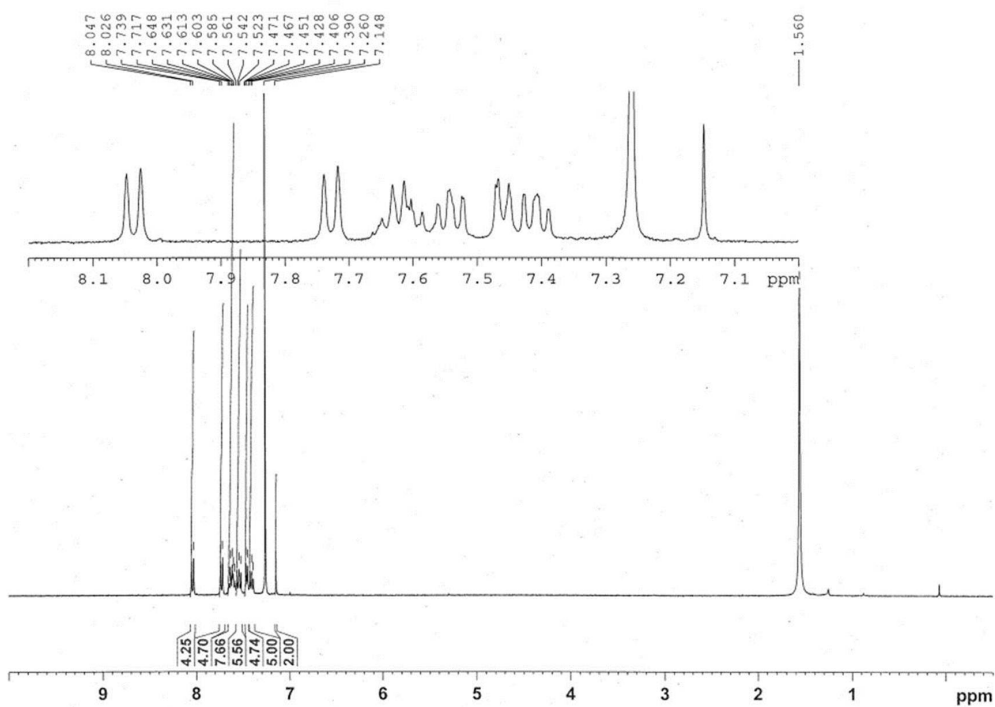
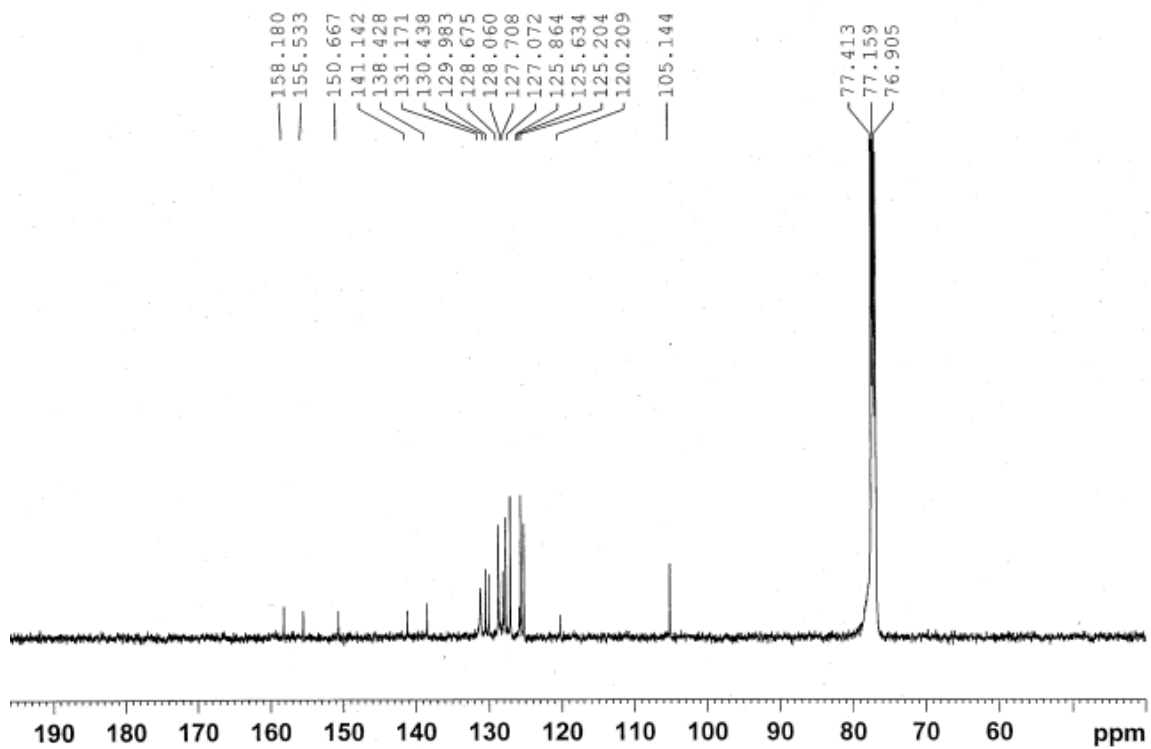


Figure S18: HR-MS mass spectrum of compound **TDAnPh**.



**Figure S19:** <sup>1</sup>H NMR spectrum of compound TDAnPh in CDCl<sub>3</sub>.



**Figure S20:** <sup>13</sup>C NMR spectrum of compound TDAnPh in CDCl<sub>3</sub>.

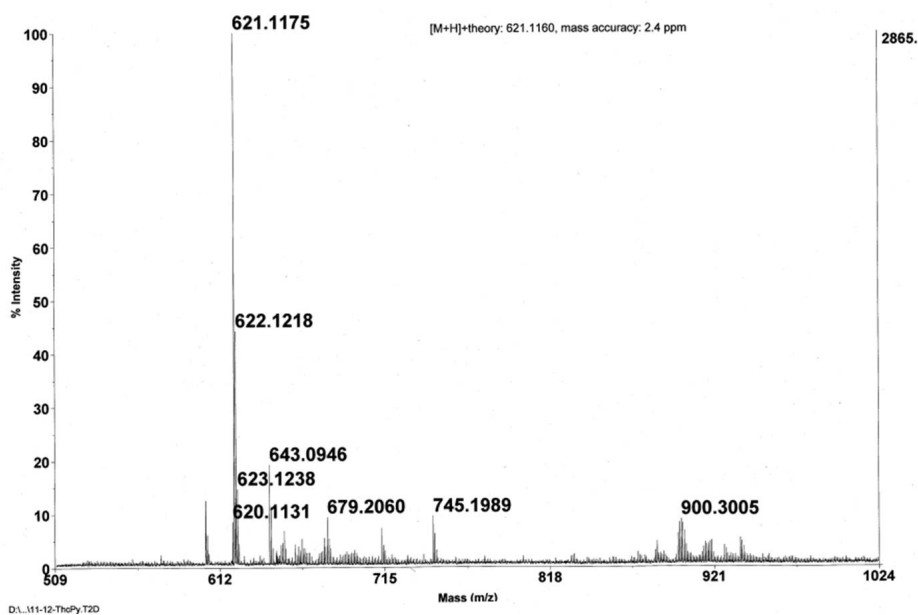
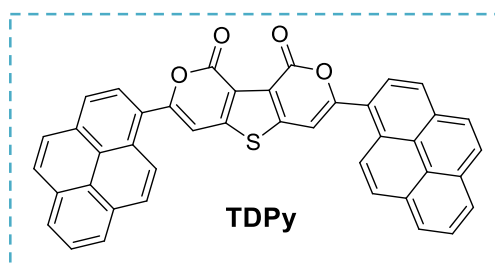


Figure S21: HR-MS mass spectrum of compound TDPy.

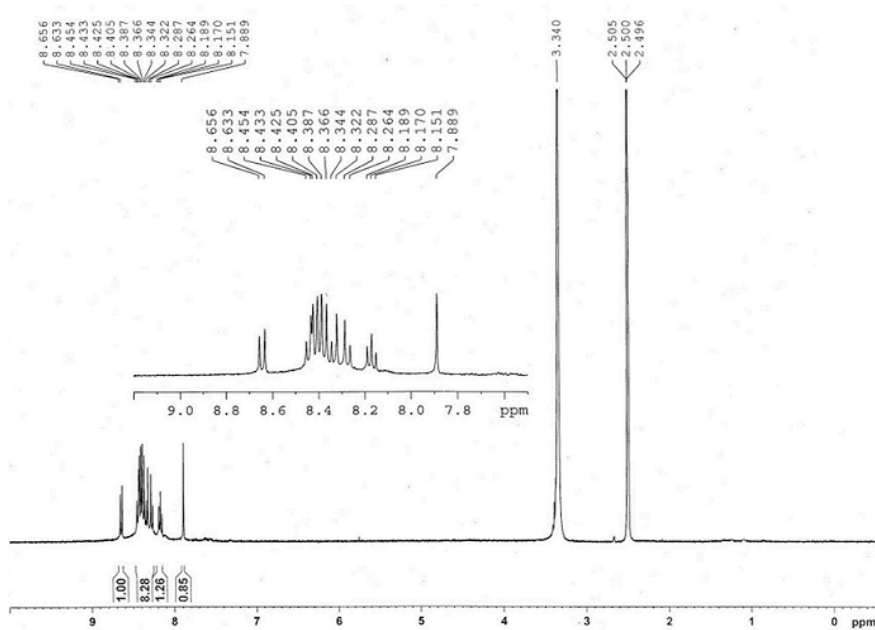
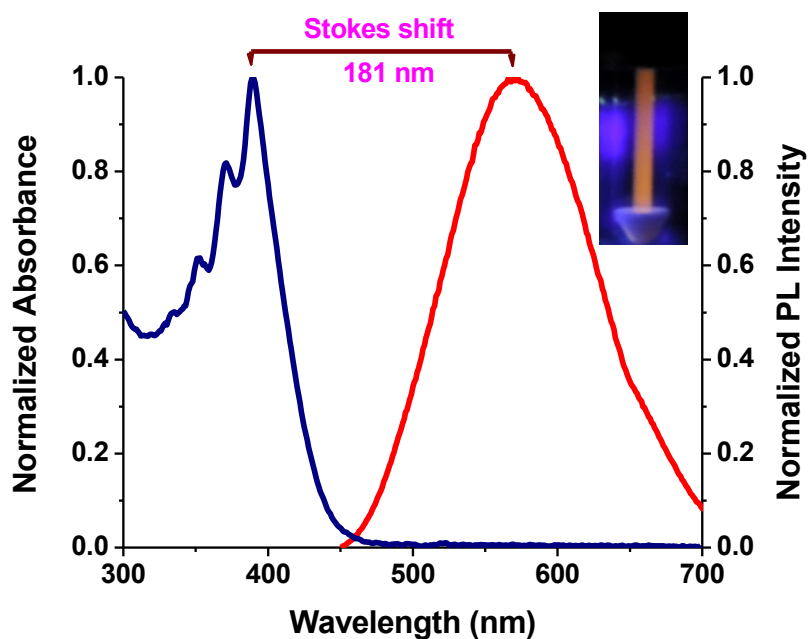
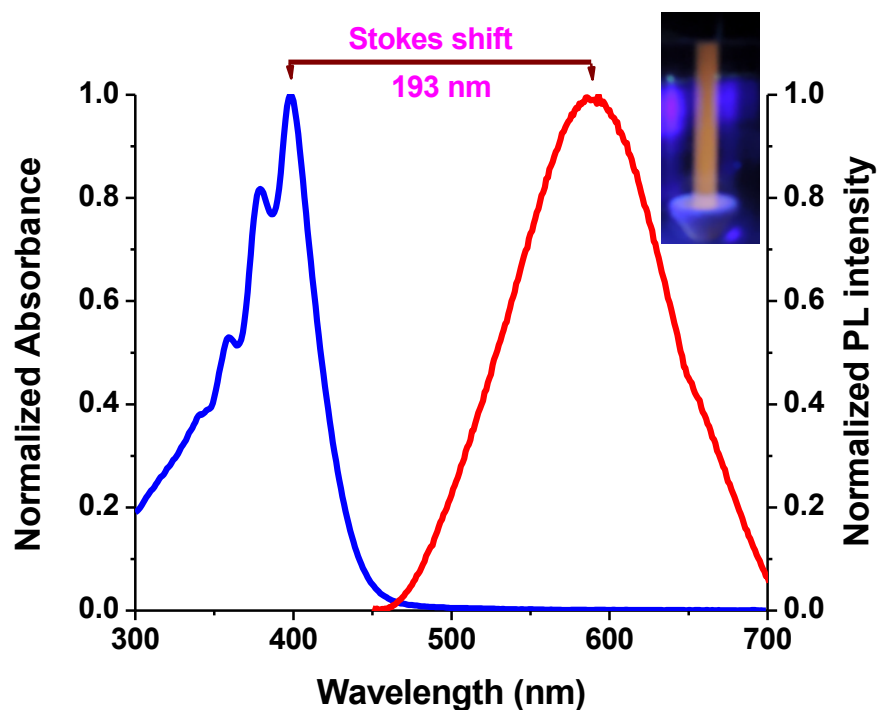


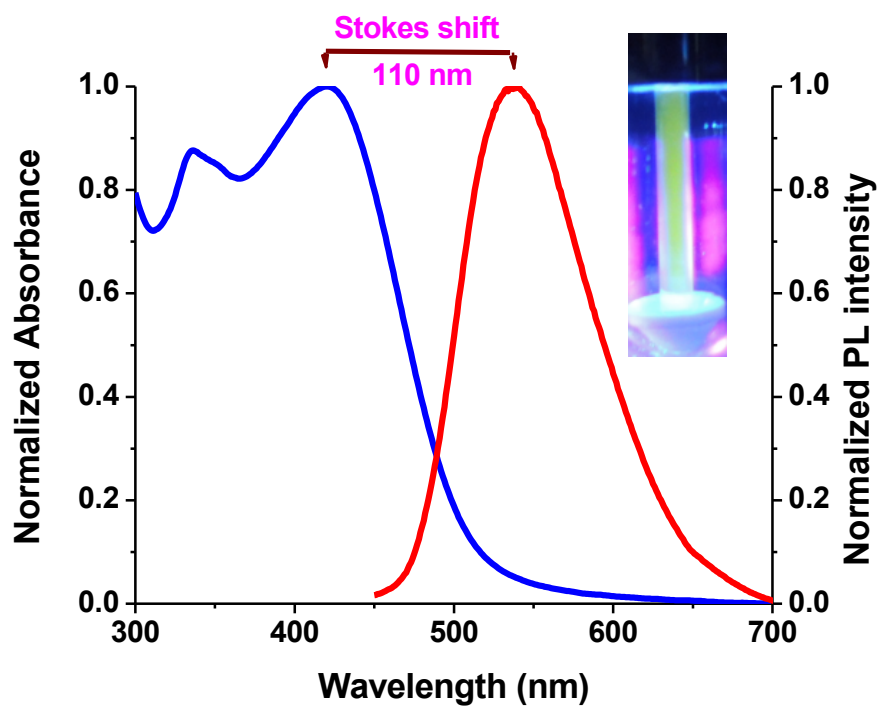
Figure S22: <sup>1</sup>H NMR spectrum of compound TDPy in [D<sub>6</sub>]-DMSO.



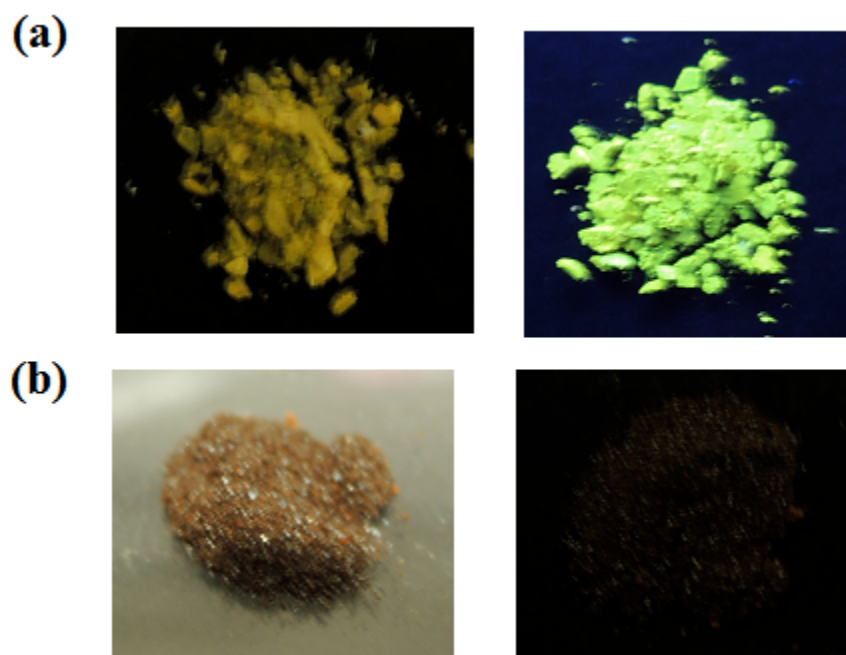
**Figure S23:** Normalized absorption ( $10^{-5}$  M) and PL ( $10^{-6}$  M) spectra of **TDAn** in toluene. Inset shows the colour of the compound under UV lamp (365 nm).



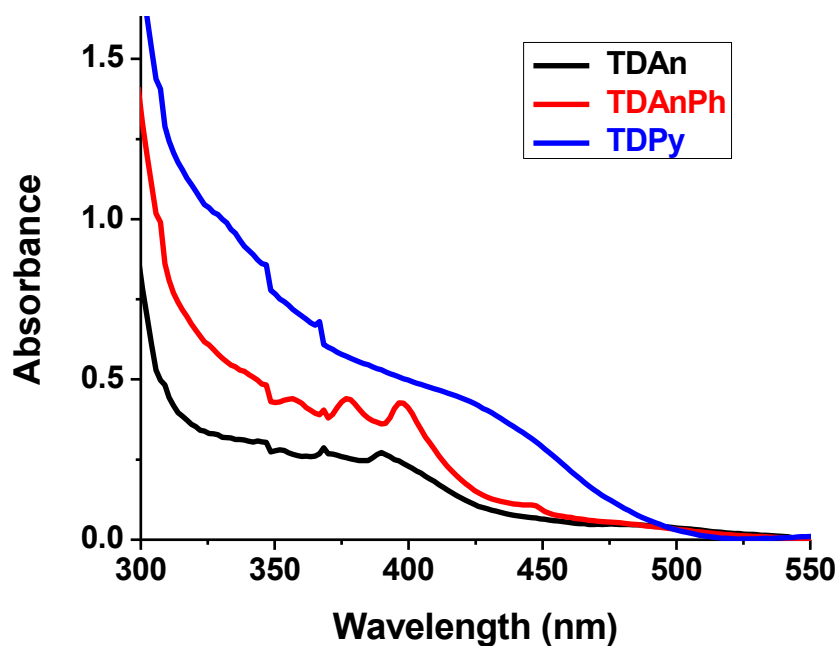
**Figure S24:** Normalized absorption ( $10^{-5}$  M) and PL ( $10^{-6}$  M) spectra of **TDAnPh** in toluene. Inset shows the colour of the compound under UV lamp (365 nm).



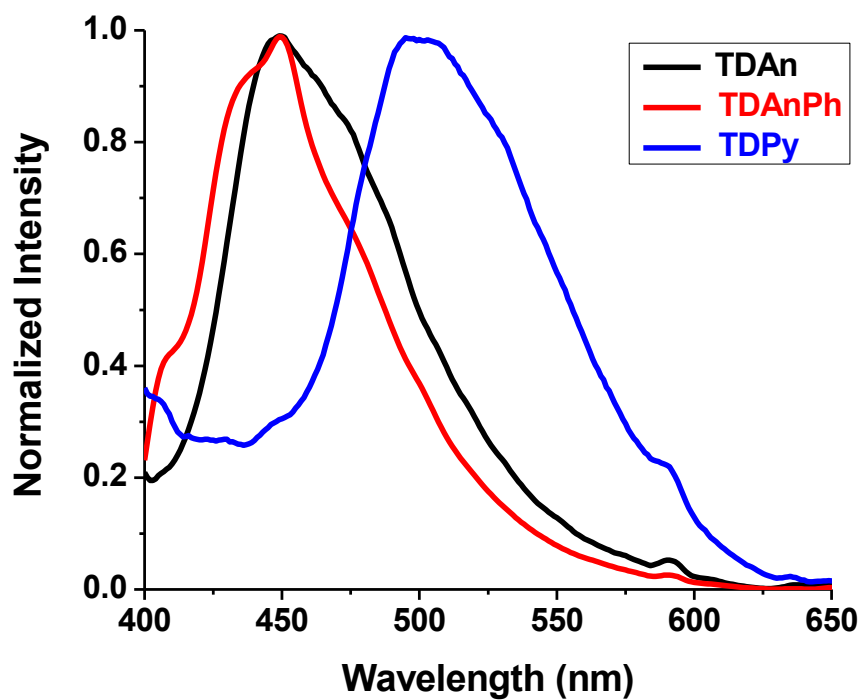
**Figure S25:** Normalized absorption ( $10^{-5}$  M) and PL ( $10^{-6}$  M) spectra of TDPy in toluene. Inset shows the colour of the compound under UV lamp (365 nm).



**Figure S26:** Photographs of TDAnPh (a) and TDPy (b) under ambient light (left) and UV-lamp (right).

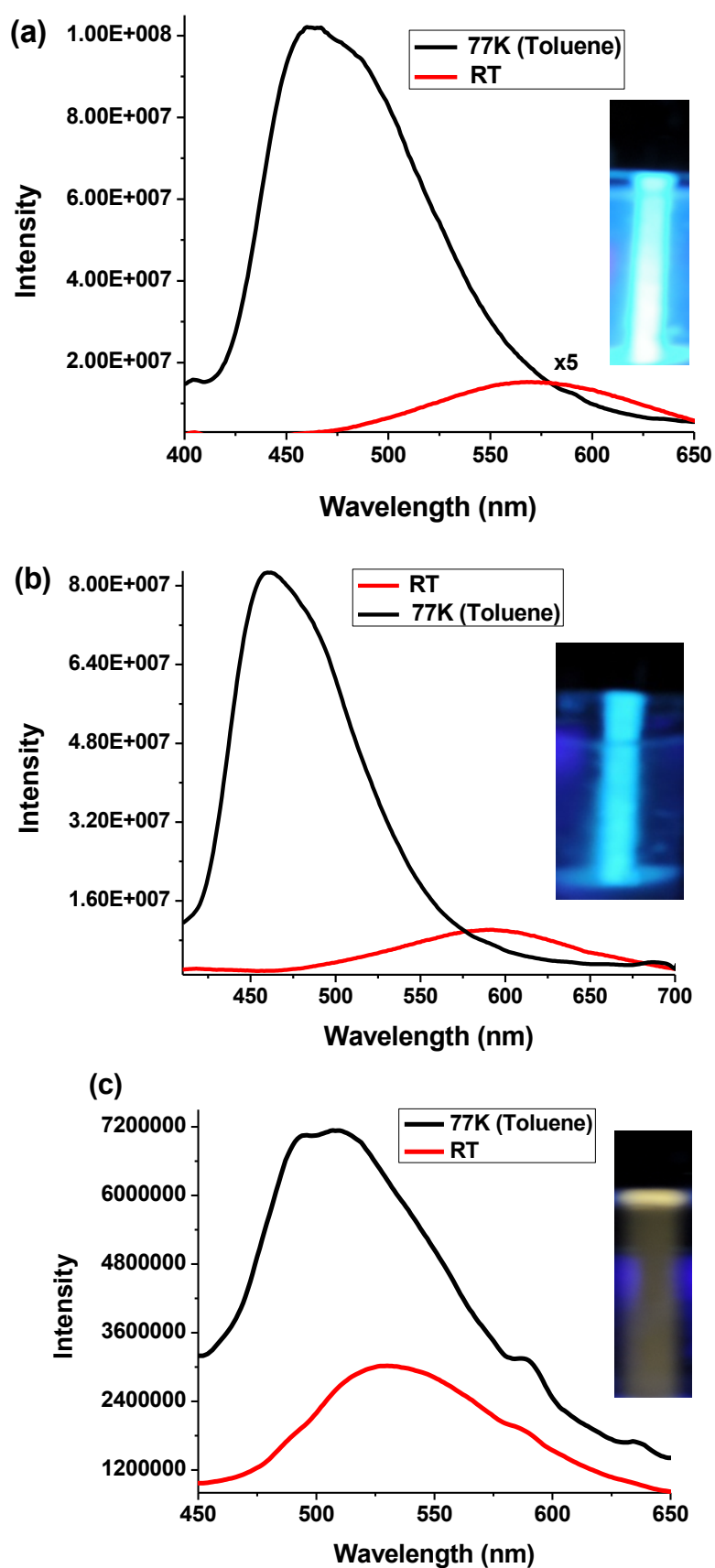


**Figure S27:** Comparison of absorption spectra of TDA n, TDA nPh and TDPy in 2-methyl THF at 77K ( $5 \times 10^{-6}$  M).

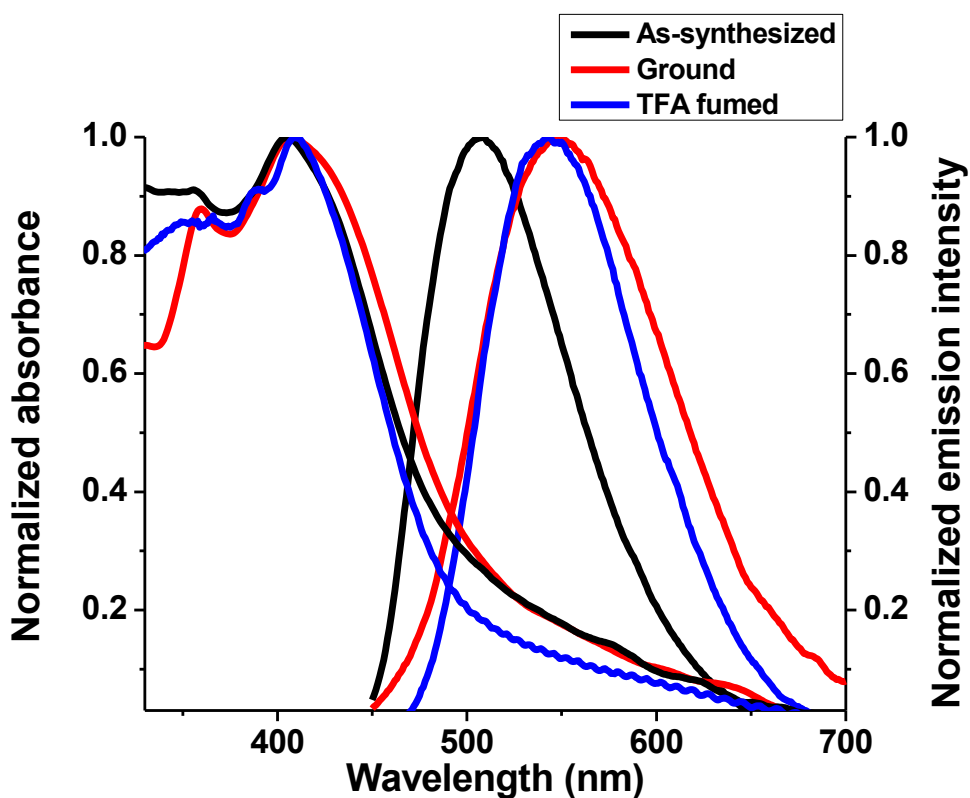


**Figure S28:** Comparison of PL spectra of TDA n, TDA nPh and TDPy in 2-methyl THF at 77K ( $1 \times 10^{-6}$  M).

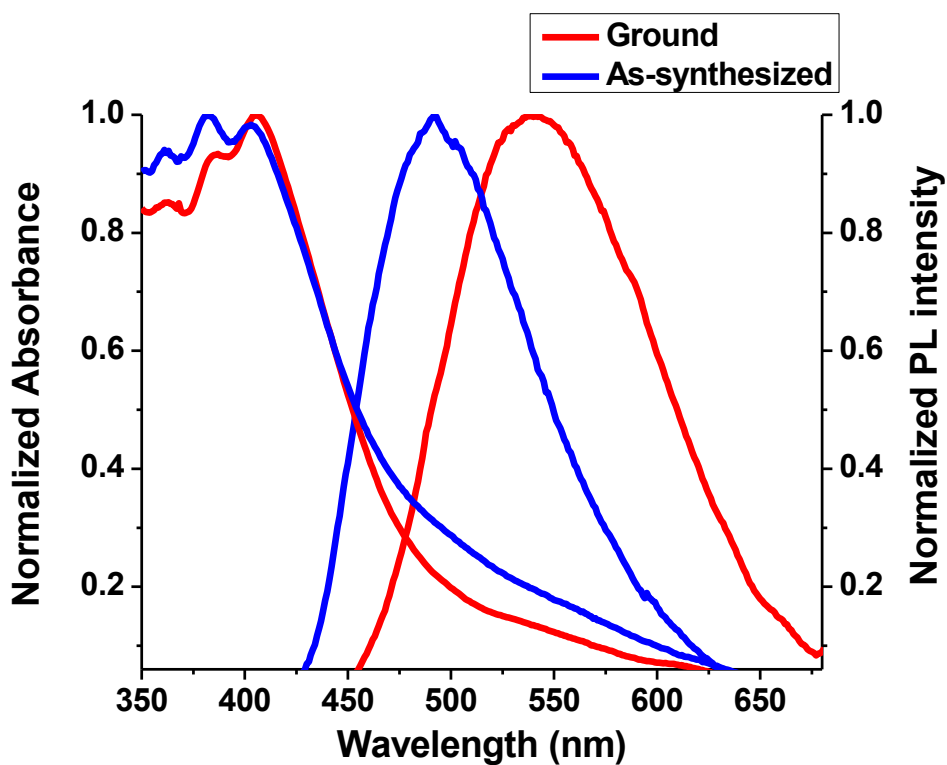




**Figure S29:** Comparison of PL spectra of (a) TDAn, (b) TDAnPh and (c) TDPy at room temperature and 77K ( $5 \times 10^{-6}$  M).



**Figure S30:** Solid-state absorption and emission spectra of TDAn in as-synthesized, ground and acid-fumed states.



**Figure S31:** Solid-state absorption and emission spectra of TDAnPh in as-synthesized, ground states.

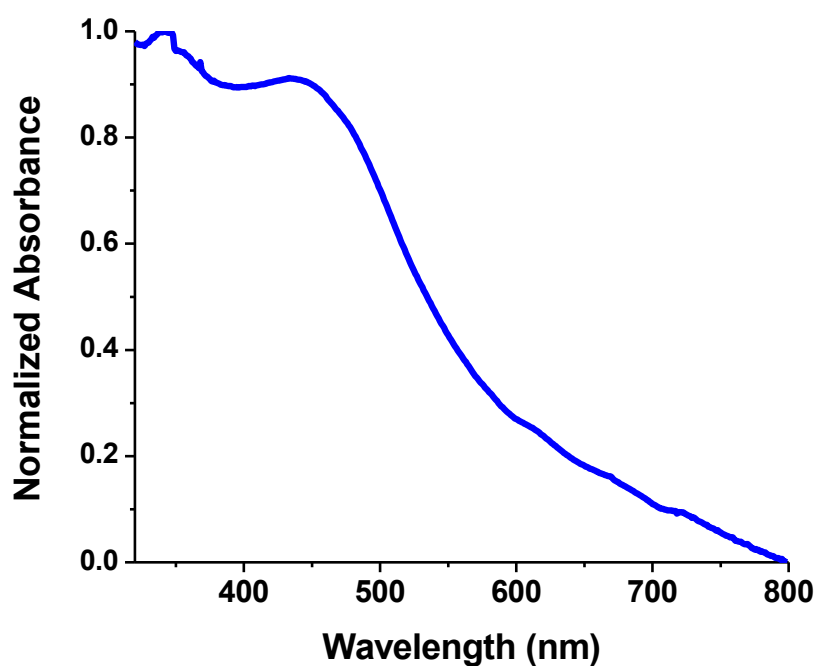


Figure S32: Solid-state absorption spectrum of TDPy.

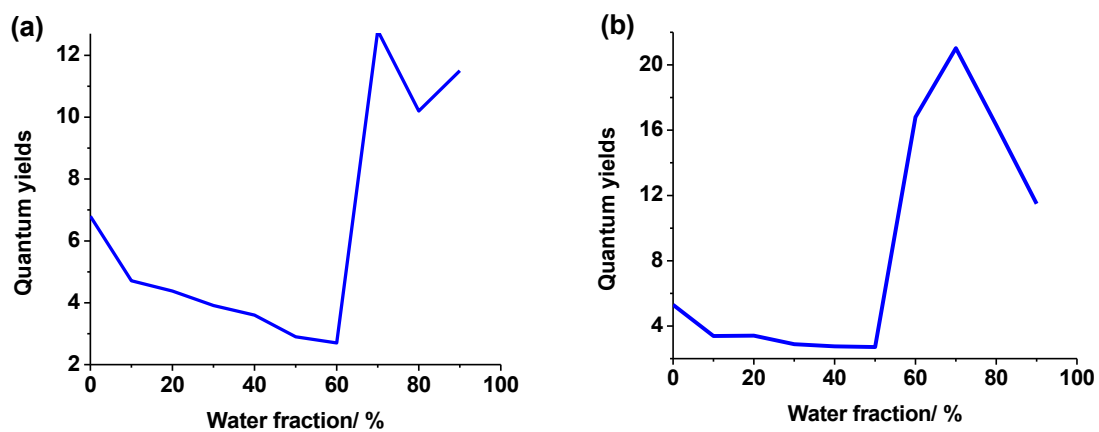


Figure S33: Change in quantum yields ( $\Phi$ ) of (a) TDAn (b) TDAnPh in different THF/water fraction.

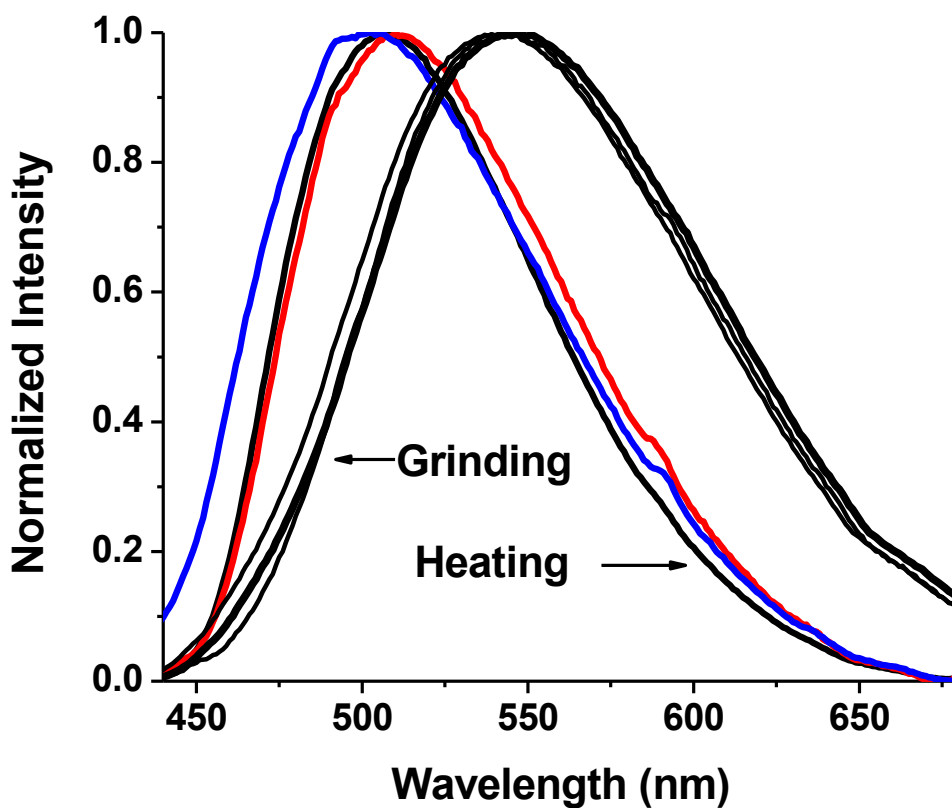


Figure S34: The normalized PL curves of TDAn reversible grinding and heating states

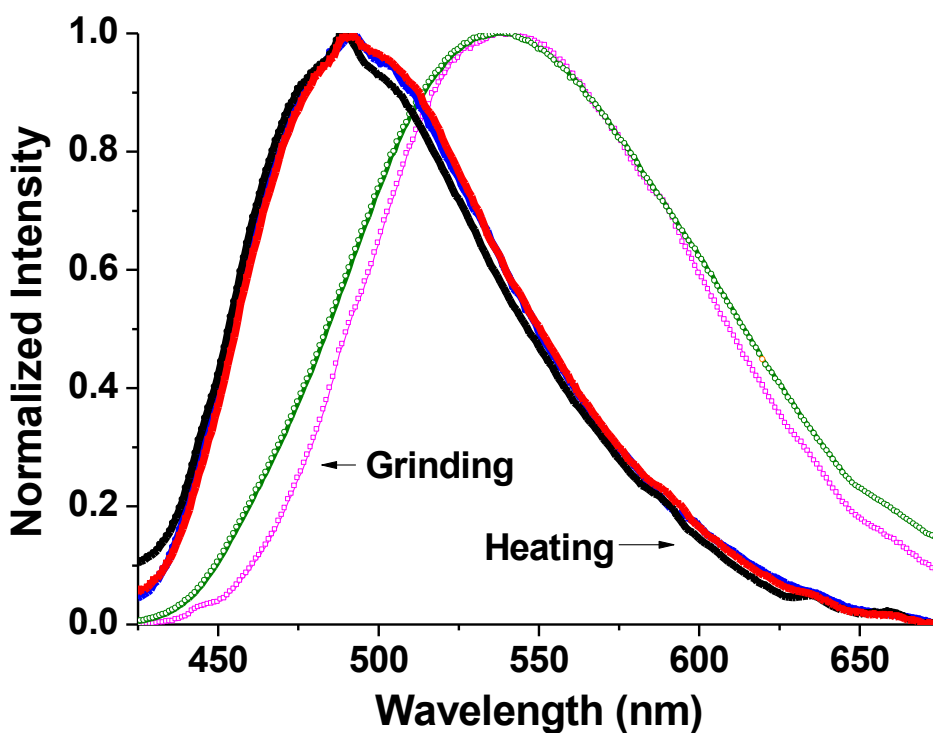
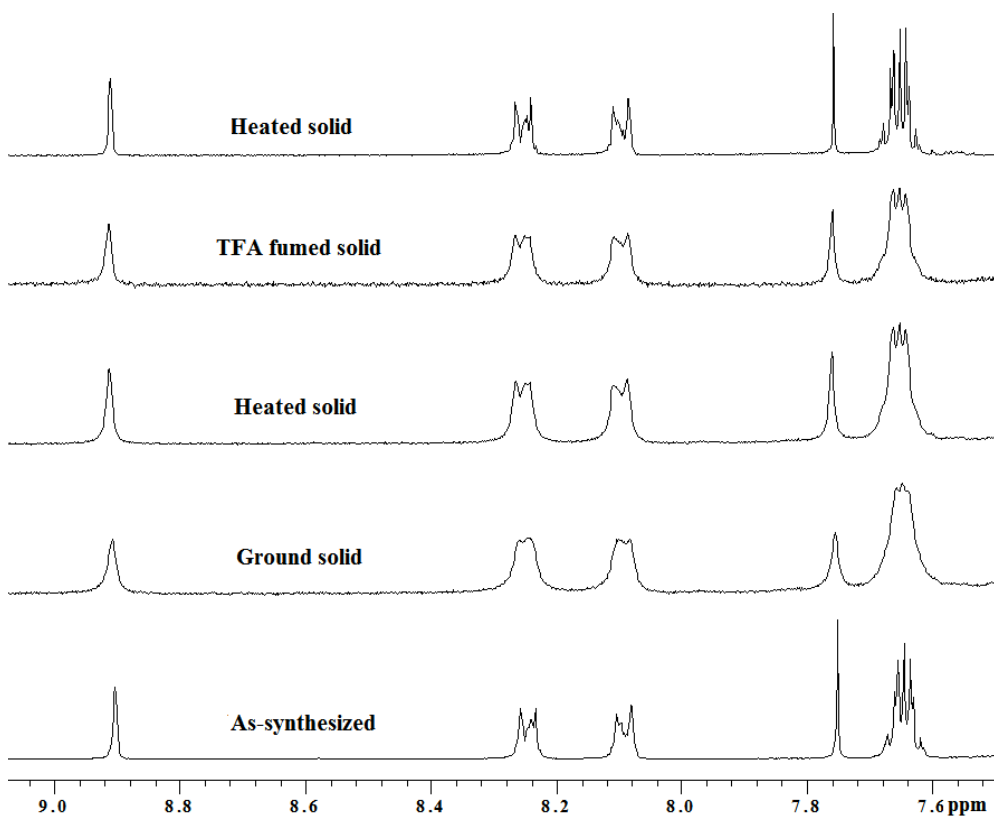
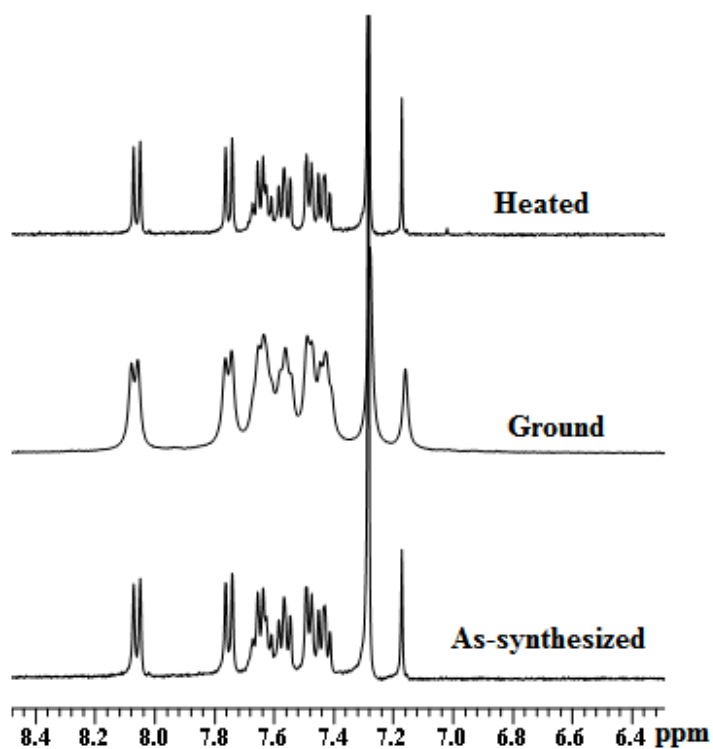


Figure S35: The normalized PL curves of TDAnPh reversible grinding and heating states



**Figure S36:** Comparison of  $^1\text{H}$  NMR spectra of TDAn in various states.



**Figure S37:** Comparison of  $^1\text{H}$  NMR spectra of TDAnPh in various states.

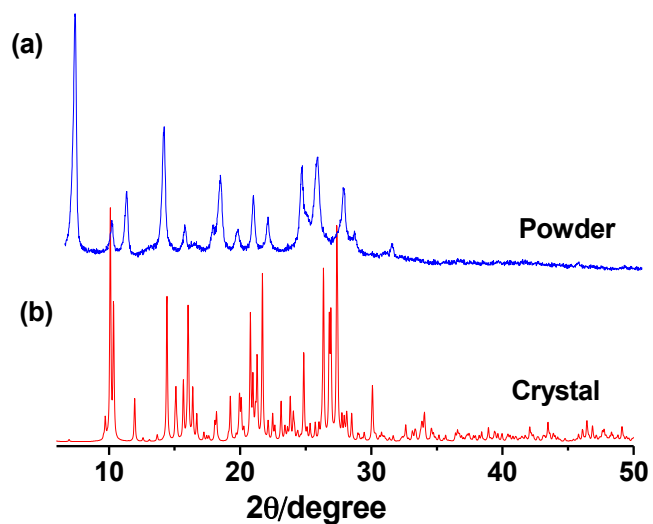


Figure S38: PXR D pattern of TDAn

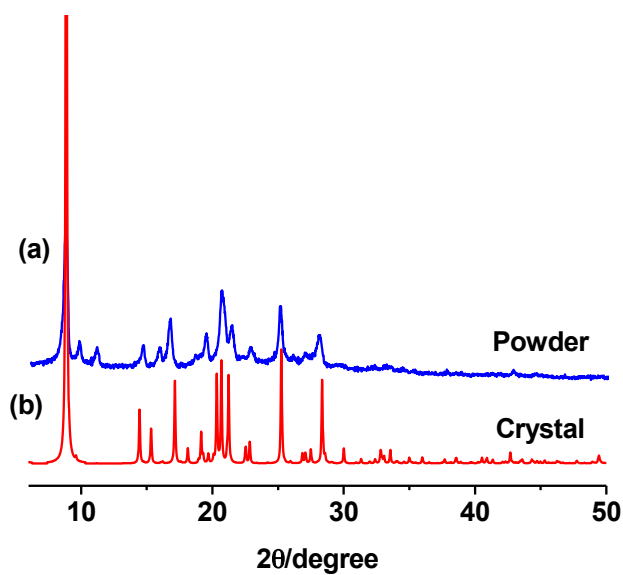


Figure S39: PXR D pattern of TDAnPh

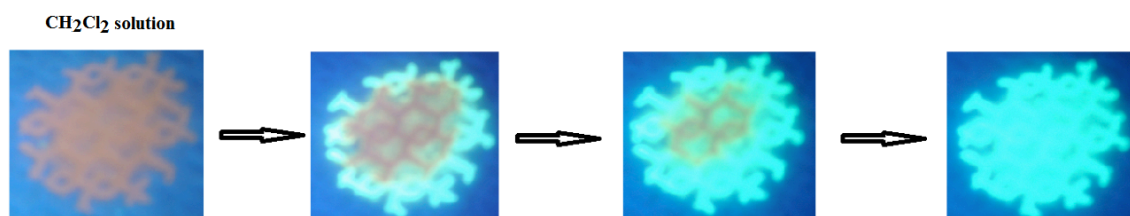
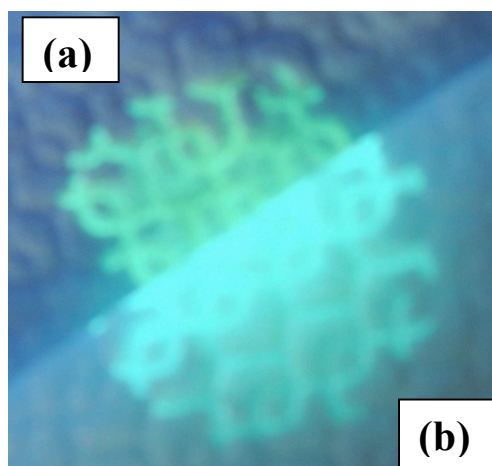


Figure S40: Photographs of TDAnPh in the process of evaporation of the solvent.



**Figure S41:** Solid state emission colours of (a) **TDAn** and (b) **TDAnPh**.

**Table S1:** Photophysical data of **TDAn**, **TDAnPh** and **TDPy** in various solvents

	Solvent	$\lambda_{\text{abs}}$ (nm)	$\lambda_{\text{em}}$ (nm)	Quantum yields ( $\phi$ )	Stokes shift (nm)
<b>TDAn</b>	Toluene	390	571	8.3	181
	CHCl <sub>3</sub>	389	573	6.5	184
	THF	391	573	6.8	182
	2-MeTHF <sup>a</sup>	390	449	49	59
<b>TDAnPh</b>	Toluene	398	591	6.2	193
	CHCl <sub>3</sub>	399	592	4.9	193
	THF	398	592	5.3	194
	2-MeTHF <sup>a</sup>	397	449	60	52
<b>TDPy</b>	Toluene	420	538	18.5	118
	CHCl <sub>3</sub>	421	537	15.4	116
	THF	418	535	17.1	117
	2-MeTHF <sup>a</sup>	422	505	42	83

<sup>a</sup>Data collected at 77k

**Table S2:** Crystal Data and Structure Refinements of **TDAn** and **TDAnPh**

	<b>TDAn</b>	<b>TDAnPh</b>
formula	C <sub>40</sub> H <sub>26</sub> O <sub>5</sub> S <sub>2</sub>	C <sub>30</sub> H <sub>30</sub> O <sub>4</sub> S
Formula weight	572.63	724.82
T, K	293 (2)	293 (2)
crystal system	monoclinic	monoclinic
space group	C 2/c	C 2/m
a, Å	36.933(5)	10.544(5)
b, Å	13.5190(17)	36.761(17)
c, Å	11.8390(13)	5.8903(19)
α, deg	90	90
β, deg	94.100(12)	101.277(16)
γ, deg	90	90
V, Å <sup>3</sup>	5896.05	2239.05
Z	8	2
density, mg/m <sup>3</sup>	1.466	1.075
μ/mm <sup>-1</sup>	0.231	0.112
θ range, deg	3.15 to 30.05°	2.05 to 25.11°
no. of reflns collected	36817	7680
no. of unique reflns		
R(int)	0.1992	0.1204
Goodness-of-fit on F <sup>2</sup>	1.013	0.876
R1 [I > 2σ(I)]	0.0962	0.0834
wR2 [I > 2σ(I)]	0.1762	0.2048
R1 (all data)	0.2354	0.2180
wR2 (all data)	0.2348	0.2504



## Computational Details

Geometry optimizations of the ground-state were done by density functional theory (DFT) employing the B3LYP functional and the 6-31G(d) basis set.<sup>1,2</sup>The frontier MO energy level distributions were estimated at the same functional level based on the optimized structures.

## References

1. R. Ditchfield, W. J. Hehre, and J. A. Pople, "Self-Consistent Molecular Orbital Methods. IX. An Extended Extended Gaussian-type basis for molecular-orbital studies of organic molecules," *J. Chem. Phys.* 1971, **54**, 724.
2. W. J. Hehre, R. Ditchfield, and J. A. Pople, "Self-Consistent Molecular Orbital Methods. XII. Further extensions of Gaussian-type basis sets for use in molecular-orbital studies of organic-molecules," *J. Chem. Phys.* 1972, **56**, 2257.

All calculations were performed with the Gaussian 09 program package:

Gaussian 09, Revision A.02, M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, G. Scalmani, V. Barone, B. Mennucci, G. A. Petersson, H. Nakatsuji, M. Caricato, X. Li, H. P. Hratchian, A. F. Izmaylov, J. Bloino, G. Zheng, J. L. Sonnenberg, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, T. Vreven, J. A. Montgomery, Jr., J. E. Peralta, F. Ogliaro, M. Bearpark, J. J. Heyd, E. Brothers, K. N. Kudin, V. N. Staroverov, R. Kobayashi, J. Normand, K. Raghavachari, A. Rendell, J. C. Burant, S. S. Iyengar, J. Tomasi, M. Cossi, N. Rega, J. M. Millam, M. Klene, J. E. Knox, J. B. Cross, V. Bakken, C. Adamo, J. Jaramillo, R. Gomperts, R. E. Stratmann, O. Yazyev, A. J. Austin, R. Cammi, C. Pomelli, J. W. Ochterski, R. L. Martin, K. Morokuma, V. G. Zakrzewski, G. A. Voth, P. Salvador, J. J. Dannenberg, S. Dapprich, A. D. Daniels, Ö. Farkas, J. B. Foresman, J. V. Ortiz, J. Cioslowski, and D. J. Fox, Gaussian, Inc., Wallingford CT, 2009.