

SUPPORTING INFORMATION FOR

A NEW $(\text{TTF})_{11}\text{I}_8$ ORGANIC MOLECULAR CONDUCTOR: FROM SINGLE CRYSTALS TO FLEXIBLE ALL-ORGANIC PIEZORESISTIVE FILMS

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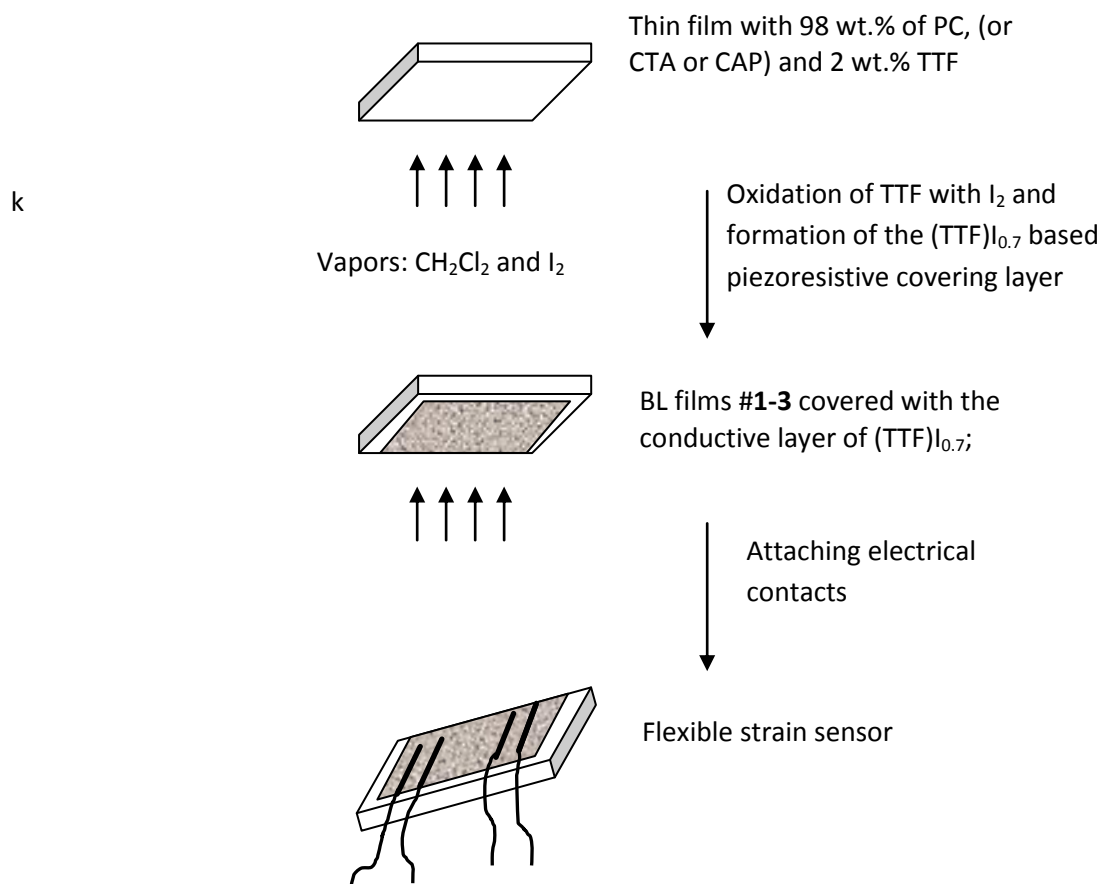


Figure S1. Scheme of the preparation method of bilayer (BL) thin films #1-3. PC=polycarbonate, CTA=cellulose triacetate, and CAP=cellulose acetate propionate.

An additional BL film (film #4), which contained copolymer of polycarbonate and siloxane as a polymeric support, was additionally prepared using this method. This BL film was only characterized by X-ray diffraction data (Fig S2b).

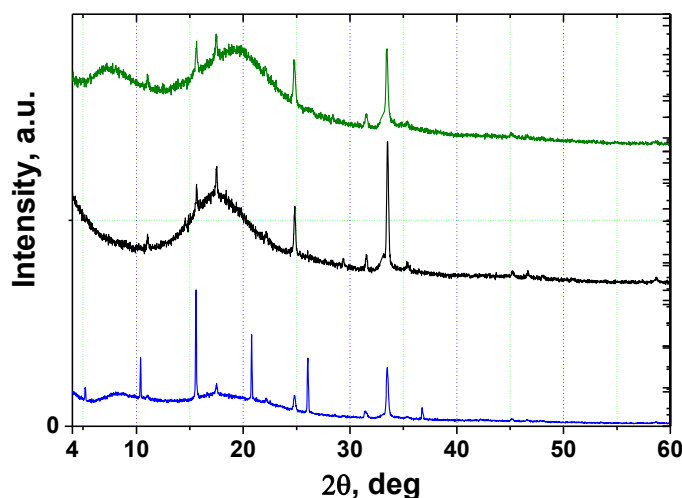


Figure S2a. X ray diffraction patterns of BL film #1 (black line), BL film #2 (blue line), and BL film #3 (green line). The film samples with dimensions $2 \times 3 \text{ cm}^2$ were attached to a glass support and X-ray diffraction data were recorded on a Rigaku “Rotaflex” RU-200B diffractometer in reflection mode with monochromatic $\text{CuK}\alpha$ radiation ($\lambda = 1.540598 \text{ \AA}$). The generator was activated at 50 kV and 80 mA

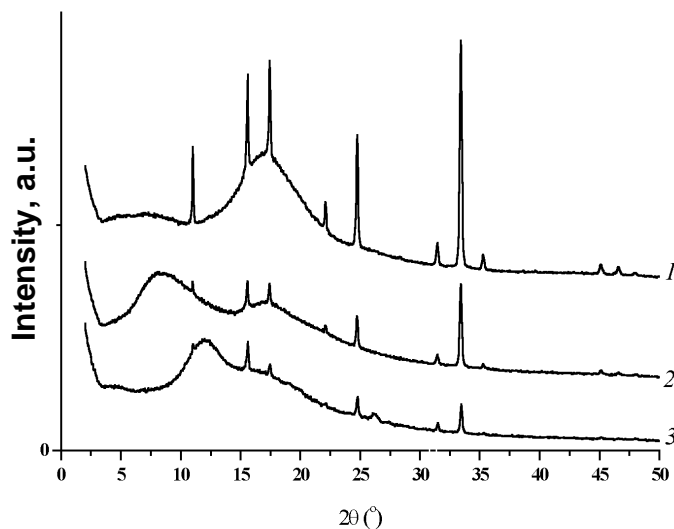


Figure S2b. X-ray patterns of BL thin films with conductive layer crystal of $(\text{TTF})\text{I}_{0.71+d}$ crystals on different polymeric substrates: polycarbonate (1) (replica of BL film #1), cellulose triacetate (2) (replica of BL film #2) and copolymer of polycarbonate and siloxane (3) (BL film #4). Data were collected in reflection mode using a Panalytical EMPYREAN instrument with a linear X²celerator detector using non-monochromated $\text{CuK}\alpha$ radiation

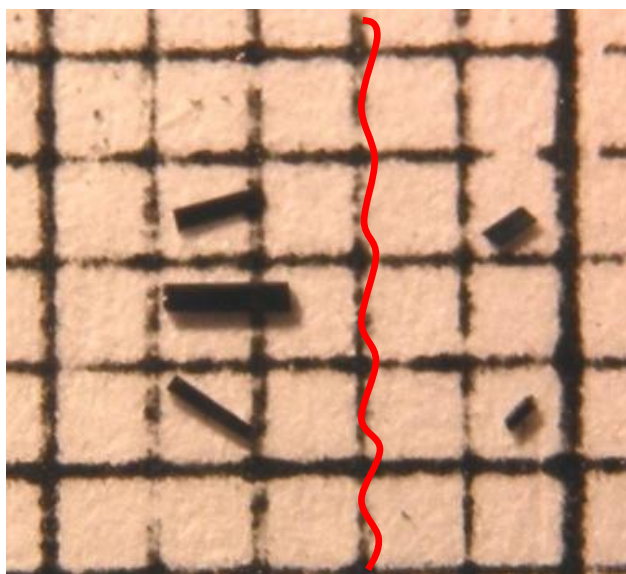


Figure S3. Photograph of $\text{TTF}_{11}\text{I}_8$ (left) and $(\text{TTF})\text{I}_3$ (right) crystals obtained using oxidation of TTF with iodine (see experimental part of the article). Dimension of cells: $1 \times 1 \text{ mm}^2$

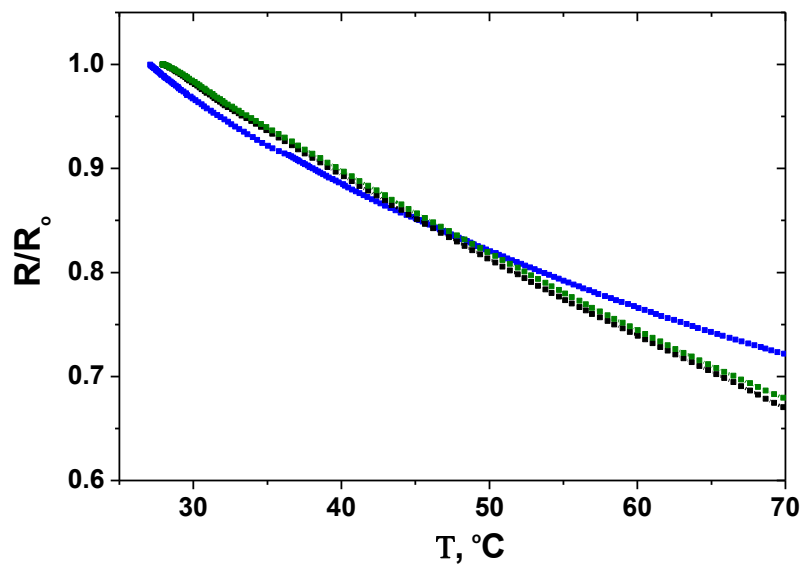


Figure S4. Temperature dependences of normalized electrical resistance of the BL film #1 (black line), film #2 (blue line), and film #3 (green line).

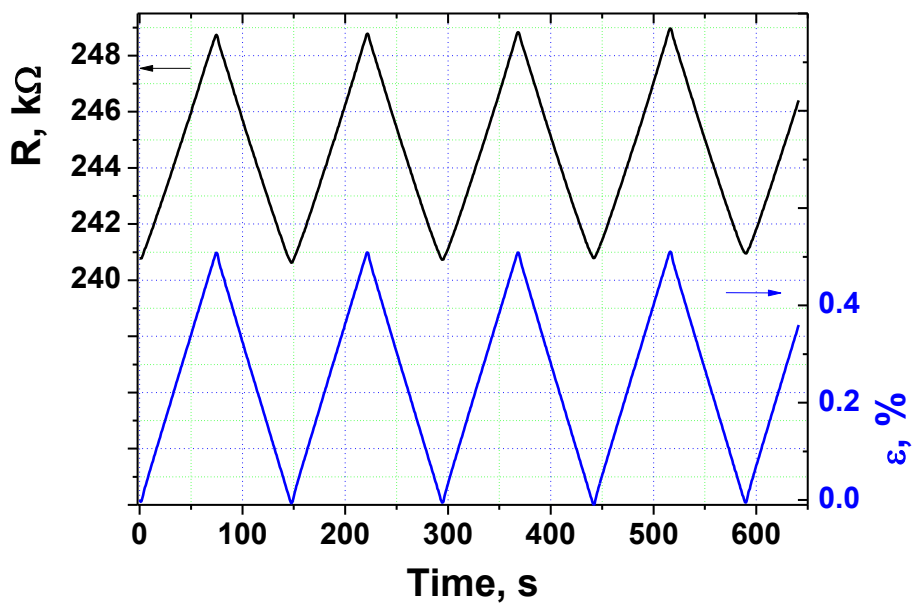


Figure S5. Electrical resistance response of the BL film #2 (CTA polymeric matrix) to perform cyclic elastic elongations using a maximum strain of $\epsilon_{\text{max}} = 0.5\%$.

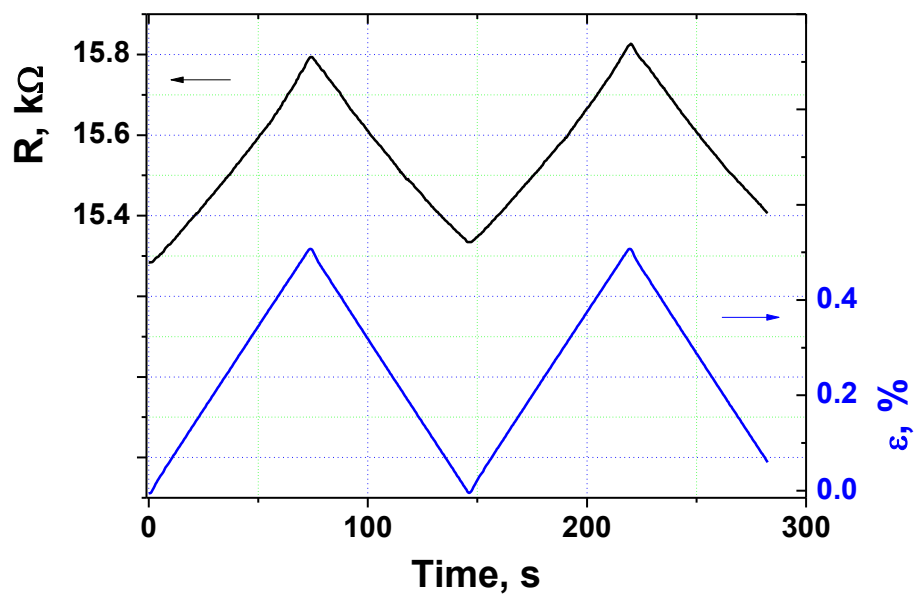


Figure S6. Electrical resistance response of BL film #3 (CAP polymeric matrix) to perform cyclic elastic elongation using a maximum of $\epsilon_{\max} = 0.5\%$.

Table S1. Distances between iodine species extracted from the resolved crystal structure of TTF₁₁I₈, as well as the values of the bond length in I-I and (I-I-I)⁻ added for comparison.

Number of I ⁻	Object1	Object2	Length, Å
1	I8	I3	4.883(5)
2	I4	I3	4.900(6)
3	I7	I4	4.892(6)
4	I1	I5	4.928(5)
5	I5	I6	4.864(6)
6	I6	I2	4.875(6)
I-I			2.68-2.72¹
I₃⁻			2,79-3,11²

Table S2. The central $C^1=C^4$ and C^4-S^4 bond length extracted from the resolved crystal structure of $TTF_{11}I_8$, the charge transfer of each molecule of TTF in asymmetric unit was calculated from Ref 3 using following equation $r=0.762+0.049q$, where r is ratio between bond lengths as it show in the table and q is a charge of TTF species.

Molecule TTF	Atom1	Atom2	Length, Å	ratio, $r=(L(C^1=C^4)/L(C^4-S^4))$	charge, q
A	S4	C4	1.723	0,796	0,700
	C1	C4	1.372		
B	S4	C4	1.722	0,809	0,958
	C1	C4	1.393		
C	S4	C4	1.712	0,801	0,804
	C1	C4	1.372		
D	S4	C4	1.722	0,809	0,958
	C1	C4	1.393		
E	S4	C4	1.732	0,798	0,733
	C1	C4	1.382		
F	S4	C4	1.722	0,797	0,721
	C1	C4	1.373		
G	S4	C4	1.722	0,803	0,840
	C1	C4	1.383		
H	S4	C4	1.722	0,809	0,958
	C1	C4	1.393		
I	S4	C4	1.712	0,807	0,923
	C1	C4	1.382		
J	S4	C4	1.712	0,813	1,043
	C1	C4	1.392		
K	C1	C4	1.414	0,825	1,295
	S1	C4	1.713		
L	S4	C4	1.722	0,810	0,970
	C1	C4	1.394		

References

1. U. Buontempo et al., J. Chem. Phys. 107 (1997), 5720.
2. Atkins et al *Inorganic Chemistry* 5th ed.(2010) Oxford University Press. p.431.
3. T.C.Umland et al., J. Phys. Chem. 92 (1986), 6456.