Supporting Information for

CEE-active Red/NIR Fluorophores with Triple-channel Solid-state "ON/OFF" Fluorescent Switching

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Fig. S1. TD-DFT calculations of the emission process for complex 1.



Fig. S2 PXRD patterns of complexes 2 (a) 3 (b) 4 (c) and 5 (d) in different states.



Fig. S3 Photoluminescent spectra of complex 4 in the solid state by acid/base vapor

fuming.



Fig. S4 Photoluminescent spectra of complex 1 in CH_2Cl_2 (1×10⁻⁵ M) acidized with different amount of HCl.



Fig. S5 PXRD patterns of complex 4 acidized with HCl.



Fig. S6 Photoluminescent spectra of TEA- and HCl-fumed samples a) 1, b) 2, c) 3 and

d) 5.



Fig. S7 Photoluminescent spectra of complexes 1–5 (a-e) upon heating, cooling and

solvent annealing.



Fig. S8 The calculated molecular conformation for complexes 1–5.



Fig. S9 The measured and simulated PXRD data for complexes 1 (a) and 4 (b).



Fig. S10 The optimized ground state (GS) and excited state (ES) structures for complex **1** with its neutral state (left), with comparison with its protonation state from the top view (upper panel) and side view (lower panel). The H atoms are omitted for clarification.



Fig. S11 PXRD patterns of complex **1** (red line: crystalline sample of form II; blue line: the sample of crystalline powder of form I after heating and solvent annealing treatment).



Fig. S12 DSC curve of the crystalline powders of complex 1.



Figure S13. ¹H-NMR spectra of 2 recorded in CDCl₃ (500 MHz).



Figure S14. ¹³C-NMR spectra of 2 recorded in CDCl₃ (125 MHz).



Figure S15. ¹H-NMR spectra of **3** recorded in CDCl₃ (300 MHz).



Figure S16. ¹H-NMR spectra of 4 recorded in CDCl₃ (300 MHz).



Figure S17. ¹H-NMR spectra of 5 recorded in CDCl₃ (500 MHz).