

## Electronic Supporting Information

### Highly sensitive conjugated polymer fluorescent sensors based on benzochalcogendiazole for nickel ions in real-time detection

**Yunxiang Lei, Hui Li, Wenxia Gao, Miaochang Liu, Jiuxi Chen, Jinchang Ding,  
Xiaobo Huang\* and Huayue Wu\***

College of Chemistry and Materials Engineering, Wenzhou University, Wenzhou 325035, China

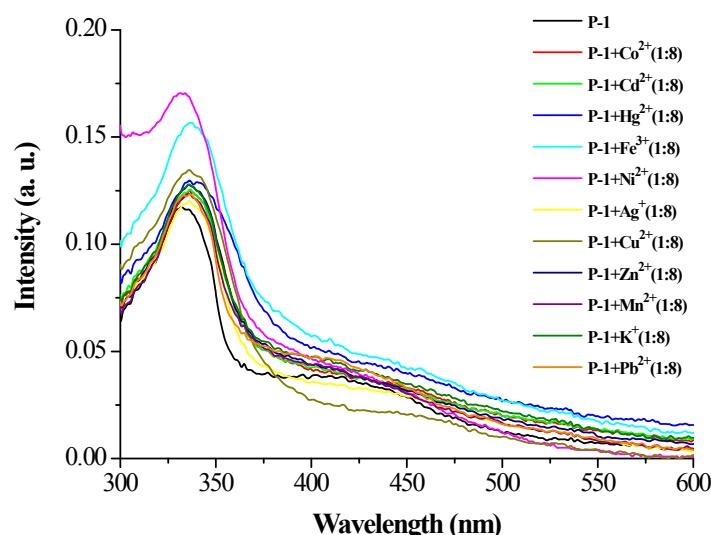
\*Corresponding authors. Tel.: +86 577 88368280; fax: +86 577 88368280

E-mail addresses: xiaobhuang@wzu.edu.cn (X. Huang); huayuewu@wzu.edu.cn (H. Wu)

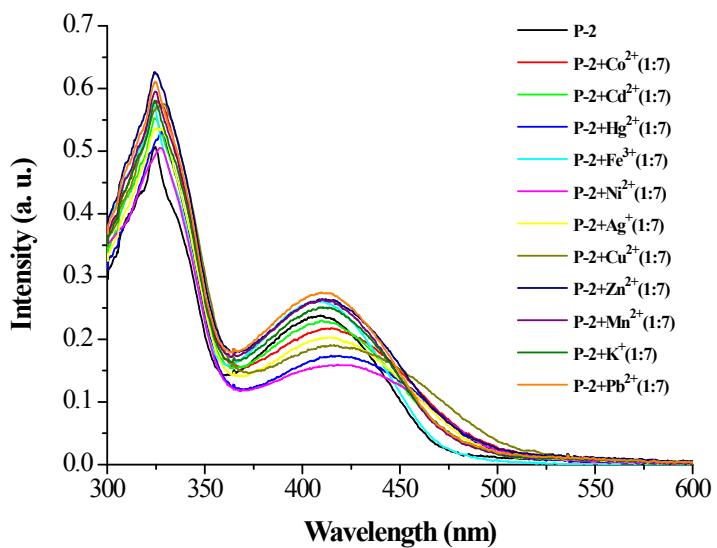
#### Contents:

1. UV-vis absorption responsive behaviors of P-1 and P-2 on various metal ions
2. Fluorescence responsive behaviors of P-1 and P-2 on various metal ions
3. NMR spectra of the important compounds and the conjugated polymers P-1 and P-2

#### 1. UV-vis absorption responsive behaviors of P-1 and P-2 on various metal ions

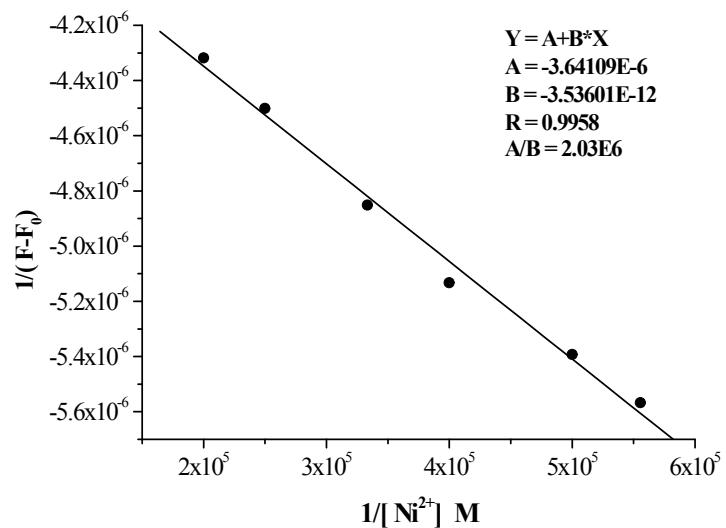


**Fig. S1** UV-vis spectra of P-1 ( $10 \mu\text{mol}\cdot\text{L}^{-1}$ ,  $\text{CHCl}_3$ ) in the presence of various metal ions (each  $80 \mu\text{mol}\cdot\text{L}^{-1}$ ,  $\text{CH}_3\text{CN}$ )

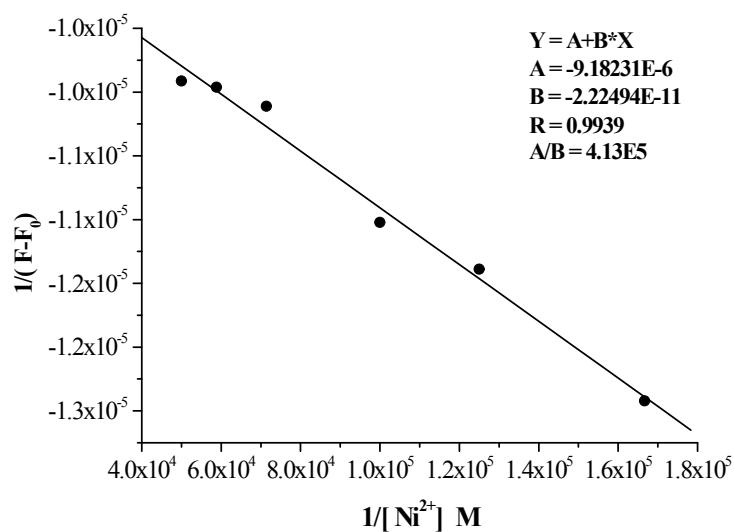


**Fig. S2** UV-vis spectra of **P-2** ( $10 \mu\text{mol}\cdot\text{L}^{-1}$ ,  $\text{CHCl}_3$ ) in the presence of various metal ions (each  $70 \mu\text{mol}\cdot\text{L}^{-1}$ ,  $\text{CH}_3\text{CN}$ )

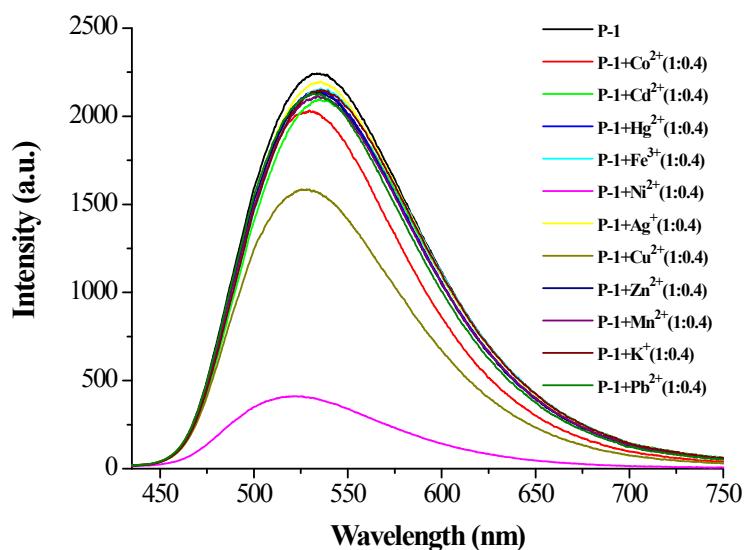
## 2. Fluorescence responsive behaviors of **P-1** and **P-2** on various metal ions



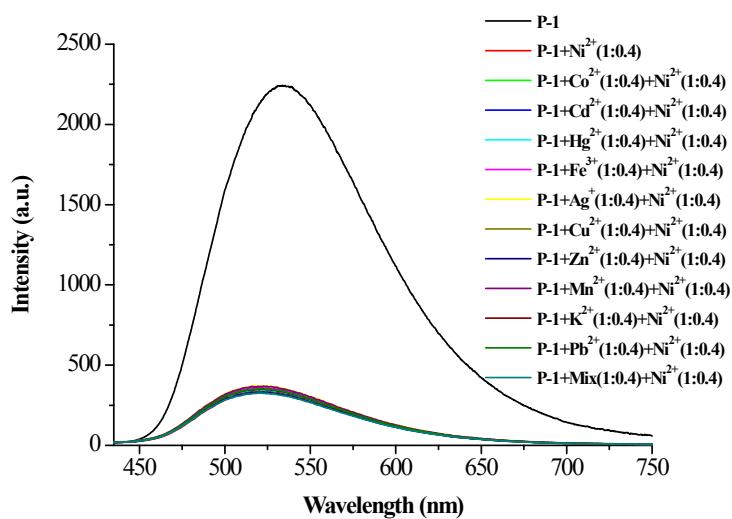
**Fig. S3** Benesi-Hildebrand plot for  $\text{Ni}^{2+}$ -bound **P-1**.



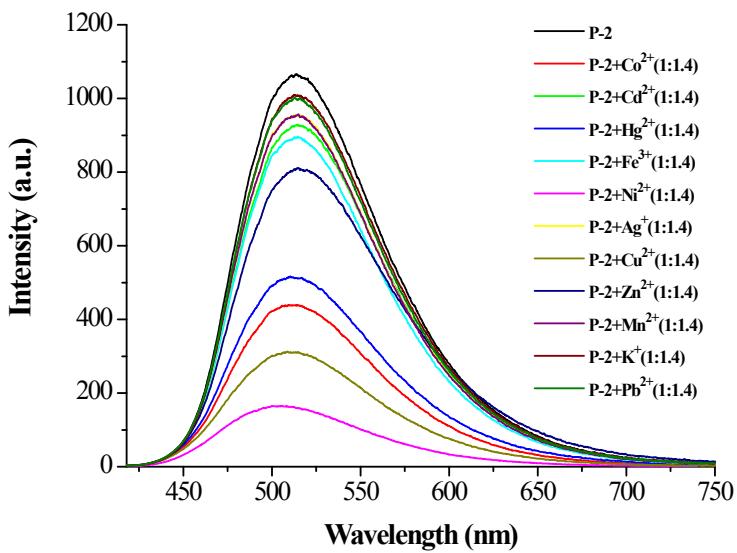
**Fig. S4** Benesi-Hildebrand plot for  $\text{Ni}^{2+}$ -bound **P-2**.



**Fig. S5** Fluorescence spectra of **P-1** ( $10 \mu\text{mol}\cdot\text{L}^{-1}$ ) in  $\text{CHCl}_3$  in the presence of in the presence of various metal ions in  $\text{CH}_3\text{CN}$  (each  $4 \mu\text{mol}\cdot\text{L}^{-1}$ ).

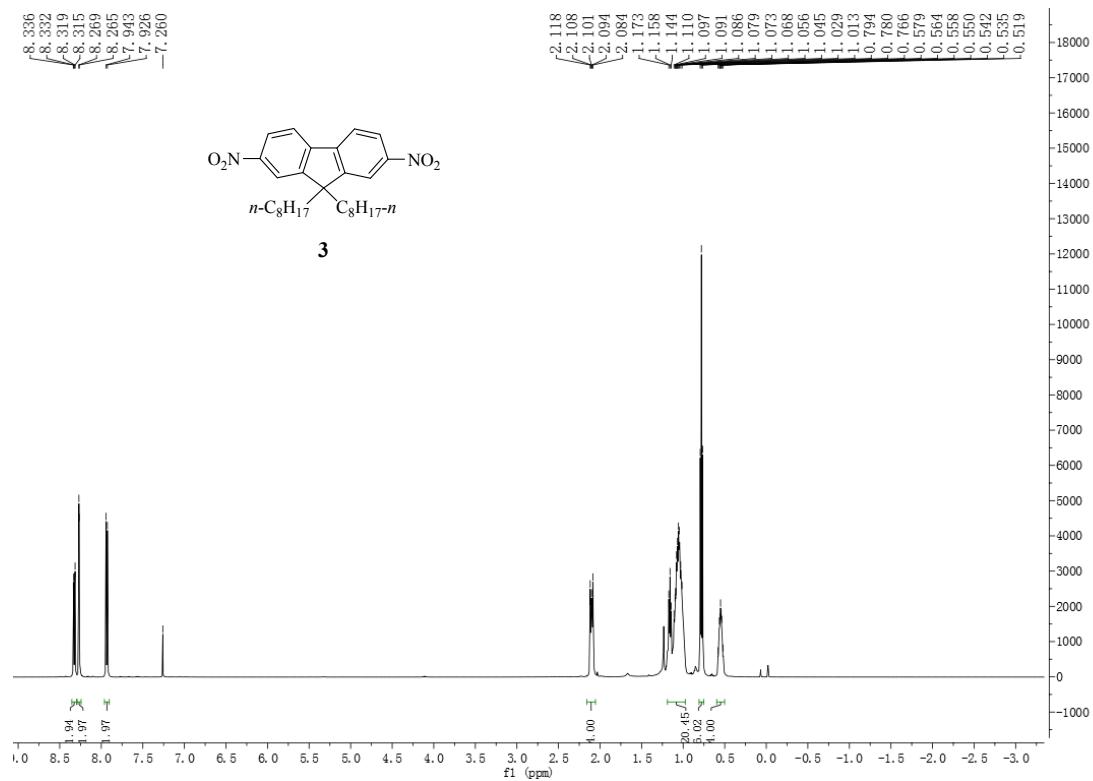


**Fig. S6** Metal specificity: the concentration of **P-1** is ( $10 \mu\text{mol}\cdot\text{L}^{-1}$ ), the concentration of  $\text{Ni}^{2+}$  is  $4 \mu\text{mol}\cdot\text{L}^{-1}$  and the other metal ions are used at  $4 \mu\text{mol}\cdot\text{L}^{-1}$ . Mix: the mixture of  $\text{Co}^{2+}$ ,  $\text{Cd}^{2+}$ ,  $\text{Hg}^{2+}$ ,  $\text{Fe}^{3+}$ ,  $\text{Ag}^+$ ,  $\text{Cu}^{2+}$ ,  $\text{Zn}^{2+}$ ,  $\text{Mn}^{2+}$ ,  $\text{K}^+$  and  $\text{Pb}^{2+}$ .

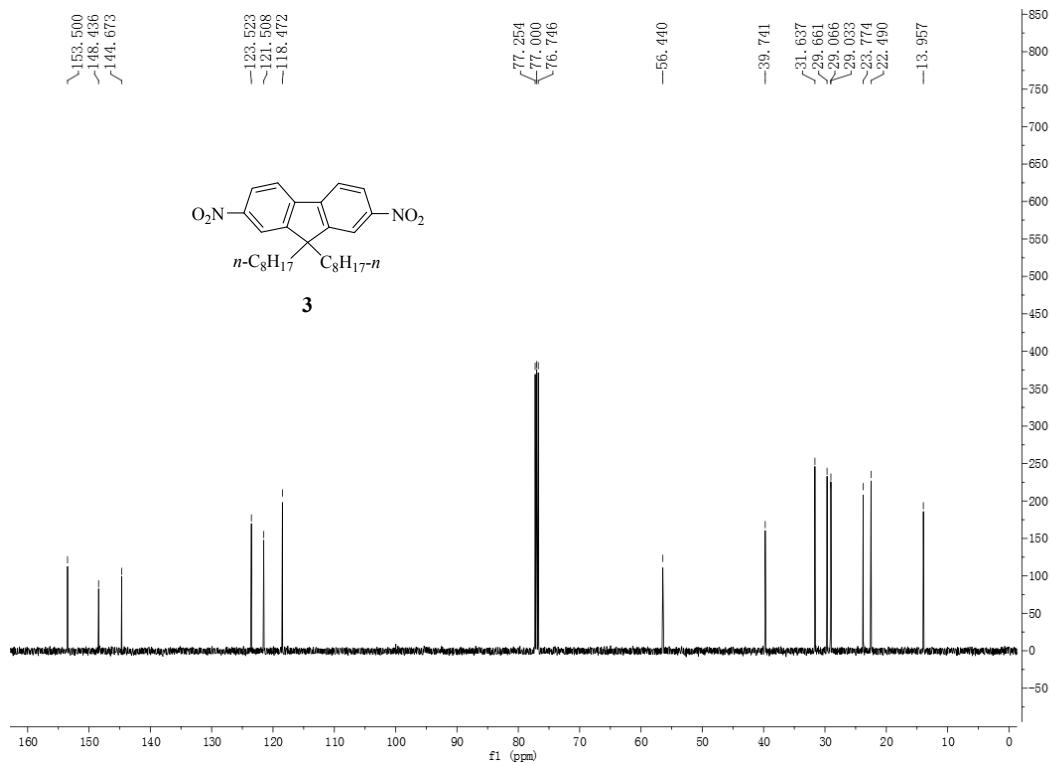


**Fig. S7** Fluorescence spectra of **P-2** ( $10 \mu\text{mol}\cdot\text{L}^{-1}$ ) in  $\text{CHCl}_3$  in the presence of in the presence of various metal ions in  $\text{CH}_3\text{CN}$  (each  $14 \mu\text{mol}\cdot\text{L}^{-1}$ ).

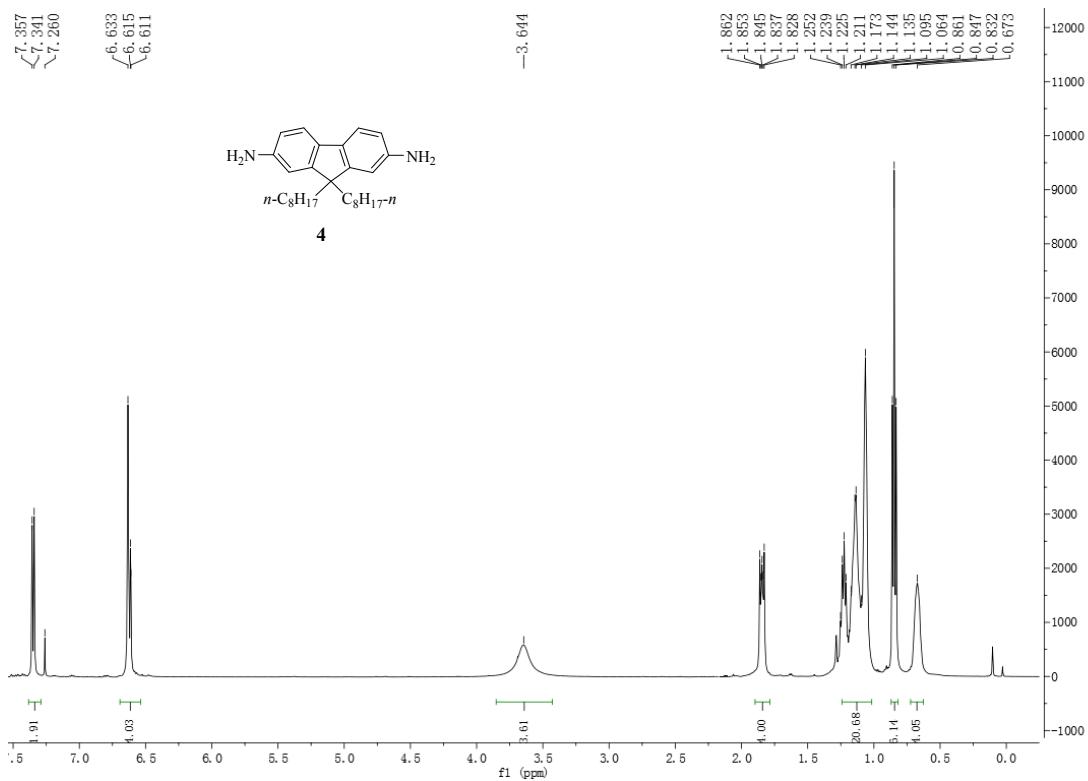
### 3. NMR spectra of the important compounds and the conjugated polymers P-1 and P-2



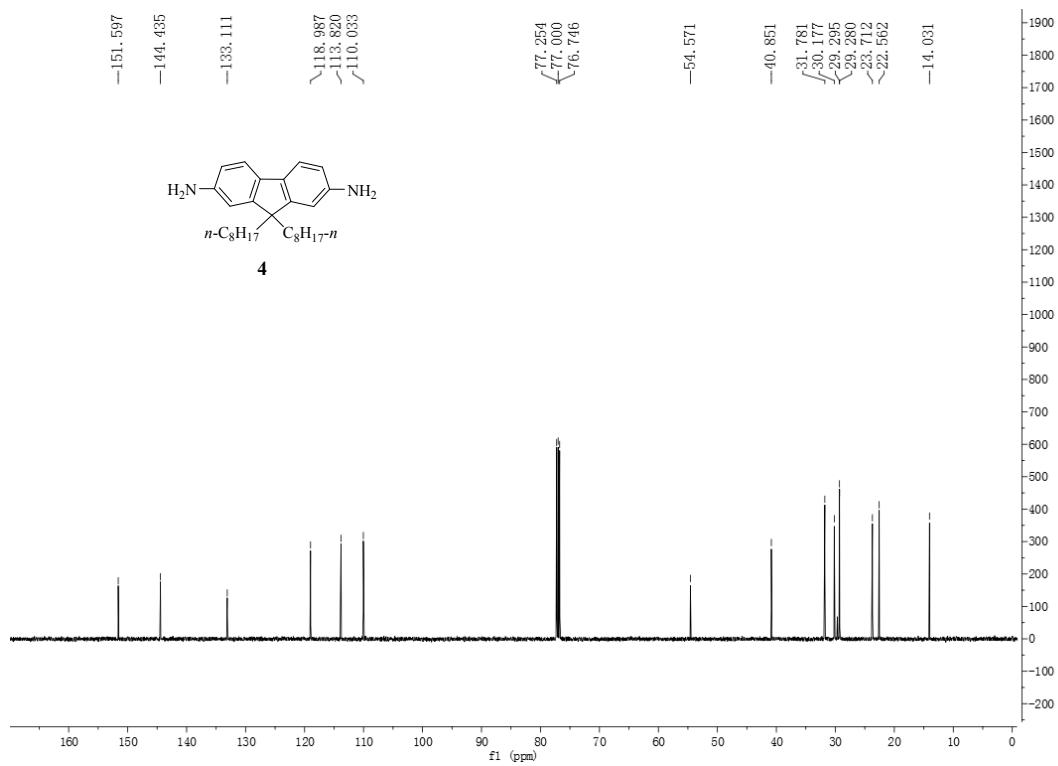
**Fig. S8**  $^1\text{H}$  NMR of compound **3** ( $\text{CDCl}_3$ , 500 MHz)



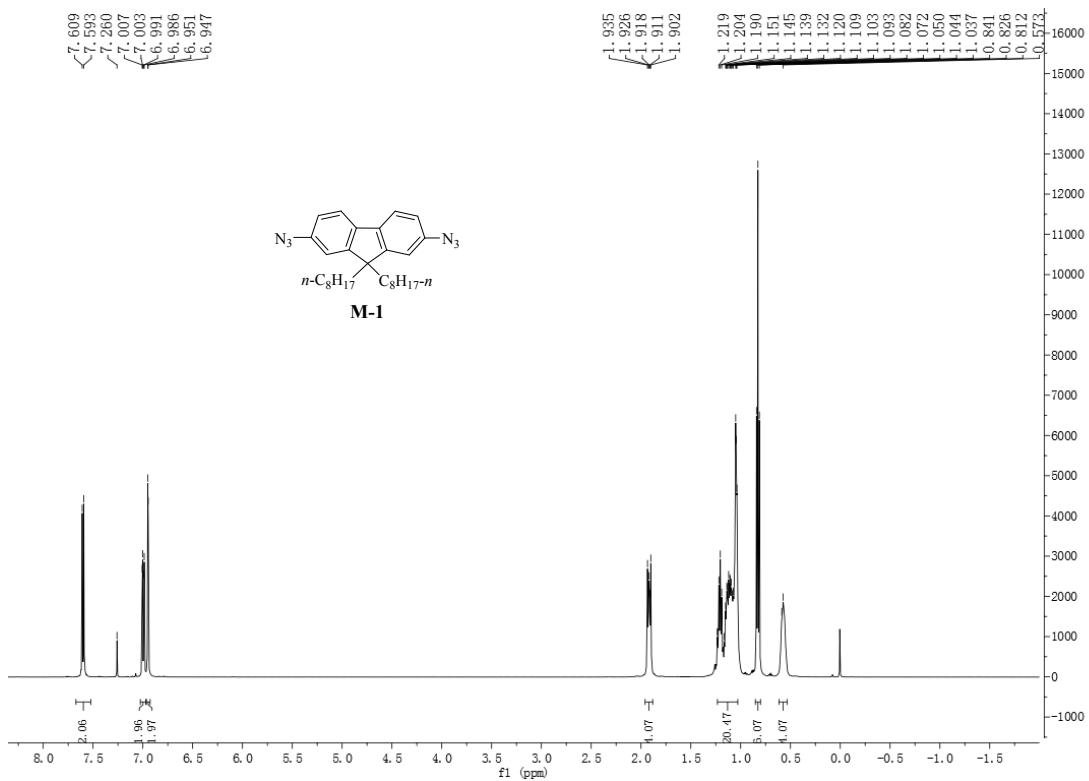
**Fig. S9**  $^{13}\text{C}$  NMR of compound **3** ( $\text{CDCl}_3$ , 125 MHz)



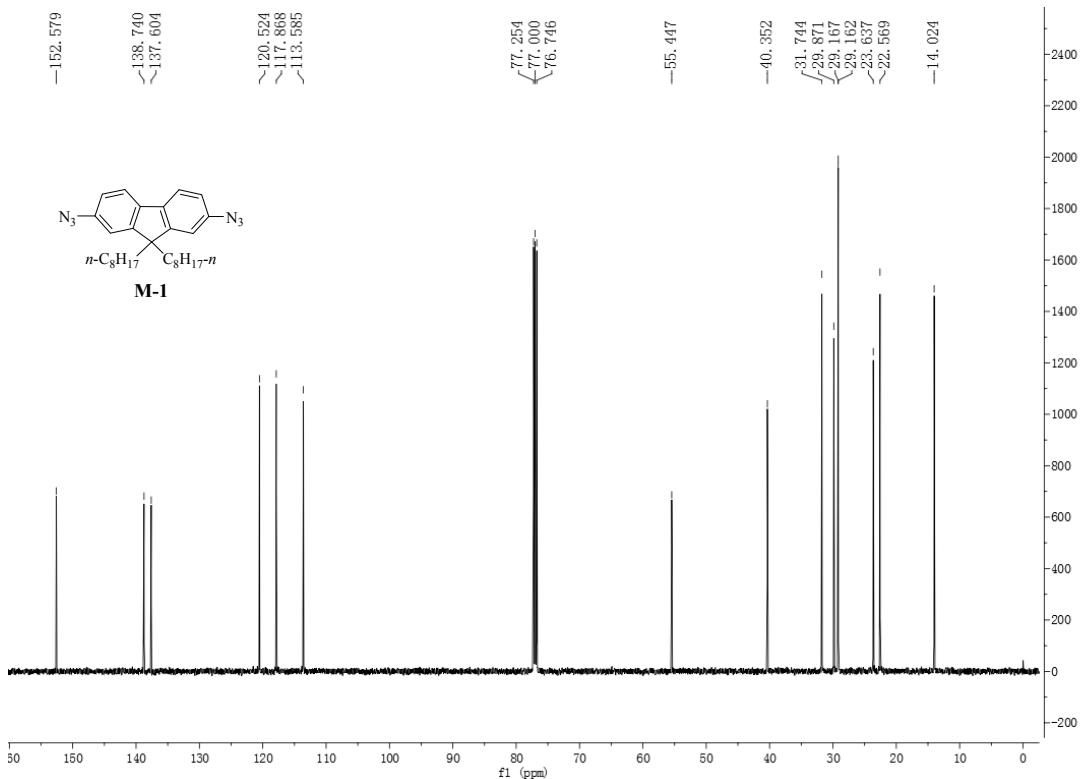
**Fig. S10**  $^1\text{H}$  NMR of compound **4** (CDCl<sub>3</sub>, 500 MHz)



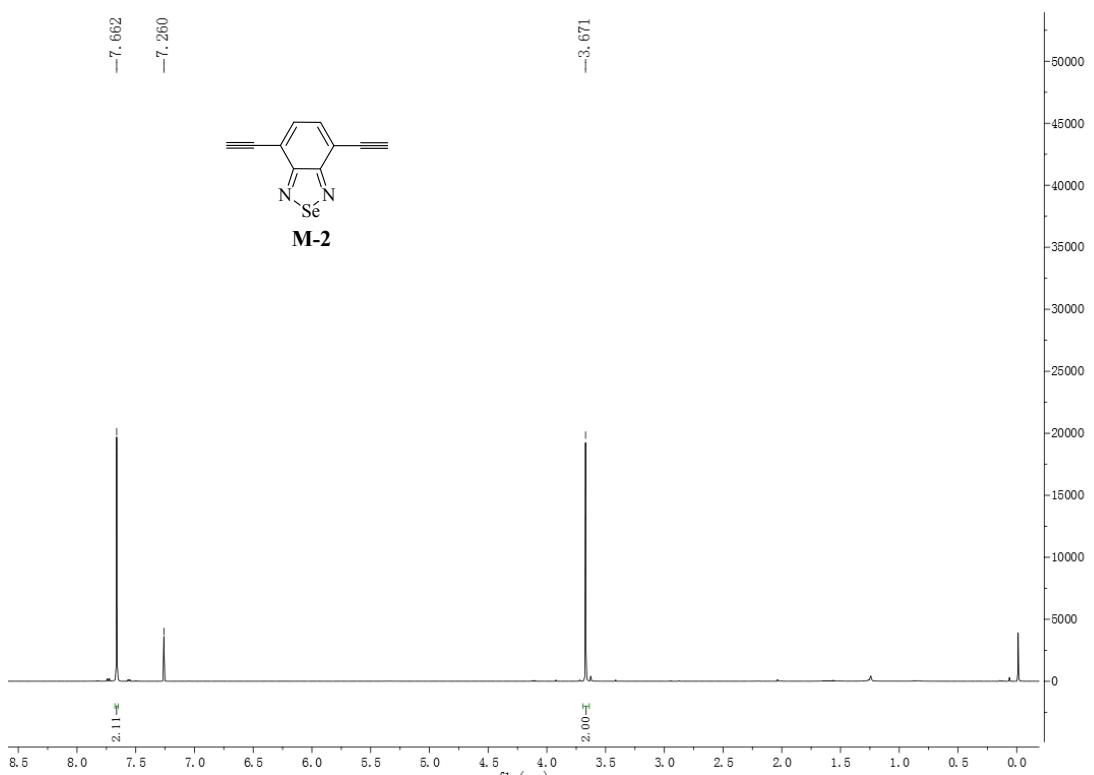
**Fig. S11**  $^{13}\text{C}$  NMR of compound **4** (CDCl<sub>3</sub>, 125 MHz)



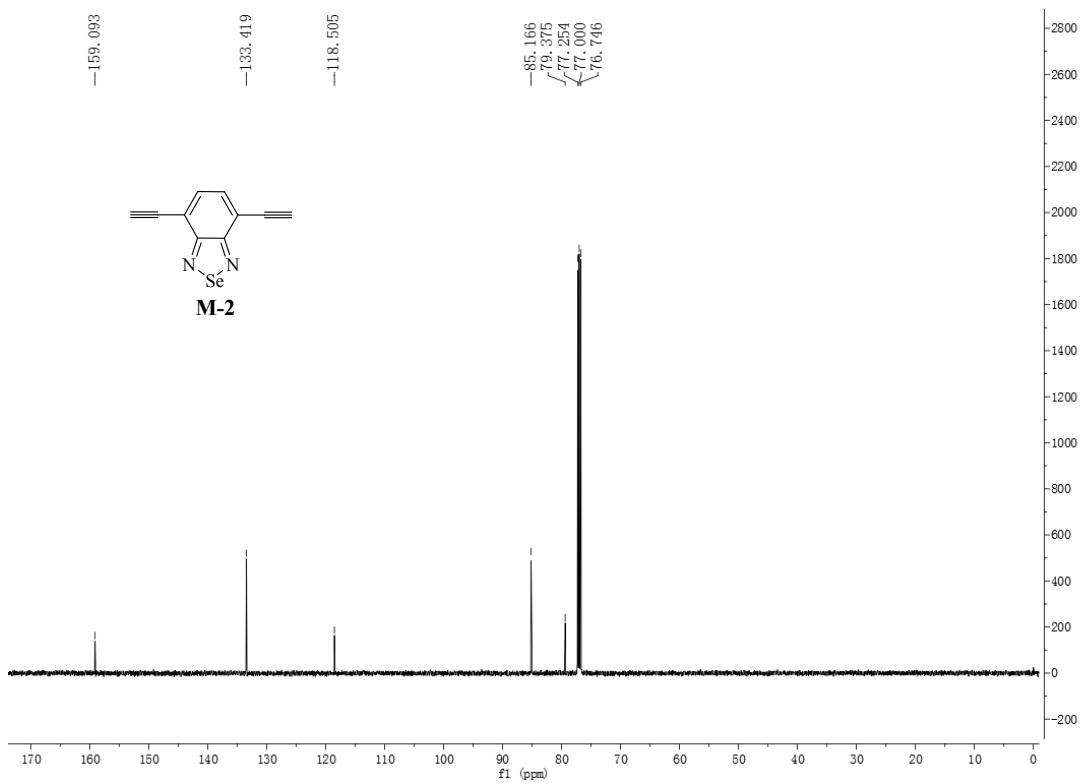
**Fig. S12**  $^1\text{H}$  NMR of compound **M-1** ( $\text{CDCl}_3$ , 500 MHz)



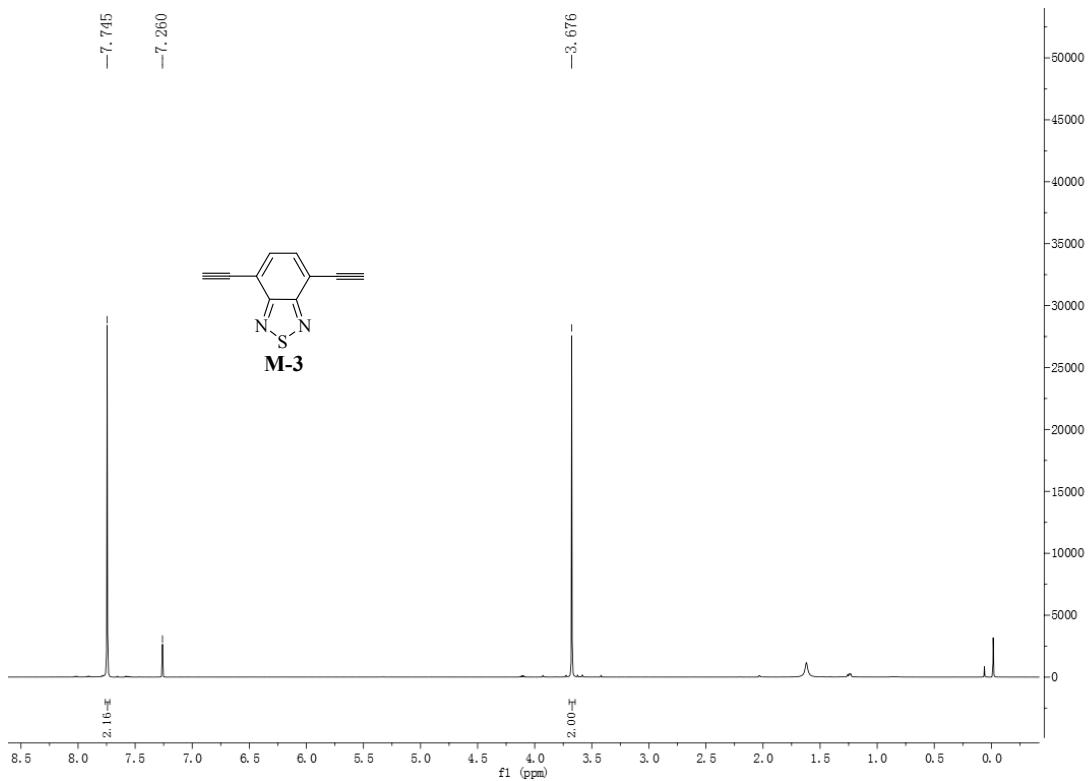
**Fig. S13**  $^{13}\text{C}$  NMR of **M-1** ( $\text{CDCl}_3$ , 125 MHz)



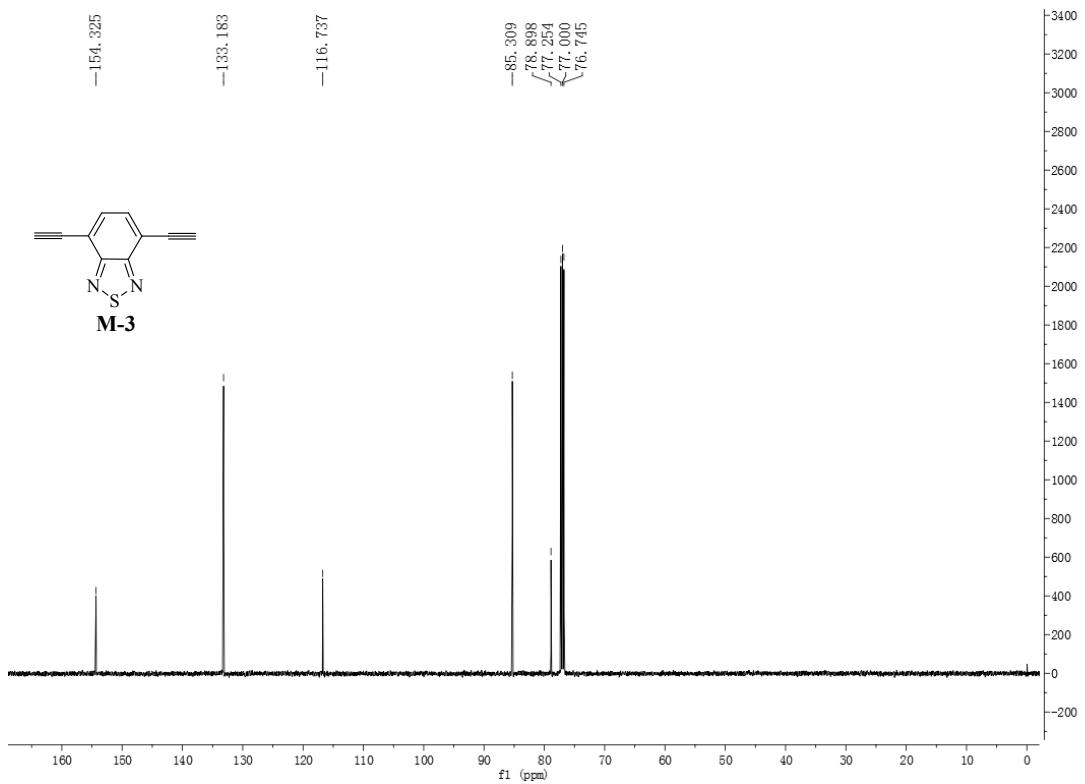
**Fig. S14**  $^1\text{H}$  NMR of **M-2** ( $\text{CDCl}_3$ , 500 MHz)



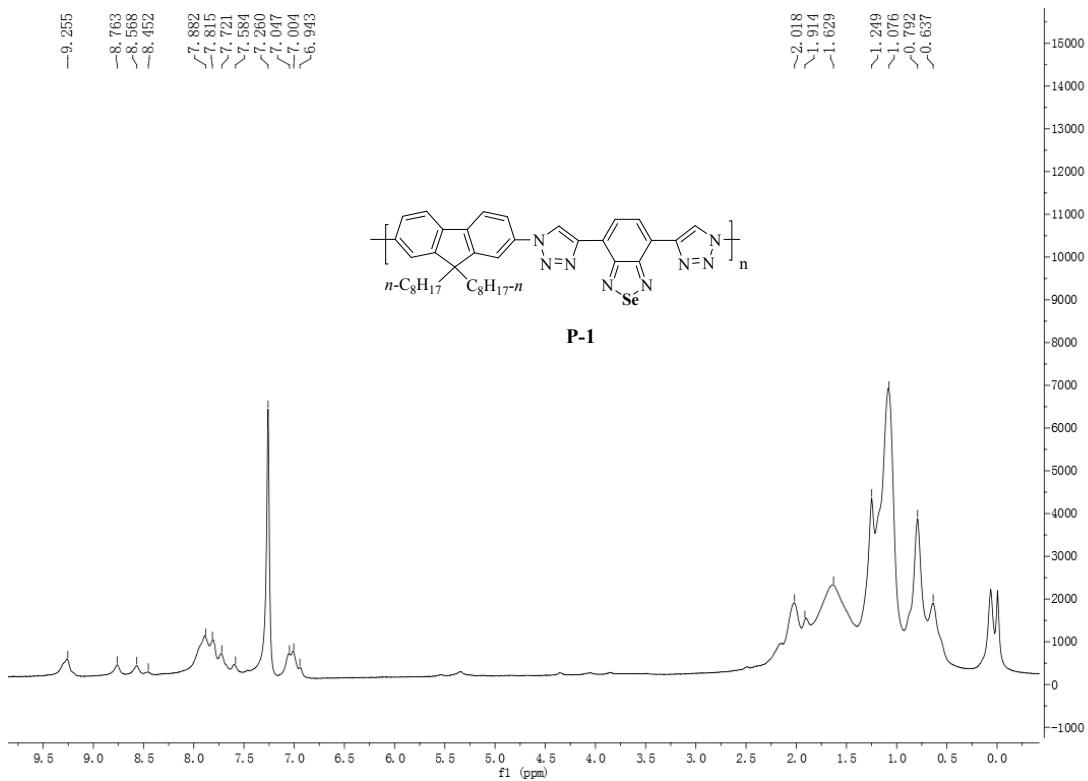
**Fig. S15**  $^{13}\text{C}$  NMR of **M-2** ( $\text{CDCl}_3$ , 125 MHz)



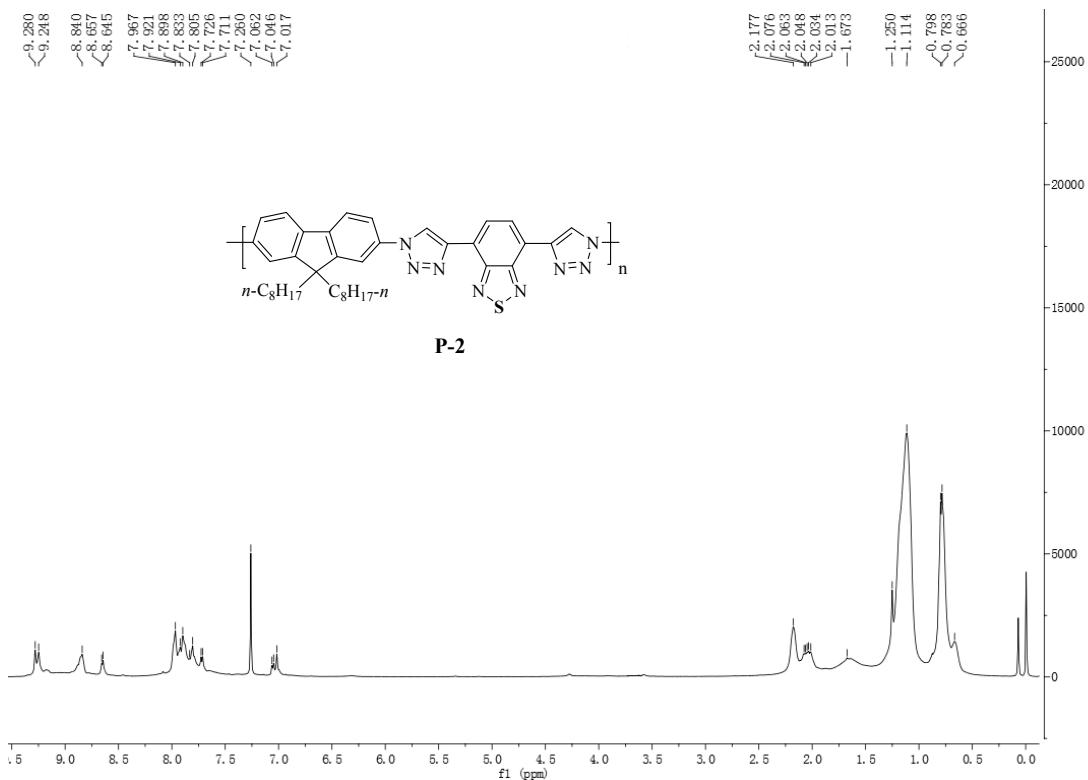
**Fig. S16**  $^1\text{H}$  NMR of **M-3** ( $\text{CDCl}_3$ , 500 MHz)



**Fig. S17**  $^{13}\text{C}$  NMR of **M-3** ( $\text{CDCl}_3$ , 125 MHz)



**Fig. S18**  $^1\text{H}$  NMR of **P-1** ( $\text{CDCl}_3$ , 500 MHz)



**Fig. S19**  $^1\text{H}$  NMR of **P-2** ( $\text{CDCl}_3$ , 500 MHz)