

Electronic Supplementary Information

Columnar discotic Pt(II) metallomesogens as luminescence multifunctional materials with chemo and thermosensor abilities

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Table S1. Selected bond distances (\AA) and angles ($^\circ$) for $[\text{Pt}(\text{pz}^{\text{R}(8,8)\text{py}}_2)_2]$ **b8**.

Pt – N1	1.975(9)	N1 – Pt – N3	79.6(4)
Pt – N3	2.033(9)	N1 – Pt – N4	179.0(4)
Pt – N3	1.976(10)	N1 – Pt – N6	100.2(4)
Pt – N3	2.018(9)	N3 – Pt – N4	99.5(4)
		N3 – Pt – N6	179.5(4)
		N4 – Pt – N6	80.8(4)

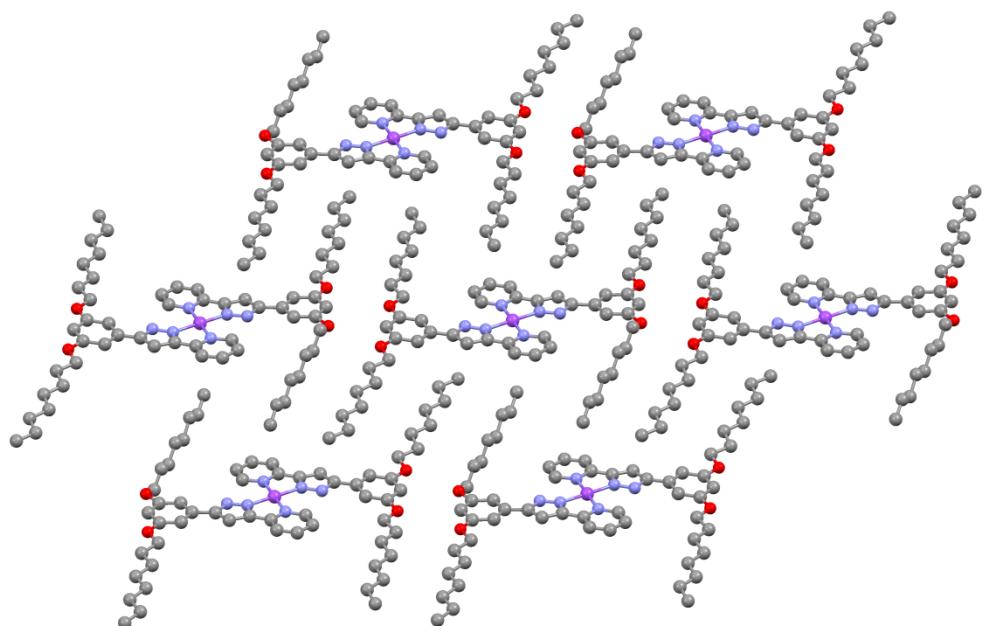


Figure S1. Packing of $[\text{Pt}(\text{pz}^{\text{R}(8,8)\text{py}}_2)_2]$ **b8** in the bc plane showing high interdigitation

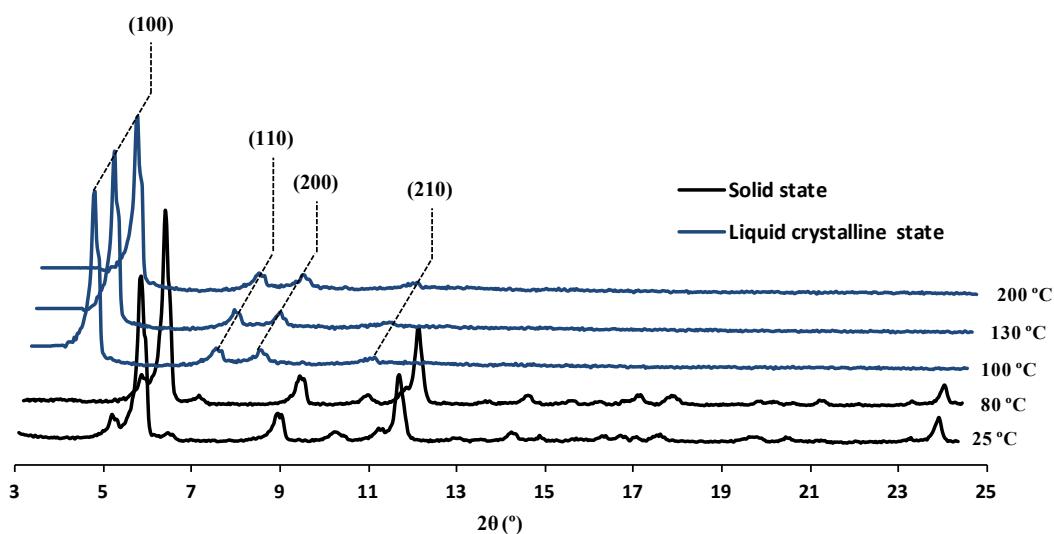


Figure S2. Powder XRD diffraction pattern for Pt(II) compound **b12** on heating

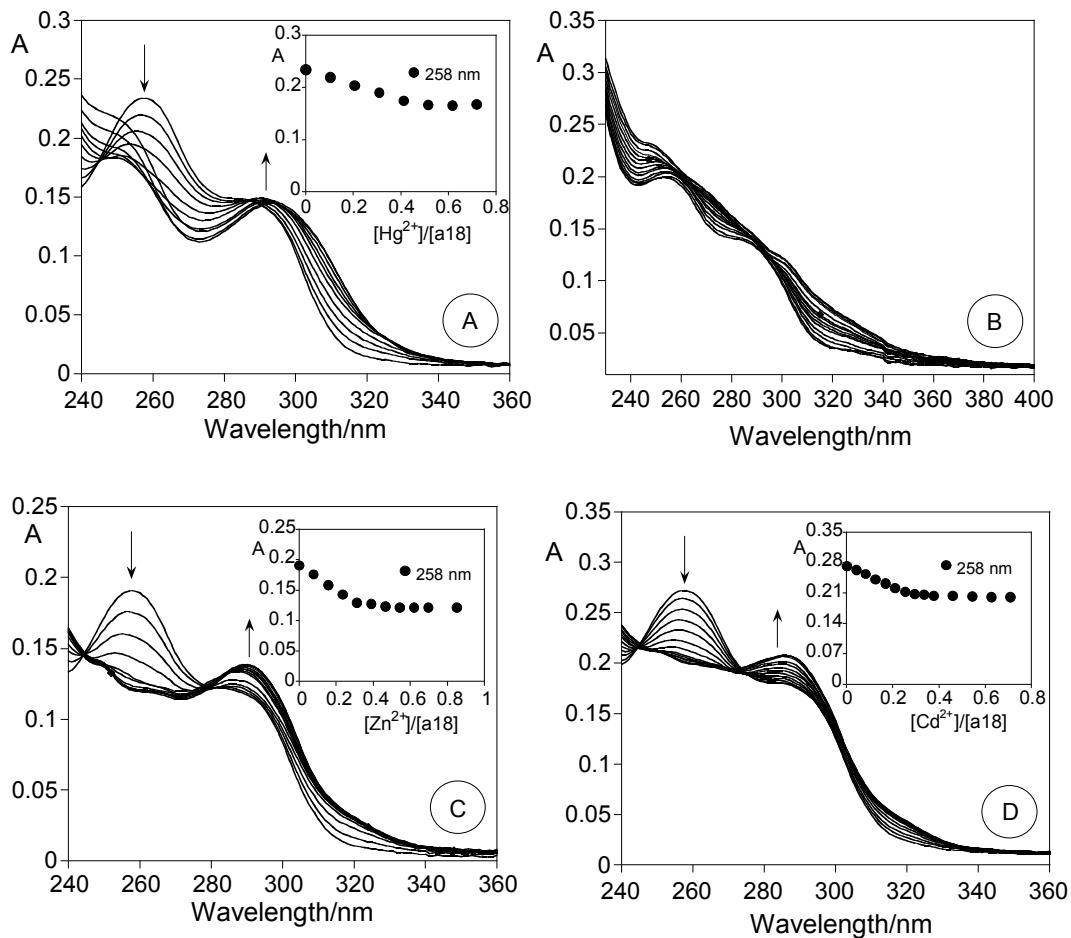


Figure S3. Absorption spectra titrations for $[\text{Hpz}^{\text{R}(18,18)\text{py}}] \mathbf{a18}$ in CH_2Cl_2 solution ($1 \times 10^{-5} \text{ M}$, $\lambda_{\text{exc}} = 286 \text{ nm}$) as a function of increasing amounts of $\text{Hg}(\text{NO}_3)_2$ (A), $[\text{Pd}(\text{CH}_3\text{CN})_4]\text{BF}_4$ (B), $\text{Zn}(\text{BF}_4)_2$ (C) and $\text{Cd}(\text{CF}_3\text{SO}_3)_2$ (D). Insets show the absorption read as a function of $[\text{Hg}^{2+}]/[\mathbf{a18}]$ (A), $[\text{Zn}^{2+}]/[\mathbf{a18}]$ (C) and $[\text{Cd}^{2+}]/[\mathbf{a18}]$ (D).

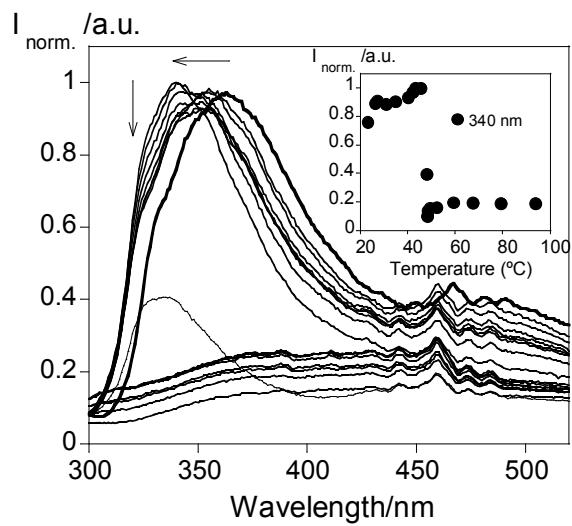


Figure S4. Fluorescence spectra of compound **a6** in solid state as a function of temperature (heating) in the range: 25-100°C. The inset represents the fluorescence intensity maximum of initial solid ($\lambda = 340$ nm) on heating.

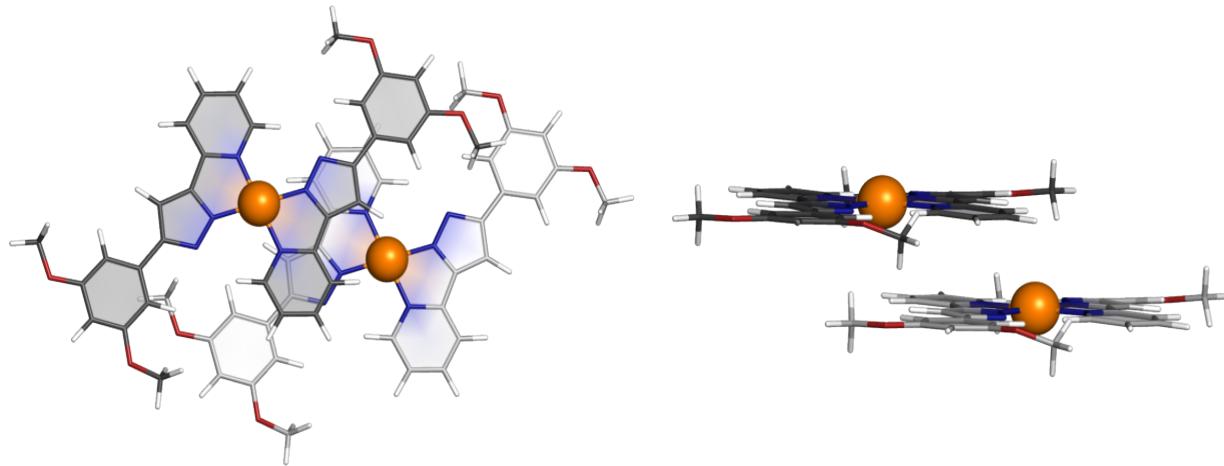


Figure S5. Simulated parallel-displaced conformation of platinum complexes

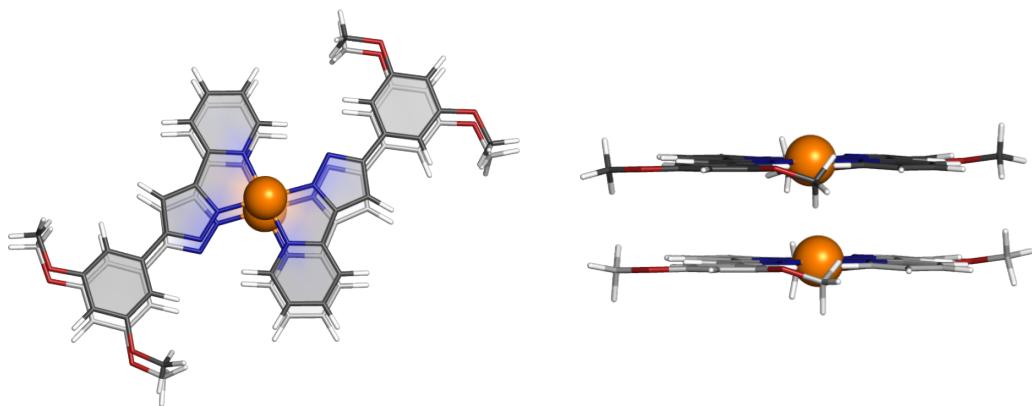


Figure S6. Simulated sandwich conformation of platinum complexes