Electronic Supplementary Information for:

Trans bis(alkylphosphine) platinum(II)-alkynyl complexes showing broadband visible light absorption and long-lived triplet excited states

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1.0 General.

All the chemicals used in synthesis are analytical pure and were used as received. Solvents were dried and distilled before used for synthesis.

Fluorescence lifetimes were measured on a OB920 luminescence lifetime spectrometer (TCSPC. Edinburgh, UK). The nanosecond time-resolved transient absorption spectra were detected by LP920 Edinburgh analytical instruments (Edinburgh Instruments, U.K.) and recorded on a Tektronix TDS 3012B oscilloscope. The lifetime values (by monitoring the decay trace of the transients) were obtained with the LP900 software. All samples in flash photolysis experiments were deaerated with argon for ca. 15 min before measurement and the gas flow is kept during the measurement.

The free energy changes of the electron transfer process (charge separation, CS), can be

calculated with the Weller equation (eq. 1 and eq. 2)

$$\Delta G^{0}_{CS} = e[E_{OX} - E_{RED}] - E_{00} + \Delta G_{S}$$

$$\Delta G_{S} = -\frac{e^{2}}{4\pi\varepsilon_{S}\varepsilon_{0}R_{CC}} - \frac{e^{2}}{8\pi\varepsilon_{0}} \left(\frac{1}{R_{D}} + \frac{1}{R_{A}}\right) \left(\frac{1}{\varepsilon_{REF}} - \frac{1}{\varepsilon_{S}}\right)$$
(Eq. 1)
(Eq. 2)

Where $\Delta G_{\rm S}$ is the static Coulombic energy, which is described by eq. 2. Where $\Delta G_{\rm S}$ is the static Coulombic energy, which is described by eq. 2. e = electronic charge, $E_{\rm OX}$ = half-wave potential for one-electron oxidation of the electron-donor unit, $E_{\rm RED}$ = half-wave potential for one-electron reduction of the electron-acceptor unit; note herein the anodic and cathodic peak potentials were used because in some cases the oxidation is irreversible therefore the formal potential $E_{1/2}$ cannot be derived; E_{00} = energy level approximated with the fluorescence emission wavelength (for the singlet excited state), or the T₁ state energy level of Bodipy. $\varepsilon_{\rm S}$ = static dielectric constant of the solvent, R_{CC} center-to-center separation distance determined by DFT optimization of the geometry, $\varepsilon_{\rm REF}$ is the static dielectric constant of the solvent used for the electron acceptor, $\varepsilon_{\rm REF}$ is the static dielectric constant of the solvent such as $\varepsilon_{\rm O}$ permittivity of free space. The solvents used in the calculation of free energy of the electron transfer is CH₂Cl₂ (ε =9.1).

2.0 NMR and HRMS spectra



Fig. S1¹H NMR of compound **1** (400 MHz, CDCl₃).



Fig. S2 ¹H NMR of compound **2** (400 MHz, CDCl₃).



Fig. S3 ¹H NMR of compound **3** (400 MHz, CDCl₃).



Fig. S4 ¹H NMR (400 MHz, CDCl₃) of complex Pt-a.



Fig. S5 ¹³C NMR (100 MHz, CDCl₃) of Pt-a.



Fig. S6 MALDI-HRMS of Pt-a.



Fig. S7 ¹H NMR (400 MHz, CDCl₃) of complex **Pt-b**.



Fig. S8 ¹³C NMR (100 MHz, CDCl₃) of **Pt-b**.



Fig. S9 MALDI-HRMS of Pt-b.









Fig. S11 ¹³C NMR (100 MHz, CDCl₃) of **Pt-1**.



Fig. S12 MALDI-HRMS of Pt-1.



Fig. S13 ¹H NMR (500 MHz, CDCl₃) of **Pt-2**.



Fig. S14 ¹³C NMR (100 MHz, CDCl₃) of **Pt-2**.



Fig. S 15 MALDI-HRMS of Pt-2.





3.0 Crystallographic data.

Table S1. Summary of unit cell, space group, data collection and structure refinementparameters for **Pt-b**.

Compound	Pt-b
Empirical formula	$C_{45}H_{72}BClF_2N_2P_2Pt$
Formula weight	982.33
Temperature (K)	173 K
Wavelength (Å)	0.71073
Volume	4964.0(9)
Space group	P21/c
a (Å)	10.0654(11)
b (Å)	36.749(4)
<i>c</i> (Å)	13.9440(15)
(°)	90
(°)	105.757(2)
(°)	90
Dx, g cm ⁻³	1.314
	4
Mu (mm ⁻¹)	2.982
Data completeness	0.994
Theta(max)	27.610
R(reflections)	0.0635(6050)
wR2(reflections)	0.1551(11457)

4.0 Nanosecond and Femtosecond Time-resolved Transient difference absorption spectra



Fig. S16 Triplet excited state decay traces of (a, b) **Pt-a** (584 nm); (c) **Pt-b** (504 nm), (d) **Pt-1** (644 nm) and (e, f) **Pt-2** (640 nm) under different atmosphere (annotated in the figures). $c = 1.0 \times 10^{-5}$ M in toluene at 20 °C.



Fig. S17 Triplet excited state decay traces of (a) **Pt-a** (584 nm), (b) **Pt-b** (504 nm), (c) **Pt-1** (644 nm), (d) **Pt-2** (640 nm). $c = 5.0 \times 10^{-6}$ M in deaerated toluene at 20 °C.



Fig. S18 Nanosecond time-resolved transient difference absorption spectra of complex **Pt-2**, after nanosecond pulsed laser excitation (λ_{ex} =503 nm). c = 1.0 ×10⁻⁵ M in deaerated toluene, 20 °C.



Fig. S19 Picosecond ultrafast transient difference absorption spectra of **Pt-2** upon femtosecond pulsed laser excitation at 640 nm. $c = 1.0 \times 10^{-5}$ M in toluene at 20 °C.

5.0 TTA upconversion details



Fig. S20 Upconversions with (a) **Pt-1** and (b) **Pt-2** upon excitation with the light of a spectrofluorometer (RF5301 PC, Shimadzu, Japan). Excitation wavelength $\lambda_{ex} = 600$ nm, 610 nm, 620 nm, *c* [sensitizers] = 1.0×10⁻⁵ M , for **Pt-1**, *c* [PBI] = 2.6×10⁻⁵ M; for **Pt-2**, *c* [PBI] = 4.0×10⁻⁵ M, in deaerated toluene. 20 °C.



6.0 Time-resolved emission spectra (TRES) of Pt-a and Pt-b

Fig. S21 TRES of **Pt-a**, **Pt-b** alone. (a) **Pt-a**, the fluorescence and phosphorescence region was measured (500 nm – 850 nm), $\tau_{\rm DF} = 13.7 \,\mu \text{s}$ monitored at 627 nm; $\tau_{\rm P} = 65.3 \,\mu \text{s}$ monitored at 805 nm. (b) **Pt-b**, the fluorescence region was measured (500 nm – 800 nm), $\tau_{\rm DF} = 189.2 \,\mu \text{s}$ monitored at 516 nm. $c = 1.0 \times 10^{-5} \,\text{M}$. In deaerated toluene. Excited with nanosecond pulsed OPO laser synchronized with spectrofluorometer (for **Pt-a** $\lambda_{\rm ex} = 560 \,\text{nm}$, for **Pt-b**, $\lambda_{\rm ex} = 503 \,\text{nm}$), 25° C.

7.0 DFT calculations



Fig. S22 Electron density maps of the frontier molecular orbitals of complex **Pt-a** based on the optimized ground state geometry. The solvent toluene was considered in the calculations (PCM model). Calculated at the B3LYP/GENCP/LANL2DZ level with Gaussian 09W.



Fig. S23 Electron density maps of the frontier molecular orbitals of complex **Pt-b** based on the optimized ground state geometry. The solvent toluene was considered in the calculations (PCM model). Calculated at the B3LYP/GENCP/LANL2DZ level with Gaussian 09W.



HOMO-6

Fig. S24 Electron density maps of the frontier molecular orbitals of complex **Pt-1** based on the optimized ground state geometry. The solvent toluene was considered in the calculations (PCM model). Calculated at the B3LYP/GENCP/LANL2DZ level with Gaussian 09W.

Table S2: Excitation Energies (eV) and corresponding Oscillator Strengths (*f*), main configurations and CI coefficients of the Low-lying Electronically Excited States of complex **Pt-a**, Calculated by TDDFT//B3LYP/LANL2DZ, based on the DFT//B3LYP/LANL2DZ Optimized Ground State Geometries

	TDDFT//B3LYP/LANL2DZ					
	Electronic transition	Energy [eV/nm]	f^{b}	Composition ^c	CId	Character
Singlet	$S_0 \rightarrow S_1$	2.36 / 525	0.4575	$H \rightarrow L$	0.68350	MLCT
	$S_{0} \!\!\rightarrow \!\! S_{3}$	3.06 / 405	0.5386	$H-2 \rightarrow L$	0.6273	MLCT
Triplet	$S_0\!\!\rightarrow\!\!T_1$	1.42 / 867	0.6677	$H\!\rightarrow\!\!L$	0.0000	MLCT

^{*a*} Only the selected low-lying excited states are presented. ^{*b*}Oscillator strengths. ^{*c*} Only the main configurations are presented. ^{*d*} The CI coefficients are in absolute values. ^{*e*} No spin-orbital coupling effect was considered, thus the *f* values are zero.

Table S3: Excitation Energies (eV) and corresponding Oscillator Strengths (*f*), main configurations and CI coefficients of the Low-lying Electronically Excited States of complex **Pt-b**, Calculated by TDDFT//B3LYP/LANL2DZ, based on the DFT//B3LYP/LANL2DZ Optimized Ground State Geometries

		TI	DDFT//B3LY	P/ LANL2DZ		
	Electronic transition	Energy [eV/nm] ^a	$f^{\ b}$	Composition ^c	CI ^d	Character
Singlet	$S_0 {\rightarrow} S_1$	2.26 / 548	0.0009	$H \!\rightarrow\! L$	0.70296	MLCT
	$S_0 \rightarrow S_2$	2.80 / 442	0.5662	$H-1 \rightarrow L$	0.69631	ILCT
	$S_0 \rightarrow S_6$	3.61 / 342	1.0131	$H \rightarrow L+1$	0.68115	ILCT
Triplet	$S_0\!\!\rightarrow\!\!T_1$	1.53 / 809	0.0000	$H \!\rightarrow\! L$	0.71025	MLCT

^{*a*} Only the selected low-lying excited states are presented. ^{*b*}Oscillator strengths. ^{*c*} Only the main configurations are presented. ^{*d*} The CI coefficients are in absolute values. ^{*e*} No spin-orbital coupling effect was considered, thus the *f* values are zero.

Optimized ground state geometry of Pt-a

0 1	
С	2.75414 9.89495 -11.42985
Р	0.6833 10.54654 -9.07646
С	-0.0519 12.19226 -8.75219
Н	-0.12324 12.3637 -7.66857
Н	-1.04555 12.22832 -9.19294
Н	0.56765 12.96851 -9.1959
С	-0.33914 9.36561 -8.11622
Н	0.09851 8.37027 -8.16805
Н	-1.33985 9.34057 -8.5394
Н	-0.38963 9.68161 -7.06425
С	2.30514 10.57203 -8.21397
Н	2.7684 9.5913 -8.2715
Н	2.96267 11.29724 -8.68593
Н	2.1551 10.84456 -7.15856
Р	0.77976 9.50729 -13.70209
С	1.49523 7.84956 -14.02661
Н	1.56289 7.67708 -15.11024
Н	0.86678 7.08143 -13.58163
Η	2.48979 7.79964 -13.58807
С	-0.83586 9.50656 -14.56902



Pt-a

Н	-1.28546 10.49304 -14.50722
Н	-1.50434 8.78746 -14.10366
Н	-0.68475 9.23816 -15.62551
С	1.82681 10.67343 -14.65332
Н	1.40391 11.67467 -14.60119
Н	2.82641 10.68305 -14.2252
Н	1.87864 10.35888 -15.70551
C	3.07783 0.81/35 -11.//808
C	6 47524 8 62071 -11 22427
C	6 26076 10 85272 -11 74106
C	6.300/0 $10.032/3$ $-11./4190$
C	7 79118 10 49777 -11 75581
C C	2 00120 11 10122 11 02600
C	8.92189 11.12188 -11.98093
C	10.160/2 10.4581/ -11.95/83
C	11.51057 10.93135 -12.14378
C	11.50974 8.66708 -11.78139
N	7.73264 9.05479 -11.42622
N	10.22369 9.07113 -11.71195
В	9.00314 8.15373 -11.43693
F	9.09003 7.6044 -10.14856
F	8.96414 7.11441 -12.38901
C	5.77922 12.23493 -11.96737
Н	6.12727 12.93793 -11.21099
Н	6.0611 12.63398 -12.94014
Н	4.69432 12.16509 -11.91331
С	11.92321 7.24347 -11.6109
Н	13.00051 7.17069 -11.72566
Н	11.6235 6.87576 -10.63008
Н	11.42765 6.61771 -12.35683
С	12.00375 12.3204 -12.43055
Н	11.71912 13.01613 -11.64288
Н	13.08887 12.28581 -12.49968
Н	11.60043 12.70309 -13.36762
C	6.1238 7.16668 -11.17073
н	6 55825 6 76803 -10 26287
Н	5 04058 7 06117 -11 15245
н	6 52004 6 50247 -12 01206
C	8 88124 12 50161 -12 26447
C	8 7072 12 05105 -12 58262
C	8 09689 19 51997 -11 91009
C	8 75960 14 49488 -19 84789
C C	8 80466 14 88556 -11 47652
U U	0.09400 14.00550 -11.4/052
	9.001/1 13.15040 -10.10952
П	8.0829/ 14.7/324 -14.8/151
П	8.93034 15.59247 -10.0503
C	8.80519 15.34421 -12.79488
H	8.77592 16.40823 -12.99969
H	8.75186 12.3405 -14.39972
	-1.53797 10.17764 -11.36275
Pt	0.73682 10.02781 -11.39831
C	12.30916 9.8027 -12.00112
Н	13.37791 9.80736 -12.05265

Optimized ground state geometry of Pt-b

01	
С	-10.13944 2.2579 -2.16761
Pt	-8.1705 2.20045 -1.72852
Р	-7.85734 1.17746 -3.82795
С	-8.43274 2.23467 -5.21897
Н	-8.33461 1.71101 -6.17597
н	-7 82044 2 15240 -5 24760
Ч	-0.47007 + 2.40052 = 0.468
II C	-9.4/90/2.49952 -5.0400
	-0.13/93 0./1952 -4.20009
H	-5./38// 0.01402 -3.5555
H	-5.50766 1.6124 -4.27614
H	-6.11377 0.26632 -5.28534
C	-8.81069 -0.38412 -4.01829
Н	-8.45545 -1.11958 -3.29031
Н	-9.86525 -0.17641 -3.81983
Н	-8.69658 -0.79212 -5.02837
Р	-8.50386 3.22556 0.36718
С	-10.22965 3.66808 0.81746
Н	-10.26093 4.13495 1.80758
Н	-10.84808 2.76697 0.81628
Н	-10.63653 $A.35645$ 0.07252
C	-7.02526 2.17060 1.76580
U U	697004 + 100066 + 60660
и П	8 50460 1 05180 1 70040
п	
H	-8.0408/ 2./0336 2./2095
C	-7.56762 4.79779 0.55902
H	-6.50909 4.6035 0.36778
H	-7.92666 5.52799 -0.17239
H	-7.69191 5.20683 1.5675
C	-11.33569 2.27842 -2.44748
C	-19.4603 0.68528 -4.45634
C	-17.30848 0.3463 -3.72964
C	-17.48214 1.70445 -4.12917
C	-16.57626 2.80179 -4.14053
С	-17.02243 4.07318 -4.59789
С	-16.34047 5.31997 -4.72264
С	-18.52017 5.54122 -5.41047
Ν	-18.8091 1.86774 -4.56561
Ν	-18.34872 4.2528 -5.02943
В	-19.45136 3.17392 -5.07945
F	-10.01138 3.00747 -6.30608
F	-20 52282 2 54065 -4 26205
C	-16 08076 -0 24026 -2 18752
с н	
Ч	-15.250/2 -0.34209 -3.90049
П Ц	-15./155 0.15040 $-2.2/142$
II C	-10.33020 - 1.30094 - 2.95450
U U	10,0000,00000,0000,0000,0000,0000,0000,0000
11 U	-19./0593 /.142/0 -0.15304
11 11	-20.130/7 5.53007 $-0.81/14$
п	-20.005/5 5.94729 -5.17624
C	-14.91571 5.66497 -4.40362



Η	-14.20034 5.07223 -4.98543
Н	-14.74312 6.72191 -4.63187
Η	-14.67036 5.50267 -3.34772
С	-20.88915 0.49949 -4.82831
Н	-21.05726 0.78306 -5.87374
Н	-21.18171 -0.543 -4.68168
Н	-21.53497 1.14741 -4.22327
C	-15.17005 2.62307 -3.66582
C	-14.83236 2.85234 -2.32255
С	-14.15483 2.22299 -4.54898
C	-13.51984 2.68729 -1.8737
C	-12.84102 2.05688 -4.10396
Н	-14.40184 2.0425 -5.59195
Н	-13.27901 2.8695 -0.82949
Н	-12.06893 1.74699 -4.80359
С	-12.51996 2.28873 -2.76429
Н	-15.60882 3.16206 -1.6279
Cl	-5.94515 2.13276 -1.23695
С	-18.54917 -0.29282 -3.93665
Н	-18.7591 -1.32294 -3.73749
C	-17.28045 6.23986 -5.23265
Н	-17.08946 7.27117 -5.44433

Optimized ground state geometry of Pt-1

01	
С	-4.84175 -0.21047 0.04298
С	-6.06742 -0.30014 0.0438
С	-10.09315 -0.64532 -0.00463
Pt	-8.0823 -0.47139 0.02119
Р	-7.91367 -0.73491 -2.31114
С	-7.16563 0.72549 -3.14378
Н	-7.03472 0.53954 -4.21525
Н	-6.19653 0.93284 -2.68306
Η	-7.81326 1.59646 -3.00525
С	-6.83072 -2.13883 -2.80204
Н	-7.26421 -3.07663 -2.44168
Η	-5.85181 -2.00473 -2.33532
Н	-6.72052 -2.18669 -3.89074
С	-9.47118 -1.03012 -3.24369
Н	-9.94071 -1.94945 -2.8839
Н	-10.1648 -0.20417 -3.06753
Н	-9.26684 -1.11831 -4.31606
Р	-8.23693 -0.2035 2.35298
С	-9.01045 -1.64573 3.19409
Н	-9.13335 -1.45355 4.26545
Н	-8.37981 -2.52919 3.05654
Н	-9.98499 -1.83797 2.73809
С	-6.66981 0.06542 3.2759
Н	-6.17926 0.96842 2.9037
Н	-5.99609 -0.77795 3.10534
Η	-6.86784 0.169 4.3481
С	-9.29151 1.22341 2.84126



Η	-8.8346 2.1519 2.48575
Η	-10.27025 1.1134 2.36741
Η	-9.40785 1.27185 3.92933
С	2.50498 -1.16181 0.07994
С	2.52642 1.11513 -0.026
С	3.35594 -0.02017 0.02007
С	1.18014 0.63664 0.00358
C	-0.05579 1.30675 -0.02259
Ċ	-1.27019 0.59902 0.01349
Ċ	-2.63097 1.03526 -0.01272
Č	-2.53794 -1.23916 0.10889
N	122120 - 0.7641 - 0.06844
N	-1.26728 - 0.8010 - 0.08722
B	-0.00852 -1.71124 -0.12074
F	-0.000033 - 1.71124 - 0.13074
F	-0.00503 -2.5/703 -0.90730
r C	0.01292 - 2.44032 - 1.3104
U U	3.04102 2.5211 -0.09220
	2./3310 3.03134 -1.01198
п	2.68237 3.13426 0.74152
H	4.1346 2.50379 -0.05952
C	-2.90376 -2.68317 0.19032
H	-3.99068 -2.78968 0.21218
H	-2.50084 -3.23385 -0.66677
H	-2.47502 -3.13974 1.08902
C	-3.19024 2.42344 -0.09173
H	-2.87394 2.94557 -1.00156
Н	-4.28313 2.36935 -0.09007
Н	-2.87563 3.04635 0.75282
C	2.91577 -2.59433 0.1494
Η	2.47405 -3.16365 -0.67515
Н	4.00449 -2.67046 0.10354
Н	2.55884 -3.05396 1.07795
С	-0.07931 2.80084 -0.08271
С	-0.10247 3.55418 1.09928
С	-0.07883 3.4652 -1.31689
С	-0.1243 4.94943 1.04732
С	-0.10157 4.86077 -1.36733
Н	-0.06112 2.88638 -2.23629
Н	-0.14158 5.52235 1.97044
Н	-0.10156 5.36455 -2.33007
С	-0.12411 5.60574 -0.18595
Н	-0.14147 6.69134 -0.22601
Н	-0.10294 3.04413 2.05869
С	-3.42424 -0.12489 0.04585
C	4.77544 -0.06087 0.0108
C	6.00339 -0.11111 0.00689
Ċ	10.04012 -0.27809 -0.04907
Pt	8.02361 -0.10/65 -0.02176
P	8.14459 -0.620 2 28758
Ĉ	0.08067 0.6580 2.20/00
й	0 18674 0 20844 4 26058
H	8 57/85 1 62017 2 111/6
н	10 08060 0 75114 0 74007
п С	6 55655 0 75000 0.0440
C	0.55055 - 0.75392 - 3.20443

Н	5.9563 -1.56511 2.78488
Н	5.9948 0.17595 3.08594
Н	6.7386 -0.94253 4.26776
С	9.0193 -2.19859 2.68421
Н	8.45792 -3.04229 2.27171
Н	10.00769 -2.1762 2.21809
Н	9.122 -2.32878 3.76693
Р	7.8886 0.23498 -2.33151
С	9.46893 0.32868 -3.2667
H	9.27928 0.52357 -4.32756
Н	10.01035 - 0.61464 - 3.15847
н	10,0010,0000,0000,0000,0000,0000,0000,
C	6 0106 -1 02652 -2 22244
н	5.021 -1.10411 -2.77264
и П	7.40400 - 2.00821 - 2.14005
и П	7.40499 - 2.00031 - 3.14095
II C	0.01250 - 0.7/899 - 4.29254
	7.04130 1.82019 -2./2531
п	0.05403 1.01520 -2.25544
H	7.61976 2.65421 -2.31663
H	6.93571 1.95048 -3.80773
C	-11.31824 -0.7417 -0.01019
C	11.3471 -0.3366 -0.11598
C	18.70079 -2.32427 0.05264
С	19.02852 -0.01206 0.08421
C	17.58826 -0.31462 0.03727
C	16.42317 0.54182 0.01393
С	15.09245 -0.01064 -0.02494
С	13.76647 0.62232 -0.05308
С	13.55036 -1.70901 -0.07817
Ν	17.44023 -1.7456 0.01967
Ν	14.9024 -1.43986 -0.04227
В	16.06054 -2.54105 -0.03148
F	16.01107 -3.34259 -1.22582
F	15.94906 -3.38558 1.12709
С	19.77205 1.35307 0.12051
Н	19.54662 1.97615 -0.79041
Н	19.47141 1.96831 1.01477
Н	20.88542 1.19357 0.16604
С	12.99463 -3.13432 -0.10873
Н	11.87331 -3.09356 -0.1638
Н	13.30322 -3.70331 -0.90527
Н	13.30456 -3.70712 0.81076
C	13.0834 2 11744 - 0.05075
н	13.40034 - 2.11744 - 0.03073 12.85020 - 2.67122 - 0.02100
н	1228876 221525 -0.10885
н	12.200/0 2.21333 0.10003
C	13./5931 2.04200 0.00240
U U	18.9432 - 3.04909 - 0.04/00
и П	10.4/030 - 4.31035 - 0.00230
11 U	20.04102 - 4.00/32 0.053/4
п С	10.45909 - 4.32502 0.94011
	10.03093 2.00879 0.02933
	10.00393 2.78773 1.2782
C	16.77973 2.7965 -1.2061
C	16.84337 4.21253 1.29029

C	16.95831 4.22142 -1.19228
Н	16.75481 2.24362 -2.17665
Н	16.86762 4.75878 2.26464
Н	17.07189 4.77476 -2.15638
С	16.99102 4.93223 0.05574
Н	17.12992 6.04078 0.06586
Н	16.54941 2.22805 2.23852
С	12.81665 -0.45019 -0.08432
С	19.72185 -1.27921 0.09285
Н	20.82285 -1.44207 0.12553
С	-13.67912 -1.86832 -0.11489
С	-13.6259 0.46059 0.05276
С	-15.02092 -0.00719 -0.00363
С	-16.27907 0.70509 0.03111
С	-17.53576 0.00213 -0.03279
С	-18.92738 0.47411 -0.01522
С	-18.86743 -1.8609 -0.17821
Ν	-14.99943 -1.44212 -0.10578
Ν	-17.55619 -1.43605 -0.1343
В	-16.27626 -2.39102 -0.19617
F	-16.24635 -3.12142 -1.43593
F	-16.27239 -3.30925 0.91054
С	-13.04809 1.89968 0.16611
Н	-13.35718 2.54397 -0.70463
Н	-13.40757 2.42151 1.09728
Н	-11.92318 1.8708 0.19543
С	-19.25159 -3.33775 -0.29053
Н	-20.37051 -3.42695 -0.33643
Н	-18.80021 -3.80258 -1.21182
Н	-18.86441 -3.92269 0.59126
С	-19.45921 1.91356 0.07851
Н	-19.08802 2.56696 -0.76076
Н	-20.58319 1.88139 0.033
Н	-19.16059 2.42128 1.03935
C	-13.25872 -3.35072 -0.21296
Ĥ	-13.68491 -3.81661 -1.14584
Н	-12.13988 -3.45713 -0.22703
Н	-13.67167 -3.9325 0.65892
C	-16.2526 2.24238 0.13645
C	-16.28843 2.88575 1.4258
Ċ	-16.20664 3.05373 -1.05418
Č	-16.27807 4.31859 1.5219
Č	-16.19713 4.48649 -0.95635
Ĥ	-16.17877 2.55062 -2.05586
Н	-16.30578 4.80583 2.52696
Н	-16.16208 5.10/03 -1.8863/
C	-16.2323 5.12181 0.33167
Ĥ	-16.22404 6.2365 0.40715
H	-16.32371 2.26113 2.35165
Ĉ	-12.78705 -0.71282 -0.0184
č	-10.74445 - 0.60067 - 0.10482
й	-20 812/1 -0 76252 -0 08/51

Optimized ground state geometry of Pt-2

01	
С	4.90307 0.02071 1.2759
C	5.9925 -0.18279 1.81718
C	9.55239 -0.85907 3.5861
Pt	7 77132 -0 51506 2 70000
P	6 60165 -0 20270 4 77160
C	5.09103 0.202/9 4.7/109
	5.95091 1.4/240 4.94509
п	5.39914 1.5509/ 5.00/31
п	5.2/4// 1.04848 4.10445
H	6.74325 2.2266 4.92072
C	5.27774 - 1.35221 5.02563
H	5.64501 -2.38271 5.04348
H	4.58146 -1.24459 4.19006
Н	4.76245 -1.13603 5.96766
C	7.71216 -0.40504 6.28703
Н	8.11527 -1.42045 6.32085
Н	8.55429 0.29071 6.25635
Н	7.1103 -0.21853 7.1828
Р	8.86747 -0.82474 0.63879
С	9.61428 -2.497 0.46761
Н	10.171 -2.58064 -0.47194
Н	8.82451 -3.25392 0.48749
Н	10.28637 - 2.66060 - 1.31203
C	7 85678 -0 62222 -0 88265
н	7.44582 0.2008 -0.01527
и П	7.44302 0.39000 0.9133/
и П	2 46616 0 50000 1 5560
п С	8.40010 -0./9903 -1.//03
	10.2/902 $0.3299/$ $0.39//3$
п	9.90//3 1.35888 0.3/283
H	10.96496 0.2263 1.24224
H	10.8058 0.11345 -0.53785
C	-2.17243 0.09885 -1.95549
C	-1.75096 2.34458 -2.18446
C	-2.71146 1.34054 -2.42937
C	-0.62761 1.69347 -1.5941
С	0.60126 2.21205 -1.09637
С	1.5491 1.32641 -0.51156
С	2.84085 1.56551 0.04463
С	2.31363 -0.66372 0.20508
Ν	-0.91469 0.31801 -1.50351
Ν	1.26985 -0.04731 -0.39921
В	-0.0282 -0.75834 -0.83783
F	-0.65702 -1.37021 0.259
F	0.26408 - 1.73853 - 1.70083
C	-104888 270484 - 251222
ч	-1.02002 4.4247 -1.62214
Ч	1,74070 4,404/ -1,04014 -1,18004 4,17880 -0,10561
11 U	-1.10224 $4.1/009$ -3.19501
11 C	-2.9230/ 3.92354 -2.994/4
	2.34/42 -2.1229 0.49502
H	3.29205 -2.3871 0.97646
H	1.51312 -2.40887 1.14657
H	2.23355 -2.7039 -0.42789
C	3.6019 2.85201 0.17281



Н	3.0673 3.5975 0.77285
Н	4.56273 2.65317 0.65891
Н	3.80437 3.31899 -0.79796
С	-2.82792 -1.23702 -1.93851
Н	-2.85243 -1.64353 -0.92079
Н	-3.84708 -1.15893 -2.32462
Н	-2.26253 -1.95435 -2.54573
С	0.9164 3.66658 -1.23677
С	1.58025 4.14951 -2.37566
C	0.55774 4.58332 -0.23593
Ċ	1.87761 5.50762 -2.51179
Ċ	0.8534 5.94228 -0.36835
н	0.0438 4.22332 0.65156
Н	2 30104 5 86152 -3 40162
н	0.56777 6 62625 0 41807
C C	15145 640822 - 150728
U U	1.5145 0.40022 $-1.50/30$
11 Ц	1./4505 /.40519 -1.01142
П	1.80320 3.45002 -3.15849
C	3.32122 0.31618 0.48995
C	-4.32689 1.61677 -3.16842
C	-5.43526 1.80629 -3.6755
C	-9.05778 2.42573 -5.3328
Pt	-7.24439 2.11565 -4.50318
P	-6.58496 1.20666 -6.57466
C	-6.74178 2.41287 -7.95458
Н	-6.49006 1.94672 -8.91324
Н	-6.0707 3.25857 -7.7771
Н	-7.77014 2.7822 -7.98054
С	-4.85885 0.58914 -6.70725
Н	-4.69731 -0.20446 -5.97326
Н	-4.16326 1.40219 -6.48421
Н	-4.66366 0.20385 -7.7136
С	-7.6204 -0.22086 -7.09783
Н	-7.50329 -1.03884 -6.38071
Н	-8.66793 0.09099 -7.105
Н	-7.33056 -0.57061 -8.09453
Р	-7.92243 3.02927 -2.43932
С	-9.65396 3.63293 -2.31962
Н	-9.85335 4.03301 -1.31989
Н	-10.3413 2.80963 -2.52987
Н	-9.82211 4.4112 -3.0682
C	-7.76317 1.83344 -1.05025
н	-6 73055 1 47720 -1 0002
н	-8 42064 = 0.07754 = 1.2207
и И	-8 02126 2 20160 -0.00700
C C	-6 00202 4 46010 -1 01720
С U	
п u	-/.02349 5.20201 -2.03901
п	-/.20221 4.82122 -0.92418
C	-10.10186 2.61452 -5.83791
C	10.63659 -1.06848 4.1249
C	17.93401 0.14024 8.77974
С	15.96895 0.52987 7.65719
С	16.32811 -0.8465 7.55243

С	15.68747 -1.93567 6.89805
C	16.27521 -3.23027 6.95515
С	15.86259 -4.47577 6.39513
С	17.82207 -4.74625 7.56152
Ν	17.53301 -1.04054 8.25136
Ν	17.47538 -3.4404 7.65726
В	18.29242 -2.37409 8.41736
F	19.58421 -2.27176 7.87547
F	18.39414 -2.71207 9.77663
С	14.77863 1.25054 7.09659
H	14.75108 1.22731 6.00069
Н	13.82956 0.82527 7.442
Н	14 8120 2 20868 7 41166
C	10.0/125 -5.2180 -8.10/02
н	10 10207 -6 28070 7 08506
н	19.10307 0.30979 7.90390
и П	19.94359 - 4.02140 / .01093
II C	19.0200/ -5.15840 9.2/802
	14.05941 -4./9402 5.55/41
п	14.03648 -4.22256 4.62225
H	14.67174 -5.85882 5.30231
H	13.71774 -4.58165 6.07665
C	19.17295 0.29701 9.589
H	20.05042 -0.03539 9.02228
H	19.30088 1.34265 9.8791
Н	19.12995 -0.32793 10.48931
C	14.40177 -1.72382 6.16539
C	13.16783 -1.94629 6.80172
С	14.39449 -1.35608 4.81061
С	12.0246 -1.69846 6.13503
С	13.12843 -1.11859 4.03137
Н	15.3412 -1.22546 4.29287
Η	11.10668 -1.7953 6.70918
Η	13.04969 -0.82982 2.98635
С	11.97355 -1.32672 4.78931
Η	13.15937 -2.22789 7.85148
С	-17.95005 1.60049 -9.07061
С	-15.94176 1.13134 -8.06065
С	-15.99784 2.50996 -8.42209
С	-15.05365 3.56177 -8.25741
С	-15.37454 4.8691 -8.7185
С	-14.62902 6.08507 -8.69135
С	-16.67551 6.43664 -9.67189
N	-17.23992 2.75318 -9.0353
N	-16.61507 5.128/1 -0.32606
B	-17.74505 4.1068 -0.57861
F	-18 017/3 / 01088 -10 05318
F	-18 01622 4 40554 -8 00824
Ċ	-1484498 096799 -798089
й	-19 01815 0 25026 -7 06757
и П	-14 = 8601 = 0.33030 = 7.90/57
11 U	-14.50001 0.70559 -0.40144
п С	-15.10001 -0.00/97 -7.23047
U	
тт	-17.85481 7.05604 -10.3352
H	-17.85481 7.05604 -10.3352 -17.6724 8.11975 -10.50596

Н	-18.75378 6.93456 -9.71926
С	-13.25159 6.34875 -8.15852
Η	-12.48593 5.74358 -8.65767
Η	-13.00288 7.40368 -8.3143
Η	-13.16798 6.13642 -7.08643
С	-19.31709 1.49786 -9.64954
Η	-19.32066 1.82711 -10.6951
Η	-19.67204 0.46614 -9.59092
Η	-20.01498 2.15388 -9.11517
С	-13.73938 3.29792 -7.59576
С	-13.61045 3.37717 -6.1944
С	-12.63608 2.8407 -8.33605
С	-12.36315 3.21678 -5.58618
C	-11.39791 2.63579 -7.7223
Η	-12.73785 2.71058 -9.41034
Η	-12.25096 3.41458 -4.52322
Η	-10.539 2.34626 -8.32238
C	-11.26344 2.80289 -6.34189
Η	-14.4648 3.69885 -5.60451
С	-17.16592 0.5611 -8.46919
Η	-17.4462 -0.4642 -8.34634
С	-15.44557 7.06741 -9.28981
Η	-15.18215 8.09527 -9.42768
C	16.8335 -5.42574 6.77563
Н	16.82224 -6.46403 6.51731
С	16.97751 1.14856 8.42574
Н	17.01068 2.18494 8.6898

Optimized triplet state geometry of Pt-a

03	
С	1.90253 -0.07808 -0.03096
Р	3.91845 0.41903 -2.36408
С	4.6862 2.01973 -2.84505
Н	4.77204 2.09804 -3.93429
Н	5.67755 2.08349 -2.38928
Н	4.07359 2.84531 -2.4705
С	4.92241 -0.85424 -3.23277
Н	4.4621 -1.83689 -3.09272
Н	5.92348 -0.8708 -2.79469
Н	4.98805 -0.63467 -4.30395
С	2.3081 0.40319 -3.25046
Н	1.8164 -0.56064 -3.09614
Н	1.65818 1.18198 -2.84357
Η	2.45883 0.57126 -4.32206
Р	3.76891 -0.22558 2.30827
С	3.03453 -1.84409 2.7823
Η	2.93938 -1.92701 3.87027
Η	3.66944 -2.65491 2.41283
Η	2.04903 -1.93213 2.31798
С	5.36155 -0.16215 3.22373
Н	5.83466 0.81045 3.06767
Η	6.03663 -0.93 2.83858
Н	5.19043 -0.32216 4.29372



С	2.70959 1.01779 3.15484
Н	3.1398 2.01449 3.0179
Н	1.7167 1.00074 2.69815
Η	2.62973 0.80245 4.22573
С	0.68048 -0.187 -0.03342
С	-1.44428 -1.54651 -0.13041
С	-1.68952 0.71397 0.04691
С	-0.73395 -0.31307 -0.03626
С	-2.97072 0.07968 0.00454
С	-4.27556 0.59665 0.05016
С	-5.39644 -0.25159 -0.01047
С	-6.80433 0.01991 0.02185
С	-6.44939 -2.21904 -0.15503
Ν	-2.76456 -1.30436 -0.10417
Ν	-5.23087 -1.63841 -0.11954
В	-3.87569 -2.39018 -0.18892
F	-3.77182 -3.08818 -1.39607
F	-3.75831 -3.27679 0.88543
С	-1.34497 2.16821 0.15549
H	-1.77064 2.75619 -0.66474
H	-1.71585 2.61265 1.08587
Н	-0.25699 2.28171 0.13214
C	-6.63933 -3.69715 -0.26785
Ĥ	-7.70356 -3.94519 -0.28239
Н	-6.17038 -4.07952 -1.18098
Н	-6.16318 -4.21411 0.57231
C	-7.52449 1.33241 0.12938
H	-7.29198 2.00048 -0.70684
Н	-8.60527 1.15088 0.13406
H	-7.26377 1.87361 1.04519
C	-0.86568 -2.91694 -0.24411
H	-1.23322 -3.41636 -1.14678
H	0.22457 - 2.85852 - 0.2783
Н	-1.17026 -3.53709 0.60626
C	-4.47906 2.07357 0.16697
Č	-4.56915 2.67862 1.42813
C	-4.58/83 2.86810 -0.08305
C	-4.76201 4.05742 1.53678
C	-4.77704 4.247 -0.87246
н	-4 51550 2 40284 -1 06203
H	-4 83084 4 51540 2 51063
H	-4 85605 4 85208 -1 77077
C	-4 86622 4 84420 0 38710
й	-5.01602 = 4.04429 = 0.30719
H	-4.4887 2.06662 2.2220
Cl	6.12078 $0.20868 - 0.02244$
Pt	3 85878 0 00627 -0 02702
C	-7 42562 -1 21788 -0 06051
ч	-8 = 0.002 = 1.21/00 = 0.00951
11	-0.50392 -1.39324 -0.0/45

Optimized triplet state geometry of Pt-b

0 3 C 2.50952 0.00691 0.00567



Pt	4.47032 -0.0133 -0.0133
Р	4.31963 -1.67937 -1.67782
С	3.40876 -1.11108 -3.17157
Н	3.28484 -1.92813 -3.89027
Н	3.96261 -0.2947 -3.64468
Н	2.42942 -0.73551 -2.86423
С	5.89888 -2.34931 -2.33516
Н	6.47816 -2.78649 -1.51829
Н	6.48788 -1.53734 -2.76836
Н	5.70034 -3.11058 -3.09725
C	3.39364 -3.16565 -1.11414
Ĥ	3.03777 -3.6/1002 -0.202/0
н	2 / 12 / -2 / 85103 - 0 / 707
н	2.27157 - 2.88508 - 1.02082
P	4 60706 1 65241 1 65284
C	2.02598 2.01989 2.01494
н	2 21508 2 07604 2 07621
и П	3.21390 3.07094 3.07021
и П	2.4441/1.49/50 2.75120
II C	2.44193 $2./4053$ 1.49050
U U	5.52009 1.06/05 3.14551
11 Ц	0.5008/0.7033 2.83502
п	4.90903 0.2009/ 3.01093
П	5.64124 1.90592 3.86291
	5.51941 3.14001 1.0859
п	6.49948 2.83/28 0./1442
H	4.96746 3.61983 0.26832
H	5.6402 3.86395 1.90478
C	1.28276 0.00091 -0.00089
C	-6.98304 -1.9362 1.62229
C	-4.70983 -1.98386 1.65018
C	-5.16859 -0.93045 0.79019
C	-4.47571 0.01373 0.01347
C	-5.16797 0.96134 -0.75963
C	-4.70853 2.01274 -1.62171
C	-6.98177 1.97614 -1.58212
N	-6.56989 -0.94322 0.80784
N	-6.56926 0.98098 -0.76999
B	-7.49925 0.02077 0.02086
F	-8.30646 -0.70338 -0.86237
F	-8.29989 0.74816 0.90741
C	-3.30788 -2.4013 1.98574
Н	-2.7403 -2.69304 1.0958
Н	-2.742 -1.59723 2.46864
Н	-3.33157 -3.25597 2.66899
C	-8.42503 2.28184 -1.82062
Н	-8.52873 3.09396 -2.54439
Н	-8.9506 1.39708 -2.19477
Н	-8.91912 2.57168 -0.88659
С	-3.30631 2.42338 -1.96454
Н	-2.74753 1.61709 -2.45204
Н	-3.32938 3.27925 -2.64629
Н	-2.73205 2.71058 -1.07744
С	-8.42653 -2.235 1.86807
Н	-8.92645 -2.52333 0.93667

Н	-8.53048 -3.04607 2.59298
Н	-8.94619 -1.34752 2.24398
С	-2.98158 0.01018 0.00969
С	-2.2624 0.77796 0.93682
C	-2.27082 -0.76099 -0.92109
С	-0.87025 0.77654 0.93452
С	-0.8787 -0.76591 -0.92575
Н	-2.81421 -1.36138 -1.64576
Н	-0.32844 1.37672 1.65906
Н	-0.3434 -1.36883 -1.65316
C	-0.14429 0.00368 0.00244
Н	-2.7993 1.38089 1.66423
Cl	6.93154 -0.02967 -0.02928
С	-5.8551 -2.59236 2.15432
Н	-5.88726 -3.4297 2.8398
С	-5.8534 2.62693 -2.11981
Н	-5.885 3.46463 -2.80487

Optimized triplet state geometry of Pt-1

03	
С	-4.81501 -0.21494 0.00301
С	-6.03098 -0.29468 0.01218
С	-10.08621 -0.50106 0.02474
Pt	-8.0533 -0.41344 0.02011
Р	-7.93503 -0.75673 -2.30459
С	-7.12781 0.64083 -3.18869
Η	-7.02572 0.42276 -4.25731
Η	-6.14161 0.80825 -2.74819
Η	-7.72685 1.54747 -3.06057
С	-6.92293 -2.223 -2.76427
Η	-7.39028 -3.12862 -2.36606
Η	-5.93097 -2.11633 -2.31827
Η	-6.83423 -2.31134 -3.85255
С	-9.51604 -1.00512 -3.20894
Η	-10.03542 -1.87761 -2.80461
Η	-10.15976 -0.1337 -3.06586
Η	-9.32681 -1.15289 -4.27756
Р	-8.1678 -0.08989 2.34728
С	-9.01093 -1.47817 3.21244
Η	-9.10706 -1.27258 4.28408
Η	-8.43523 -2.39792 3.07142
Η	-10.00114 -1.61501 2.77031
С	-6.58172 0.10677 3.25535
Η	-6.03445 0.96315 2.85362
Η	-5.96511 -0.7839 3.11077
Η	-6.76803 0.25601 4.32428
С	-9.14289 1.3954 2.82684
Η	-8.64914 2.2951 2.44739
Η	-10.13495 1.3234 2.37416
Н	-9.2354 1.46668 3.91606
С	2.54238 -1.135 0.05384
С	2.55308 1.14177 -0.06222
С	3.38502 0.01126 -0.00534





С	1.20943 0.65661 -0.03809
C	-0.02983 1.31985 -0.07224
С	-1.24116 0.60671 -0.03551
С	-2.60349 1.03655 -0.06393
C	-2.50064 -1.23667 0.06699
Ν	1.25733 -0.74375 0.03348
Ν	-1.23196 -0.7938 0.04424
В	0.03156 -1.69635 0.09178
F	0.04328 -2.56256 -1.00554
F	0.05296 -2.43247 1.28054
C	3.06045 2.55023 -0.13035
Η	2.7416 3.06127 -1.04566
H	2.70555 3.15884 0.7084
Н	4.15438 2.53856 -0.10632
C	-2.86063 -2.68207 0.14915
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Optimized triplet state geometry of Pt-2

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H H H	3.11878 2.73082 0.67173 2.81915 3.46081 -0.08878 2.77141 3.12069 1.63459 4.21192 2.68771 0.69355
H H H C	3.11878 2.73082 0.67173 2.81915 3.46081 -0.08878 2.77141 3.12069 1.63459 4.21192 2.68771 0.69355 -2.91507 -2.27814 -0.33205
H H H C H	3.11878 2.73082 0.67173 2.81915 3.46081 -0.08878 2.77141 3.12069 1.63459 4.21192 2.68771 0.69355 -2.91507 -2.27814 -0.33205 -4.00366 -2.36868 -0.33215
H H H C H H	3.11878 2.73082 0.67173 2.81915 3.46081 -0.08878 2.77141 3.12069 1.63459 4.21192 2.68771 0.69355 -2.91507 -2.27814 -0.33205 -4.00366 -2.36868 -0.33215 -2.52238 -2.60242 -1.302
С Н Н С Н Н Н	3.11878 2.73082 0.67173 2.81915 3.46081 -0.08878 2.77141 3.12069 1.63459 4.21192 2.68771 0.69355 -2.91507 -2.27814 -0.33205 -4.00366 -2.36868 -0.33215 -2.52238 -2.60242 -1.302 -2.40366 -2.0520 0.42007
С Н Н С Н Н Н Н	3.11878 2.73082 0.67173 2.81915 3.46081 -0.08878 2.77141 3.12069 1.63459 4.21192 2.68771 0.69355 -2.91507 -2.27814 -0.33205 -4.00366 -2.36868 -0.33215 -2.52238 -2.60242 -1.302 -2.49366 -2.9529 0.42097 -2.11224 2.72056 0.67016
С Н Н С Н Н Н Н С Н Н	3.11878 2.73082 0.67173 2.81915 3.46081 -0.08878 2.77141 3.12069 1.63459 4.21192 2.68771 0.69355 -2.91507 -2.27814 -0.33205 -4.00366 -2.36868 -0.33215 -2.52238 -2.60242 -1.302 -2.49366 -2.9529 0.42097 -3.11324 2.73956 0.67916 -2.78858 2.46821 -0.07103
С Н Н С Н Н Н С Н Н Н Н С Н Н	3.11878 2.73082 0.67173 2.81915 3.46081 -0.08878 2.77141 3.12069 1.63459 4.21192 2.68771 0.69355 -2.91507 -2.27814 -0.33205 -4.00366 -2.36868 -0.33215 -2.52238 -2.60242 -1.302 -2.49366 -2.9529 0.42097 -3.11324 2.73956 0.67916 -2.78858 3.46821 -0.07193 -4.20680 2.70402 0.6727
С Н Н С Н Н Н Н С Н Н Н Н Н Н Н Н Н Н Н	3.11878 2.73082 0.67173 2.81915 3.46081 -0.08878 2.77141 3.12069 1.63459 4.21192 2.68771 0.69355 -2.91507 -2.27814 -0.33205 -4.00366 -2.36868 -0.33215 -2.52238 -2.60242 -1.302 -2.49366 -2.9529 0.42097 -3.11324 2.73956 0.67916 -2.78858 3.46821 -0.07193 -4.20689 2.70493 0.6727
С Н Н Н С Н Н Н Н С Н Н Н Н С Н Н Н С Н Н Н С Н Н Н С Н Н Н С Н Н Н С С Н Н Н С С Н С С С Н С С Н С С С С С Н С	3.11878 2.73082 0.67173 2.81915 3.46081 -0.08878 2.77141 3.12069 1.63459 4.21192 2.68771 0.69355 -2.91507 -2.27814 -0.33205 -4.00366 -2.36868 -0.33215 -2.52238 -2.60242 -1.302 -2.49366 -2.9529 0.42097 -3.11324 2.73956 0.67916 -2.78858 3.46821 -0.07193 -4.20689 2.70493 0.6727 -2.78727 3.12499 1.6515 2.00500 -2.97840 0.27875
С Н Н Н С Н Н Н Н С Н Н Н Н С Н Н Н С Н Н Н С Н Н Н С Н Н Н С Н Н Н С С Н Н Н С С Н С С Н С С Н С С Н С С С С С Н С	$\begin{array}{rrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrr$
С Н Н Н Н С Н Н Н Н С Н Н Н Н С Н Н Н Н	3.11878 2.73082 0.67173 2.81915 3.46081 -0.08878 2.77141 3.12069 1.63459 4.21192 2.68771 0.69355 -2.91507 -2.27814 -0.33205 -4.00366 -2.36868 -0.33215 -2.52238 -2.60242 -1.302 -2.49366 -2.9529 0.42097 -3.11324 2.73956 0.67916 -2.78858 3.46821 -0.07193 -4.20689 2.70493 0.6727 -2.78727 3.12499 1.6515 2.90509 -2.27842 -0.37872 2.45298 -2.61491 -1.3176

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