

Nanocluster seed-mediated synthesis of CuInS_2 quantum dots, nanodisks, nanorods, and doped Zn-Ga-CuInS_2 quantum dots

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Supporting Information

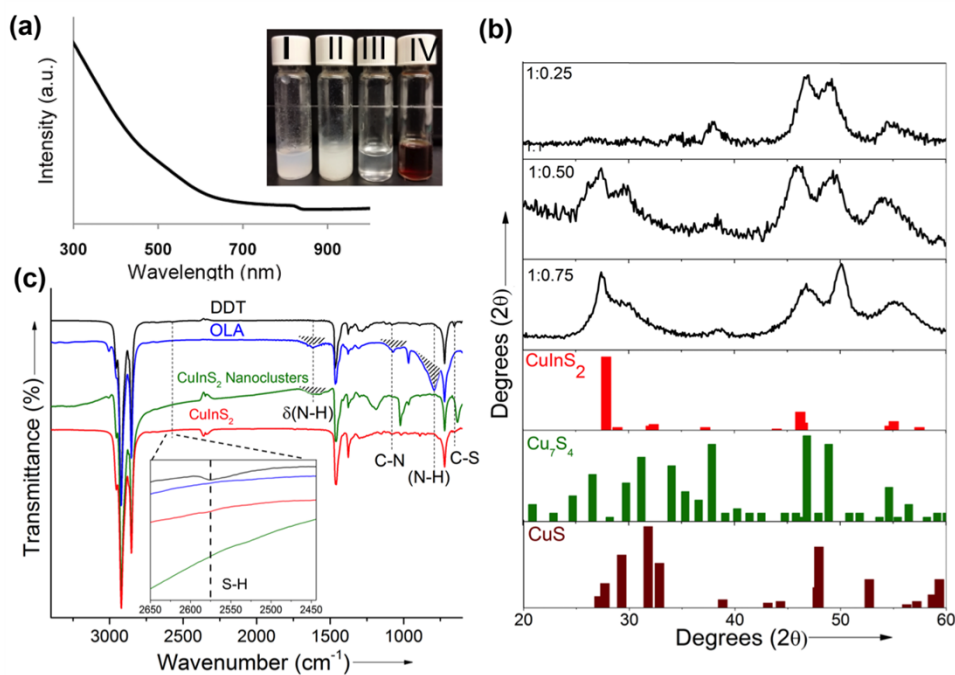


Figure S1. a) UV-vis optical absorption spectrum of CuInS_2 nanoclusters. Inset image shows a photograph of the following compounds in dodecanethiol (DDT) at room temperature: I) CuCl , II) InCl_3 , III) a mixture of Cu and In (1:1 molar ratio) and IV) CuInS_2 nanoclusters. b) XRD patterns for products obtained during the nanocluster formation stage with different Cu:In molar ratios. Increasing the molar ratio of Cu:In to 1:0.75 favors the formation of CuInS_2 , while lower molar ratios result mixtures of Cu_7S_4 and CuS . c) FT-IR spectra of DDT, oleylamine (OLA), CuInS_2 nanoclusters and CuInS_2 nanoparticles. Inset image shows the weak S-H vibrations corresponding to DDT, which are present in pure DDT and in the nanoclusters. The absence of S-H vibrations in both clusters and the final CuInS_2 nanoparticles indicates the capping of CuInS_2 by DDT.

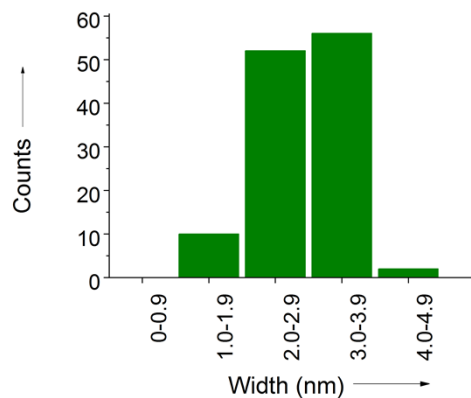


Figure s2. Distribution of the thickness of nanodisks.

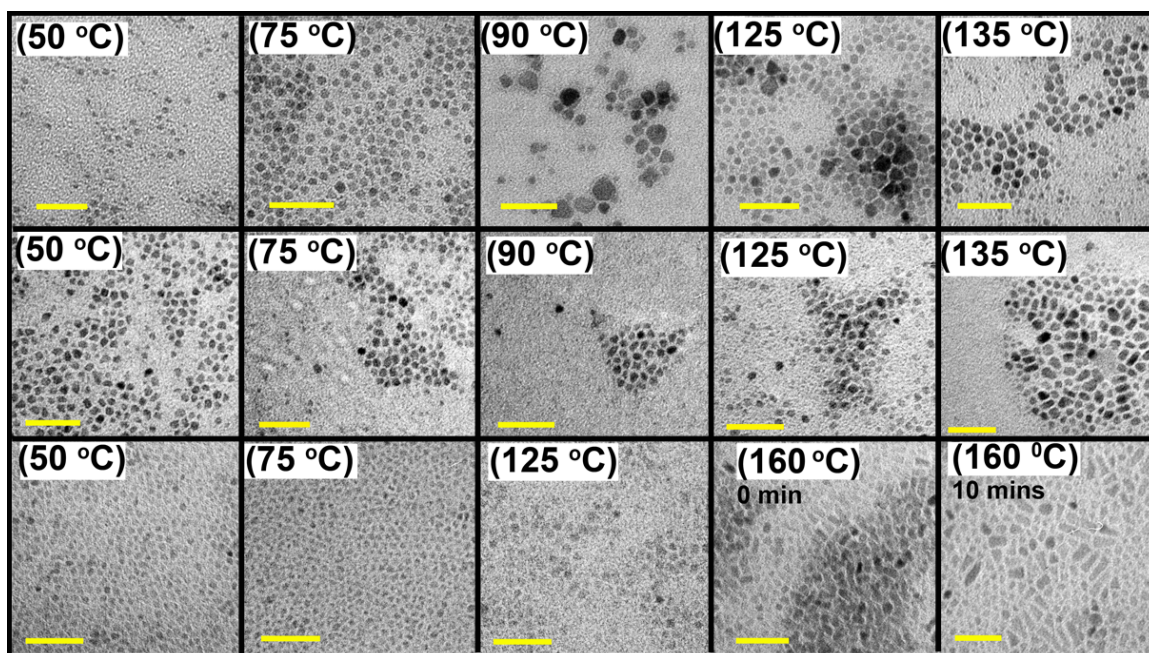


Figure s3. Aliquots of CuInS₂ q-dots (first row), nanodisks (second row), nanorods (third row) collected at different temperatures.

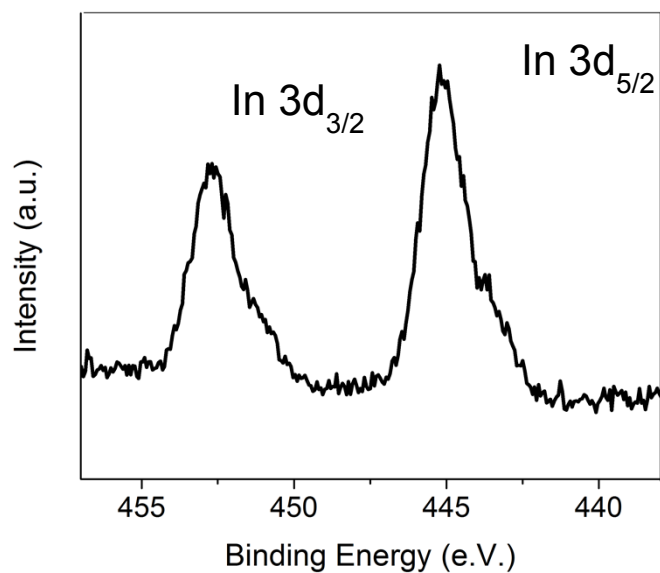


Figure s4. XPS spectra of In 3d core electrons in CuInZnS.

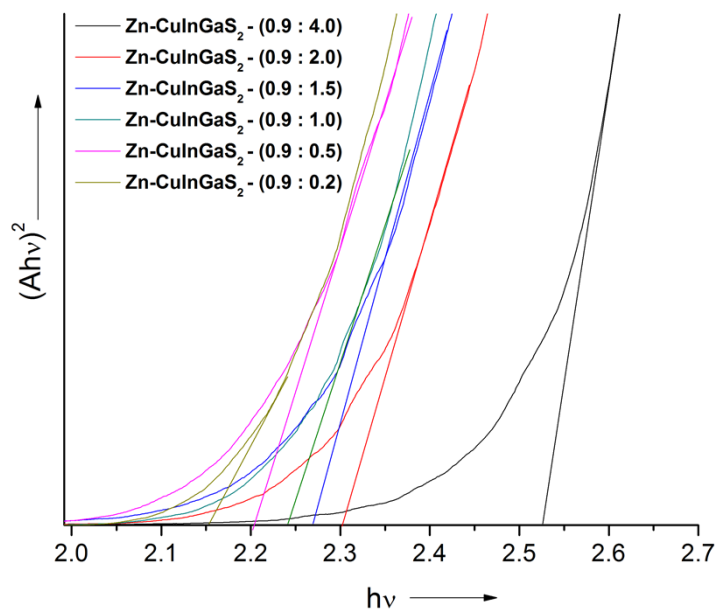


Figure s5. Least-squares fit of the linear region of the optical absorption edge corresponding to different Ga/Cu starting molar ratios. (A - absorbance, h - Planck's constant, ν - frequency).

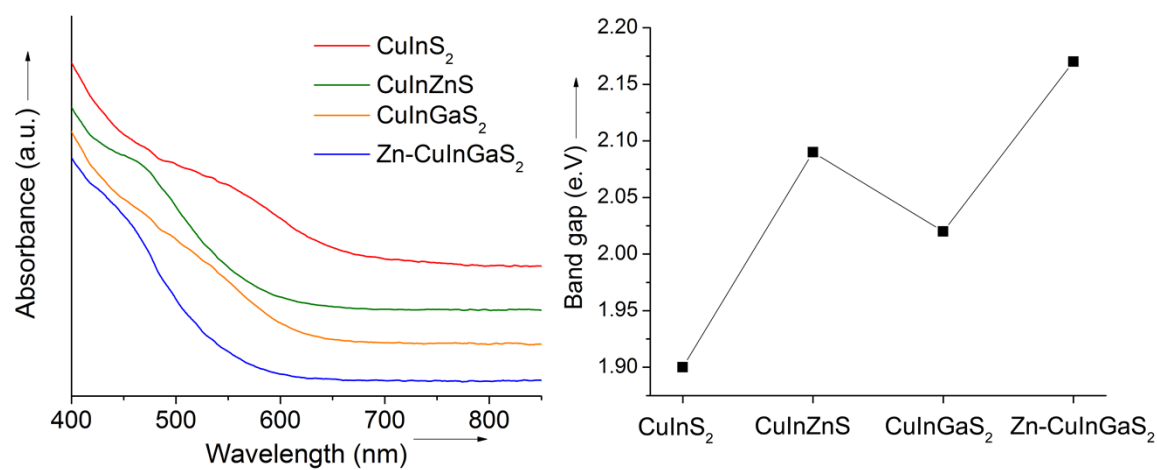


Figure s6. a) Comparison of optical absorption spectra of CuInS₂, CuInZnS, CuInGaS₂ and Zn-CuInGaS₂, which shows the synergistic effect of cation combination on the absorption edge. b) Synergistic effect of Zn and Ga on band gap. The differences in the band gaps of CuInS₂, CuInZnS, CuInGaS₂ and Zn-CuInGaS₂ clearly show the overall effect of incorporation of both Ga and Zn.