

Electronic Supplementary Information

Tuning the Field Emission Properties of AlN Nanocones by Doping

Qiang Wu,* Ning Liu, Yongliang Zhang, Weijin Qian, Xizhang Wang and Zheng Hu

Key Laboratory of Mesoscopic Chemistry of MOE, School of Chemistry and Chemical Engineering, Nanjing University, Nanjing 210093, China. E-mail: wqchem@nju.edu.cn

Detailed experimental

The CVD growth of doped AlN nanocones was carried out in a three-zone tubular furnace (see Fig. S1). For synthesizing Mg-doped AlN nanocones, anhydrous MgCl₂ was used as a new kind of dopant. Typically, about 0.5 g of anhydrous AlCl₃ and MgCl₂ were separately placed at the precursor-loaded zones I and II and a Si(100) substrate at the deposition zone III. After evacuating and Ar flushing the sealed alumina tube for several times, the three zones were heated to 125, 700 and 750 °C, respectively, under the protection of Ar gas. Flowing Ar of 300 mL min⁻¹ was then introduced to transport the AlCl₃ and MgCl₂ vapours to zone III, where they mixed and reacted with NH₃ gas (20 mL min⁻¹) for 3 h. After the system was cooled down to ambient temperature, AlN nanocone arrays with Mg doping were obtained. If a Mo grid with the diameter of 8 mm was covered on the Si substrate during the growth, patterned arrays of Mg-doped AlN nanocones were gotten by disposing of the Mo grid mask. The preparation procedure for the Si-doped AlN nanocones was quite similar except that the dopant source was SiH₄ in this case. Briefly, anhydrous AlCl₃ and Mo grid-covered Si(100) substrate were loaded at zone I and zone III, respectively. The temperatures of the three zones were also set to be 125, 700 and 750 °C though zone II was nothing-loaded in this synthesis. When the furnace reached the desired temperature, 300 mL min⁻¹ of Ar, together with 1.5 mL min⁻¹ of SiH₄/Ar (SiH₄, 5 vol%), were introduced into the system. After 3 hours of CVD growth, Si-doped AlN nanocone arrays with patterned distribution were synthesized by in situ doping.

The products were examined by X-ray diffraction (XRD, Philips X'pert Pro X-ray diffractometer), scanning electron microscopy (SEM, Hitachi S-4800) attached with an energy dispersive X-ray spectroscopy (EDS, SHIMADZU-SEDX), and high resolution transmission electron microscopy (HRTEM, JEM-2100). FE properties were tested using a parallel-plate diode configuration with a cathode-anode distance of 100 μm in a vacuum chamber of 8×10⁻⁵ Pa. Before the data recording, the field emission of the sample was carried out for 2 hours under high voltage (~3000 V) for desorbing gaseous species on the nanocone surface and stabilizing the emission.

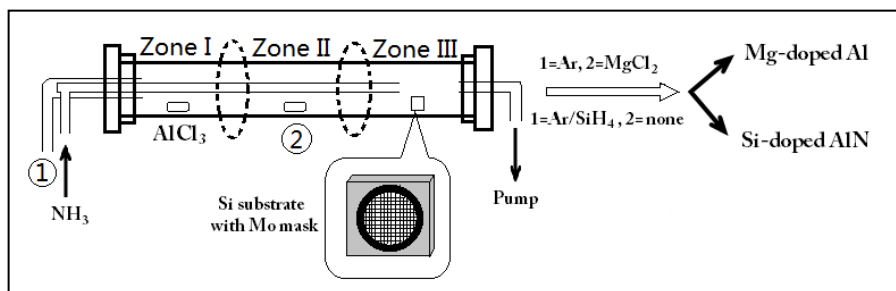


Fig. S1. Schematic of three-zone tubular furnace and the synthetic procedure of the AlN nanocone arrays.

Table S1. Doping amount of Mg in the AlN nanocones obtained at different vaporization temperature of MgCl₂.

Vaporization temperature of MgCl ₂ (°C)	700	800	900	1000
Mg content (at.%)	0.50	0.63	0.71	0.81

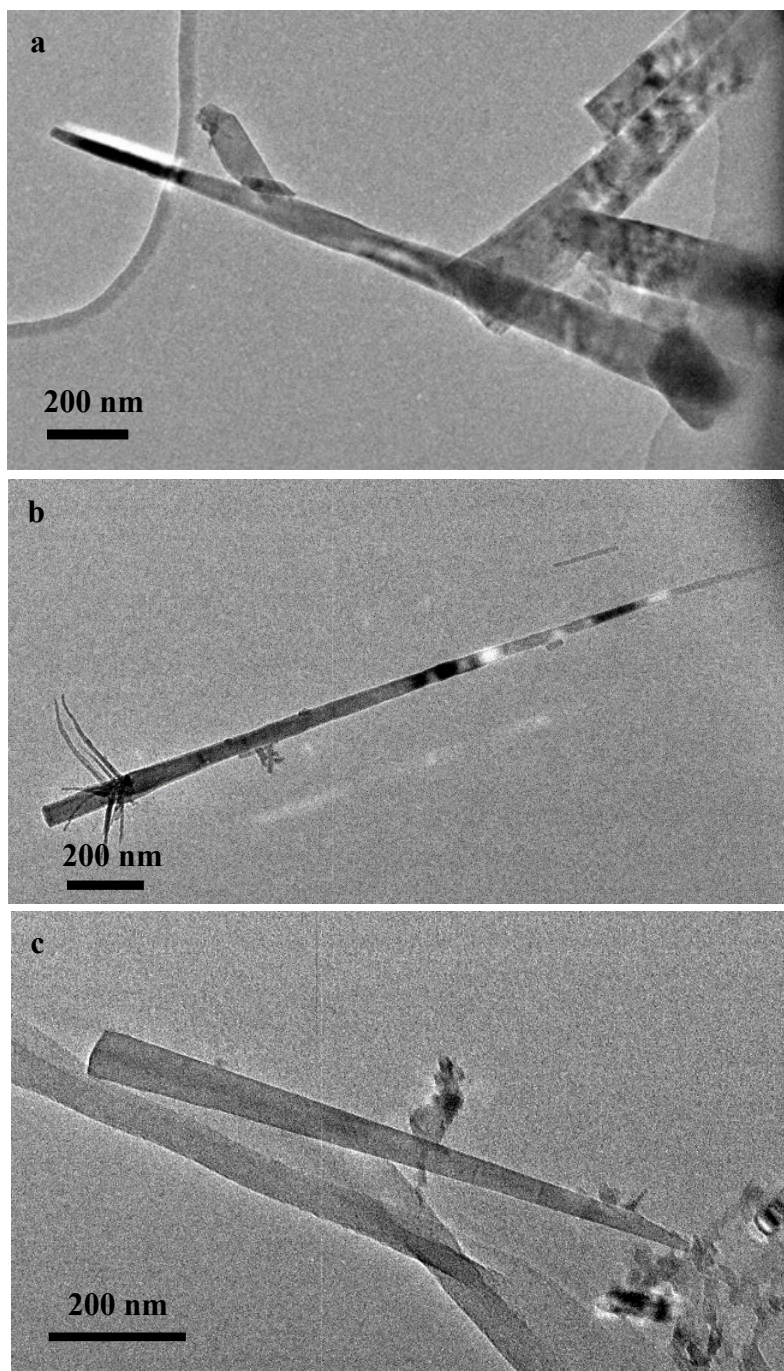


Fig. S2. Typical TEM images of the Si-doped (a), Mg-doped (b) and undoped (c) AlN nanocones. It is shown that the AlN samples have conelike morphologies with the length up to 1~2 micrometer and the diameter of 5~20 nm at the tips and ~50 nm at the roots.

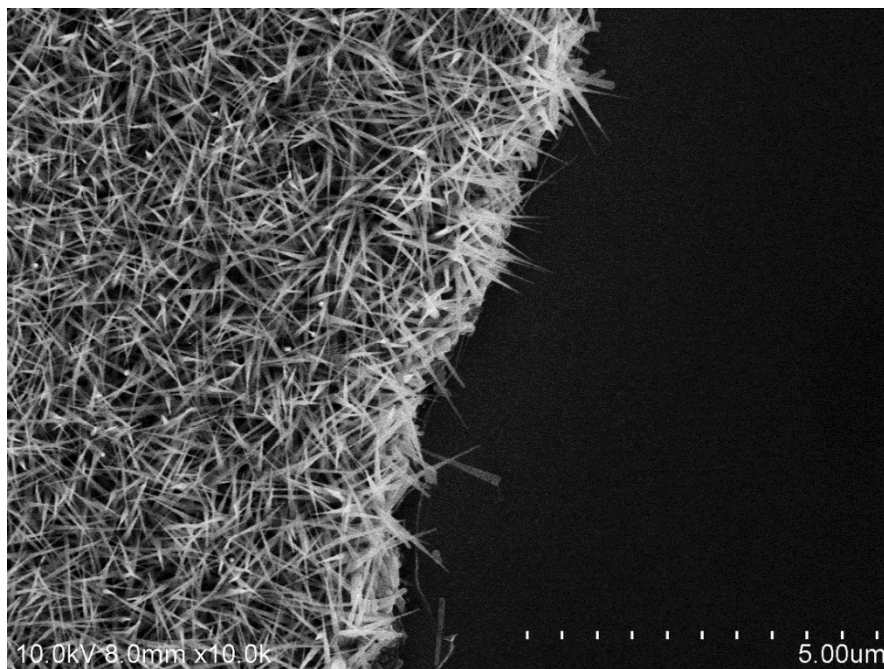


Fig. S3. SEM image of Mg-doped AlN nanocones. It is seen that the nanocones are quasi-aligned.

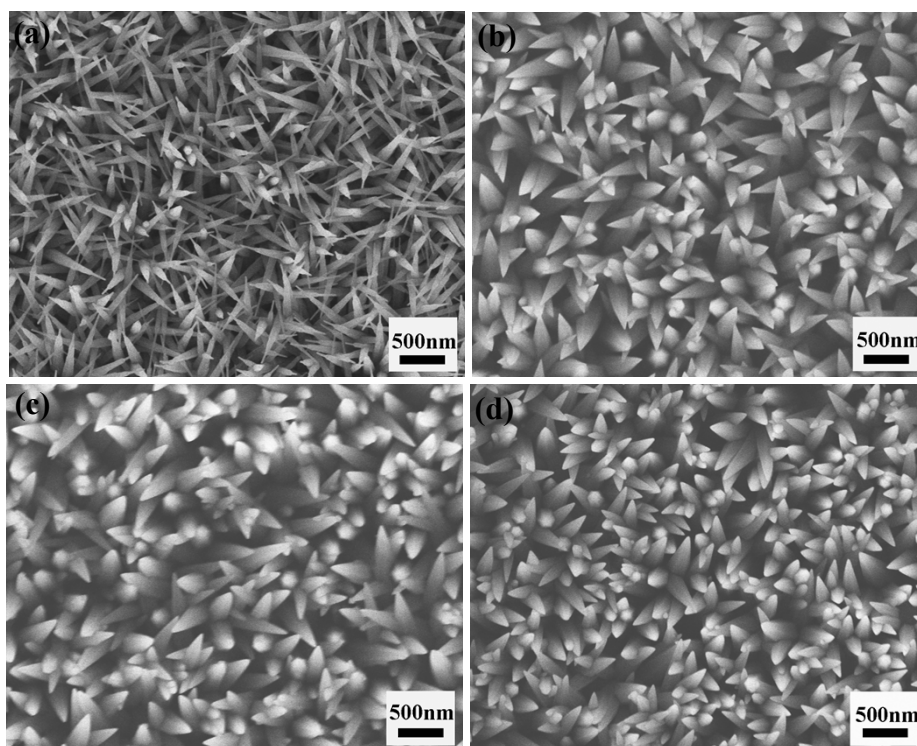


Fig. S4. SEM images of the Si-doped AlN nanocones synthesized at different flow rate of SiH_4/Ar gas: (a) 1 mL min^{-1} , (b) 2 mL min^{-1} , (c) 5 mL min^{-1} , (d) 10 mL min^{-1} .

From (a) to (d), the Si doping amounts were 2.5, 1.2, 1.0 and 0.7 at.% respectively. Meanwhile, the sharpness and length of the nanocones decreased with increasing the flow rate of SiH₄/Ar gas. It is learned that high concentration of SiH₄ was unfavorable for the growth of AlN nanocones, probably owing to the formation of Si₃N₄ species via vapor phase reaction. To obtain Si-doped AlN nanocones with suitable aspect ratio and doping amount, a flow rate of 1.5 mL min⁻¹ was preferred.

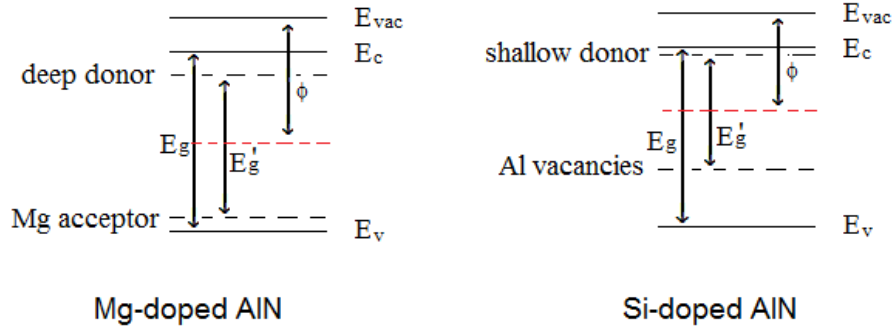


Fig. S5. Sketch of energy levels for AlN. Here, E_c , E_v and E_{vac} are conduction level, valance level and vacuum level, respectively. E_g is the energy gap between E_c and E_v , and E_g' is the energy gap between the donor and acceptor levels.

The work function can be estimated according to the literature results. Many reports have demonstrated that the Mg-doping could induce the light emission band at 4.70 eV originating from the donor-acceptor pair transition involving V_N^{3+} donor (~ 0.90 eV below the conduction band E_c) and Mg acceptor (~ 0.51 eV above the valence band E_v) [1,2]. The Si-doping could result in an enhancement of the Al vacancies in the AlN, and thus PL band at 3.50-4.0 eV could be observed, which could be assigned to recombination from a shallow donor (60 ± 20 meV below the E_c) to $(V_{Al}\text{-complex})^{2-}$ and V_{Al}^{3-} (~ 2.5 eV above the E_v) [3,4]. The work function of semiconductor could be estimated as the sum of half of the band gap ($E_g/2$) and the electron affinity (χ , i.e., the energy gap between E_c and vacuum energy level E_{vac}) [5]. Here, the E_g should be replaced by E_g' , i.e., the energy gap between the donor and acceptor levels, because the electron transitions in the doped AlN occurred indeed between these two energy levels. Based on above analysis, the ϕ of the Si-doped AlN is smaller than that of the Mg-doped sample.

References

- [1] K. B. Mam, et al., Appl. Phys. Lett. 2003, 83, 878-880.
- [2] M. L. Nakarmi, et al., Appl. Phys. Lett. 2006, 89, 152120.
- [3] B. N. Pantha, et al., Appl. Phys. Lett. 2010, 96, 131906.
- [4] E. Monroy, et al., Appl. Phys. Lett. 2006, 88, 071906.
- [5] V. N. Tondare, et al., Appl. Phys. Lett. 2002, 80, 4813-4815.