

Supporting Information

Solution-Processable Host Material of
1,3-Bis{3-[3-(9-carbazolyl)phenyl]-9-carbazolyl}benzene and
Its Application for Organic Light Emitting Diodes Employing
Thermally Activated Delayed Fluorescence

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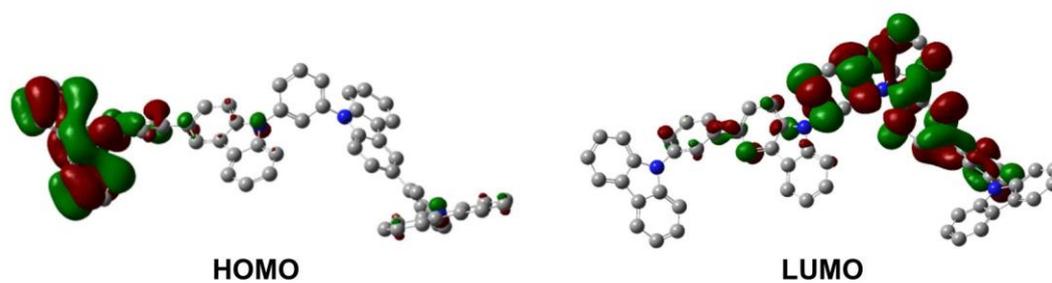


Fig. S1 The HOMO and LUMO of **CPCB** calculated at the B3LYP/6-31G* level. The calculated S_1 and T_1 energies of SPH-01 are 3.23 and 3.06 eV, respectively.

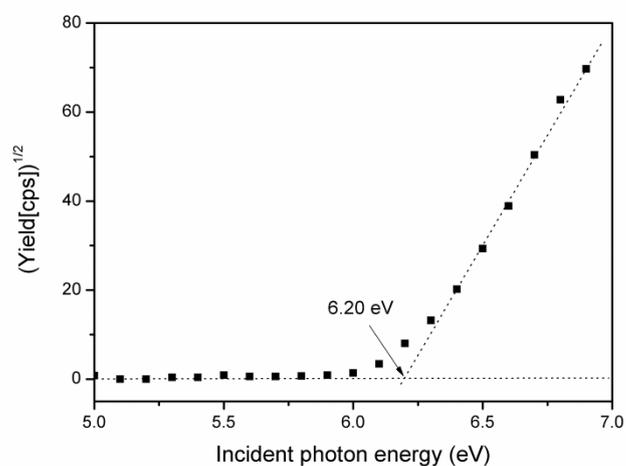


Fig. S2 The photoelectron yield spectra of the neat films of **CPCB**.

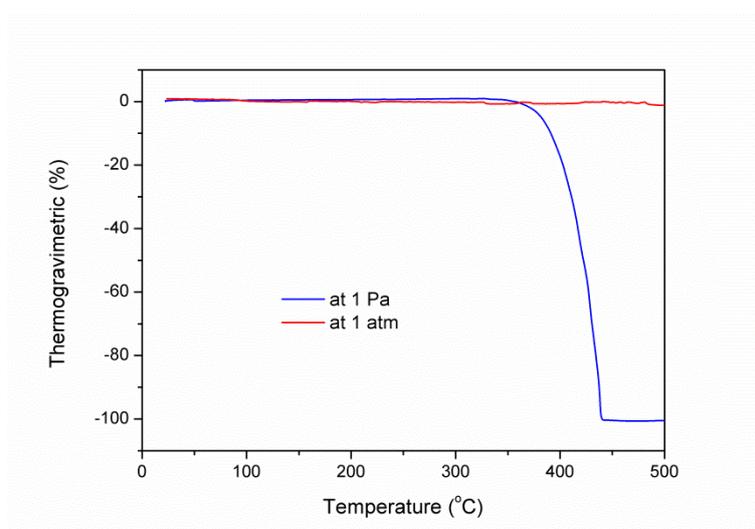


Fig. S3 Thermogravimetric analysis of **CPCB** at 1 Pa and 1 atm under N_2 .

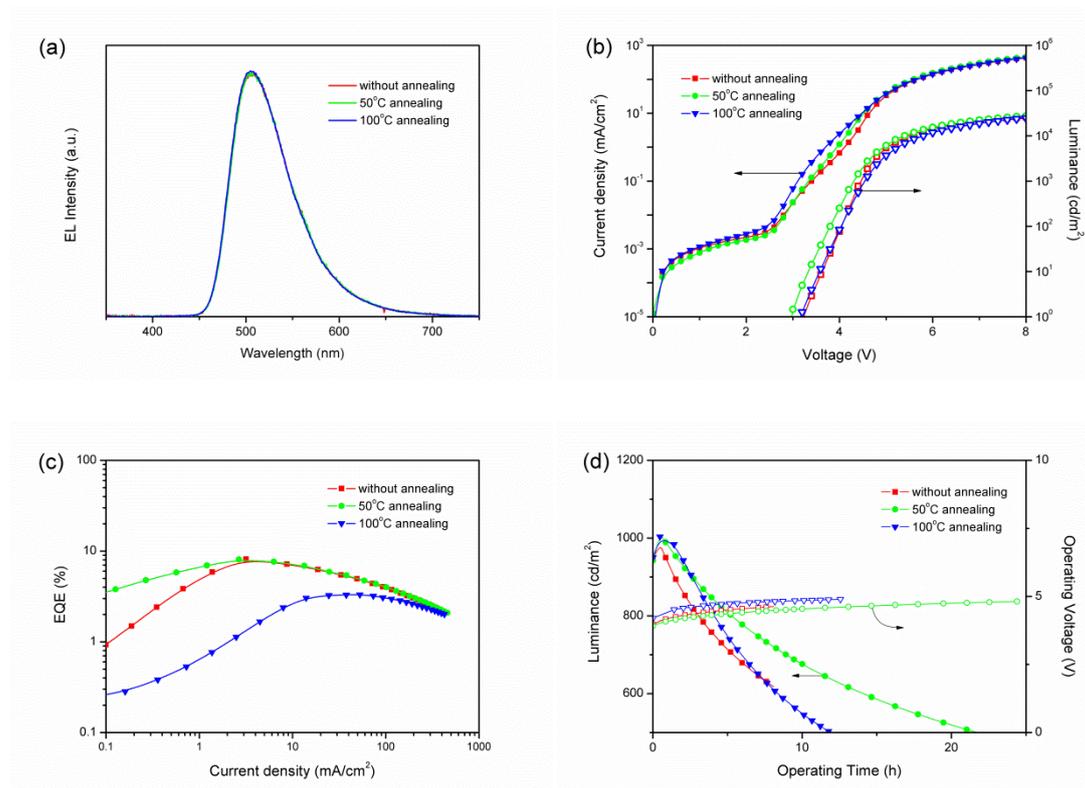


Fig. S4 (a) EL spectra at 10 mA/cm²; (b) luminance-current density-voltage characteristics; (c) EQE-current density characteristics; and (d) luminance-voltage-operating time characteristics at fixed current densities of ITO/PEDOT (40 nm)/CBP: 4CzIPN (30 nm)/T2T (10 nm)/BPyTP2(30 nm)/LiF(0.8 nm)/Al devices with an emitting layer annealed at different temperatures. The device with an emitting layer annealed at 50 °C performed best. Since the annealing temperature of 100 °C is higher than the glass transition temperature of CBP (62 °C), the corresponding device had a low EQE maximum due to the formation of CBP dimers, which have low triplet energy levels.



Fig. S5 Photographs of ~100 nm thin films deposited on quartz substrates after one week. (a) Vacuum-deposited CBP film; (b) spin-coated CBP film annealed at 50 °C for 10 min; (c) spin-coated CPCB film annealed at 220 °C for 10 min. The spin-coated CBP film partly crystallized, while the vacuum-deposited CBP and spin-coated CPCB films were still amorphous.

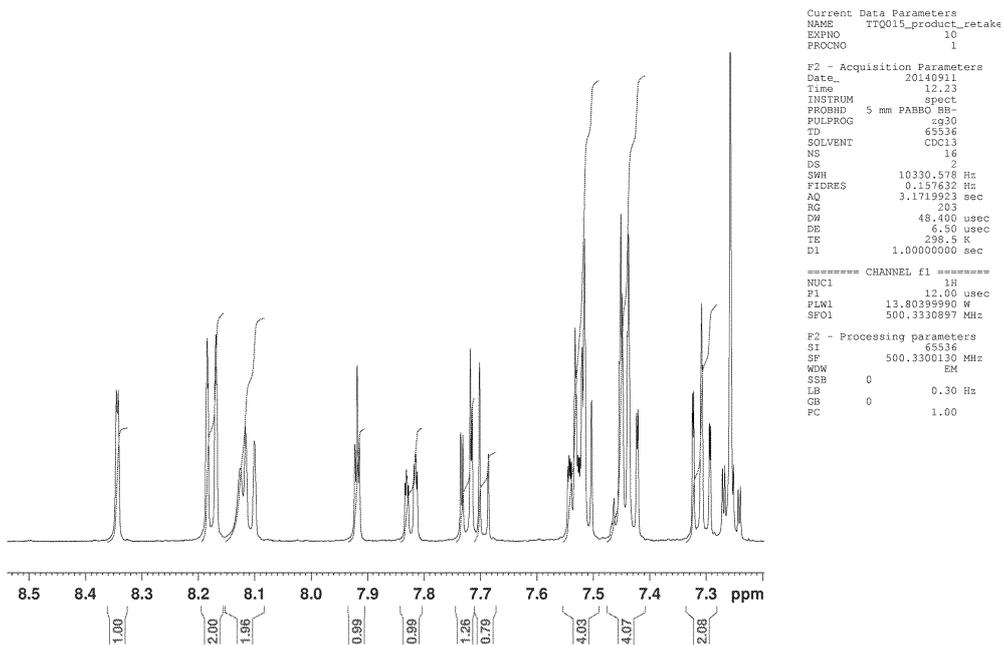


Fig. S4 ^1H NMR spectra of compound **iii**.

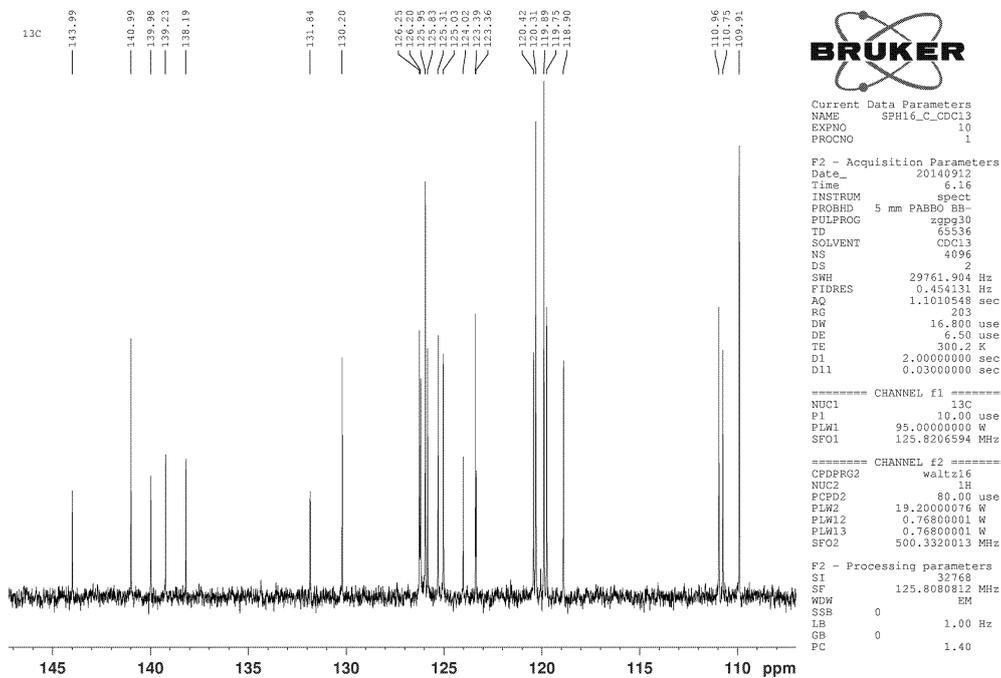


Fig. S5 ^{13}C NMR spectra of compound **iii**.

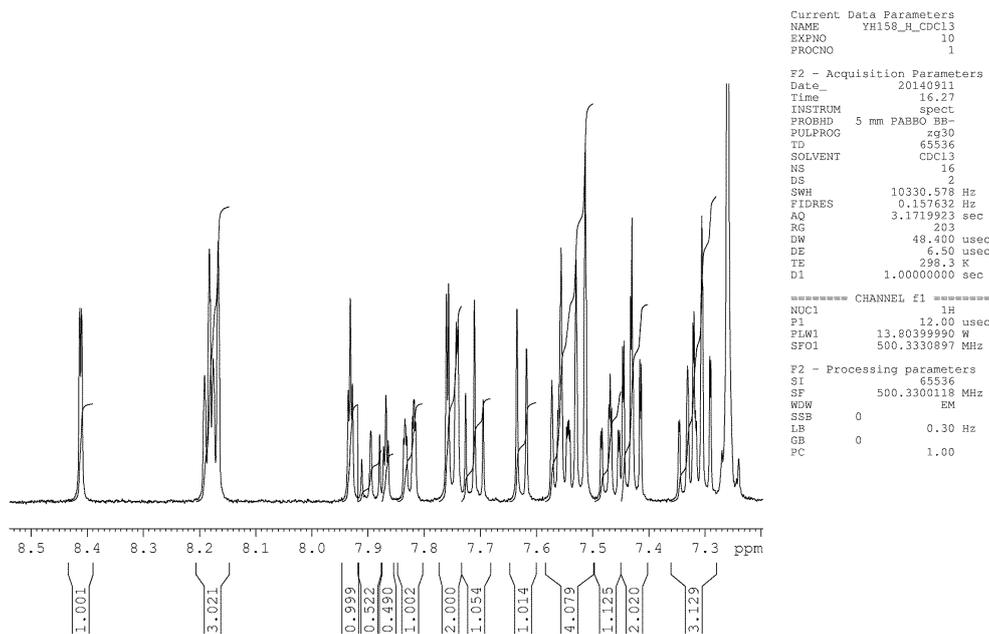


Fig. S6 ^1H NMR spectra of compound CPCB.

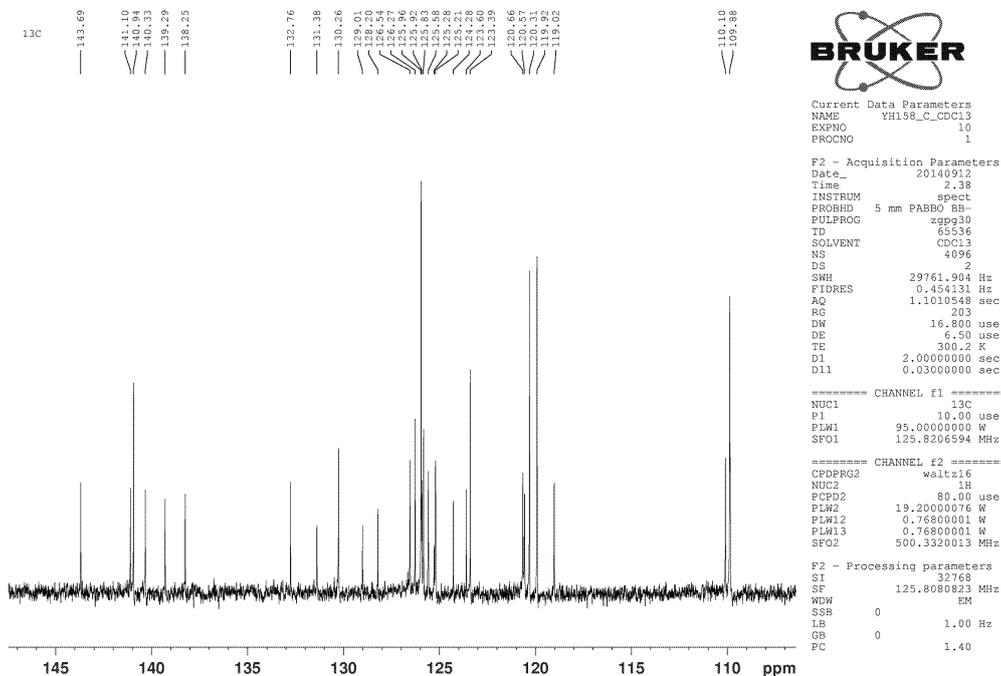


Fig. S7 ^{13}C NMR spectra of compound CPCB.