# SUPPLEMENTARY INFORMATION

# Mechanistic studies on the reactivity of sensitizing allylic hydroperoxides: investigation on the covalent modification of amino acids by carbon-radical intermediates

Dany Kao,<sup>a</sup> Alain Chaintreau,<sup>b</sup> Jean-Pierre Lepoittevin<sup>a</sup> and Elena Giménez-Arnau<sup>\*a</sup>

<sup>a</sup> Laboratoire de Dermatochimie, Institut de Chimie de Strasbourg (CNRS UMR 7177), Université de Strasbourg, 4 Rue Blaise Pascal, 67081 Strasbourg, France. E-mail: egimenez@unistra.fr

<sup>b</sup> Firmenich SA, Corporate R&D Division, 1 Route des Jeunes, 1211 Geneva 8, Switzerland

### **Table of contents**

NMR data compounds 5, 6a/6b

Reaction 4/N-Ac-Cys-OMe ESI-MS/MS analysis of *m/z* 346 and *m/z* 364

Reaction 3/N-Ac-Lys-OMe/FeSO<sub>4</sub> LC chromatogram of the crude product

Reaction 3/N-Ac-Tyr-OEt/FeCl<sub>3</sub>

LC chromatogram of the reaction and suggested adduct  $\mathbf{30}$ 

#### NMR data compounds 5, 6a/6b

In the presence of FeCl<sub>3</sub>, two fractions of new compounds could be isolated from the crude product by column chromatography on silica gel (acetone/pentane 1:9, acetone/pentane 2:8) and analyzed by NMR. One corresponded to the cysteine disulfide dimer 5, resulting from its oxidation. HSQC experiments showed two protons at 3.19 ppm correlated to a carbon atom at 40.7 ppm,  $\delta$  characteristic of the –CH<sub>2</sub> in position  $\alpha$  to the disulfide bond. NMR analysis of the second fraction suggested the formation of two diastereomers. HSOC data revealed carbon atoms at 110.5/110.1 ppm, directly correlated to protons at 4.74/4.78 ppm, proving that the vinyl group of **3** had been preserved during the reaction. HMBC experiments showed long-range correlations between protons at 4.74/4.78 ppm and a carbon atom at 20.5 ppm bearing protons at 1.71/1.74 ppm, corroborating the existence of the isoprenyl group. HSQC and HMBC data also highlighted a correlation between a carbon atom at 125.5 ppm and a proton at 5.53 ppm, this one long-range correlated with a methyl group at 21.0/1.76 ppm and a carbon atom at 68.5 ppm with an associated proton at 3.98 ppm. These  $\delta$  suggested the replacement of the initial hydroperoxide methylallyl group of **3** by a secondary alcohol methylallyl function, confirming the preservation of the menthene ring. Finally, both diastereomers had a N-Ac-Cys-OMe unit linked as the characteristic  $\delta$  of the –CH<sub>2</sub> in position  $\alpha$  to the thiol group of the side-chain at 27.2 ppm shifted to 34.9 ppm with protons at 2.96/3.10 ppm. The whole NMR information agreed with regioisomers 6a and 6b, for which two diasteromers are possible.



Combination of NMR (HSQC direct and HMBC long-range correlations) and MS data suggesting structures 5 and 6a/6b.

NMR data did not allow distinguishing between regioisomers **6a** and **6b**. Theoretically, <sup>1</sup>H and <sup>13</sup>C NMR chemical shifts of the isoprenyl group and of the sulphur atom fixing position should be slightly different. However, no significant signal was observed discriminating between both derivatives.



Simulation of NMR chemical shifts of 6a and 6b (CS Chem Draw Ultra software, version 12.0)

### Reaction 4/N-Ac-Cys-OMe

(a) ESI-MS/MS analysis of m/z 346; (b) ESI-MS/MS analysis of m/z 364



Reaction 3/N-Ac-Lys-OMe/FeSO<sub>4</sub>

LC chromatogram of the crude product



## Reaction 3/N-Ac-Tyr-OEt/FeCl<sub>3</sub>



