

Supplementary Data

For the article

An unexpected increase of toxicity of amino acid-containing ionic liquids

Ksenia S. Egorova,^{a*} Marina M. Seitkhalieva,^a Alexandra V. Posvyatenko^{b,c}

and Valentine P. Ananikov^{a*}

^a N.D. Zelinsky Institute of Organic Chemistry, Russian Academy of Sciences, Leninsky prospect 47, Moscow, 119991 Russia. * E-mail: egorova-ks@ioc.ac.ru, val@ioc.ac.ru

^b Institute of Gene Biology, Russian Academy of Sciences, Vavilova str. 34/5, Moscow, 119334 Russia.

^c D. Rogachev Federal Scientific Clinical Centre of Pediatric Hematology, Oncology and Immunology, Ministry of Health of Russian Federation, Samory Mashela str., Moscow, 117198 Russia.

1-Butyl-3-methylimidazolium glycinate ([BMIM][Gly])

Obtained 0.160 g (75%) light yellow oil.

¹H NMR (600 MHz, dms_o-d₆) δ, ppm: 0.89 (t, J=7.3 Hz, 3H), 1.25 (sext, J=7.3 Hz, 2H), 1.76 (pent, J=7.3 Hz, 2H), 2.69 (s, 2H), 3.87 (s, 3H), 4.19 (t, J=7.2 Hz, 2H), 7.77 (s, 1H), 7.83 (s, 1H), 9.75 (s, 1H); NH₂ may appear as very broad signal around 1.5 ppm.

¹³C{¹H} NMR (150.9 MHz, dms_o-d₆) δ, ppm: 13.2; 18.7; 31.4; 35.5; 46.6; 48.3; 122.2; 123.5; 137.6; 175.3.

Confirmed according to the literature data.¹

HRMS (ESI): found m/z 139.1232; calculated for cation C₈H₁₅N₂ m/z 139.1230 (Δ=1.4 ppm).

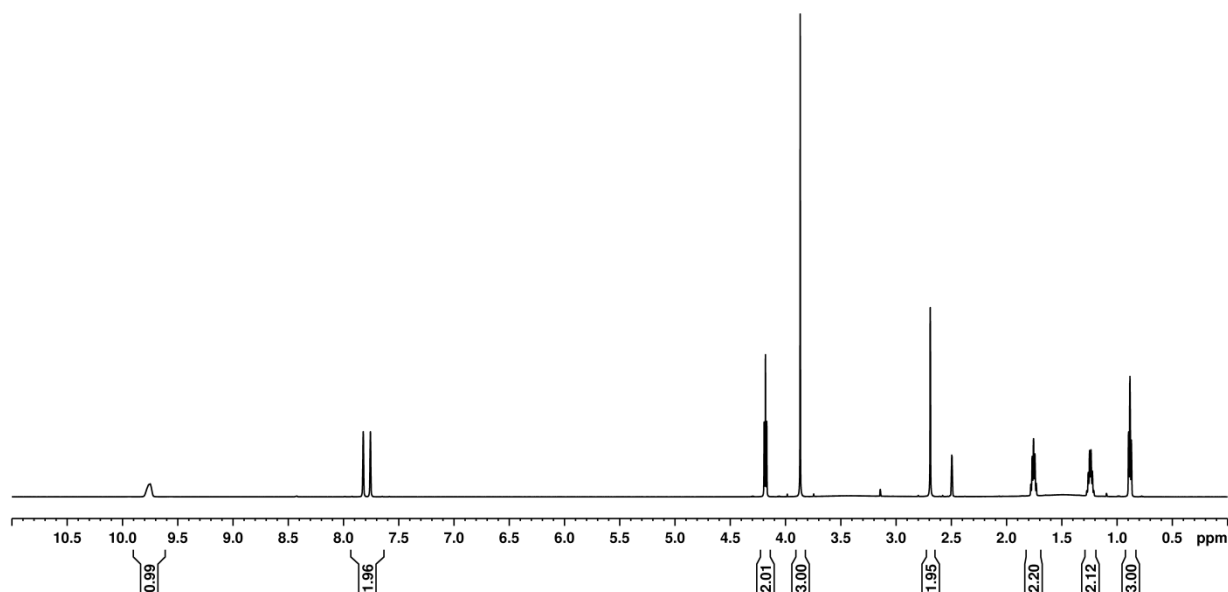


Figure S1. ¹H NMR spectrum of [BMIM][Gly].

¹ Li, M.; Wang, T.; Pham, P. J.; Pittman, C. U.; Li, T. *Sep. Sci. Technol.*, 2008, **43**, 828.

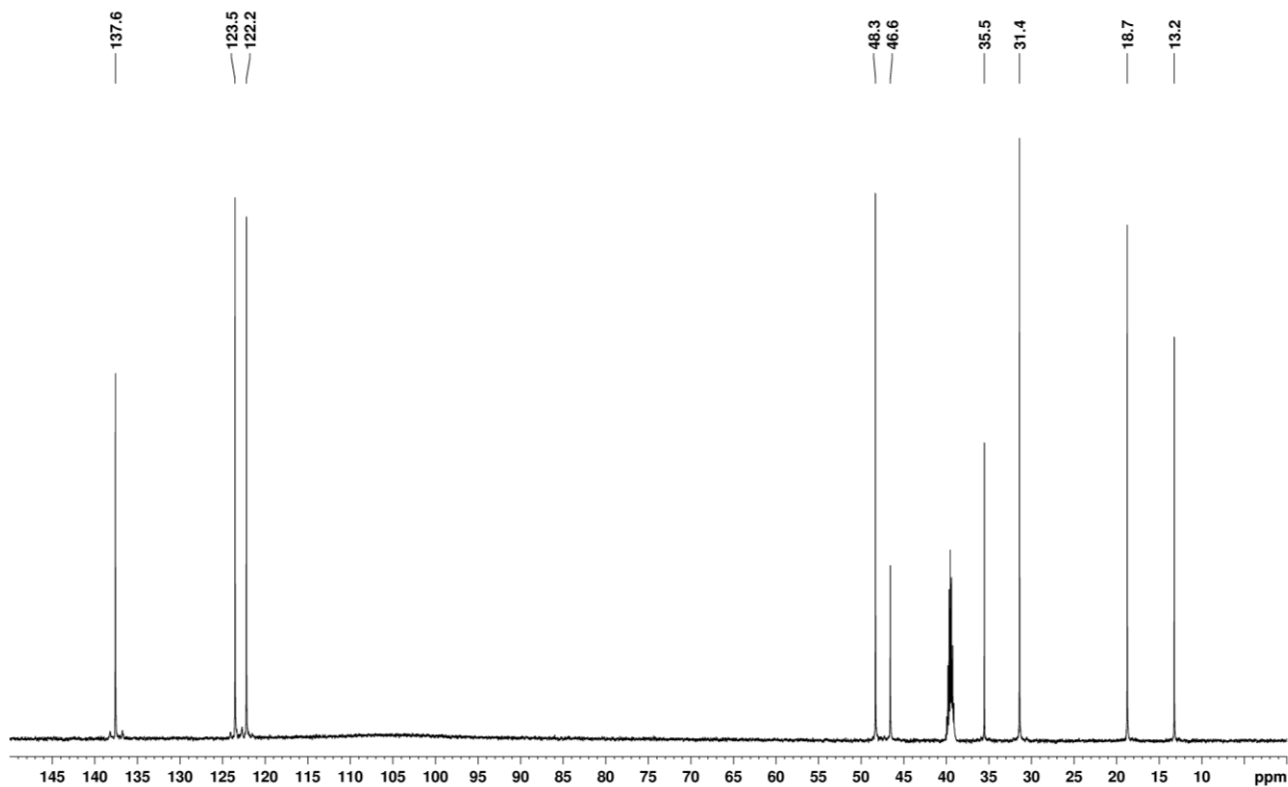


Figure S2. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of [BMIM][Gly].

1-Butyl-3-methylimidazolium alaninate ([BMIM][Ala])

Obtained 0.177 g (78%) light yellow oil.

^1H NMR (600 MHz, $\text{dms}\text{-}d_6$) δ , ppm: 0.89 (t, $J=7.4$ Hz, 3H), 1.01 (d, $J=6.8$ Hz, 3H), 1.24 (sext, $J=7.5$ Hz, 2H), 1.76 (pent, $J=7.5$ Hz, 2H), 2.84 (q, $J=6.7$ Hz, 1H), 3.87 (s, 3H), 4.19 (t, $J=7.2$ Hz, 2H), 7.76 (s, 1H), 7.82 (s, 1H), 9.81 (s, 1H); NH_2 may appear as very broad signal around 1.5 ppm.

$^{13}\text{C}\{^1\text{H}\}$ NMR (150.9 MHz, $\text{dms}\text{-}d_6$) δ , ppm: 13.2; 18.7; 22.9; 31.4; 35.5; 48.3; 51.8; 122.2; 123.5; 137.4; 177.9.

Confirmed according to the literature data.²

HRMS (ESI): found m/z 139.1233; calculated for cation $\text{C}_8\text{H}_{15}\text{N}_2$ m/z 139.1230 ($\Delta=2.2$ ppm).

² Fang, D. W.; Guan, W.; Tong, J.; Wang, Z. W.; Yang, J. Z. *J. Phys. Chem., B* 2008, **112**, 7499.

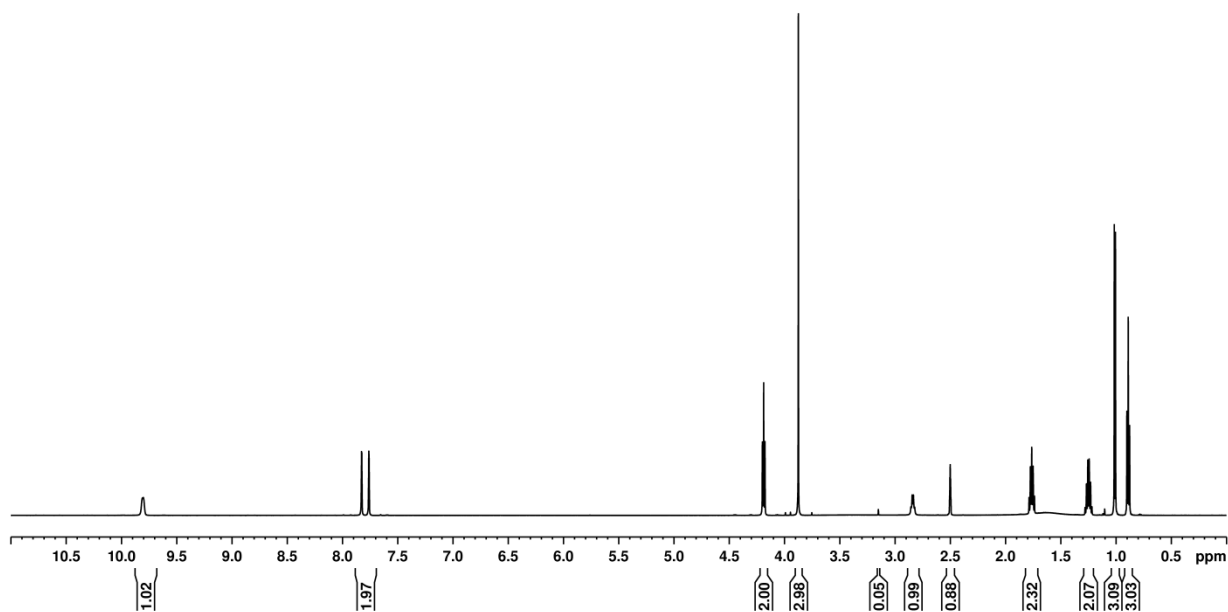


Figure S3. ^1H NMR spectrum of [BMIM][Ala].

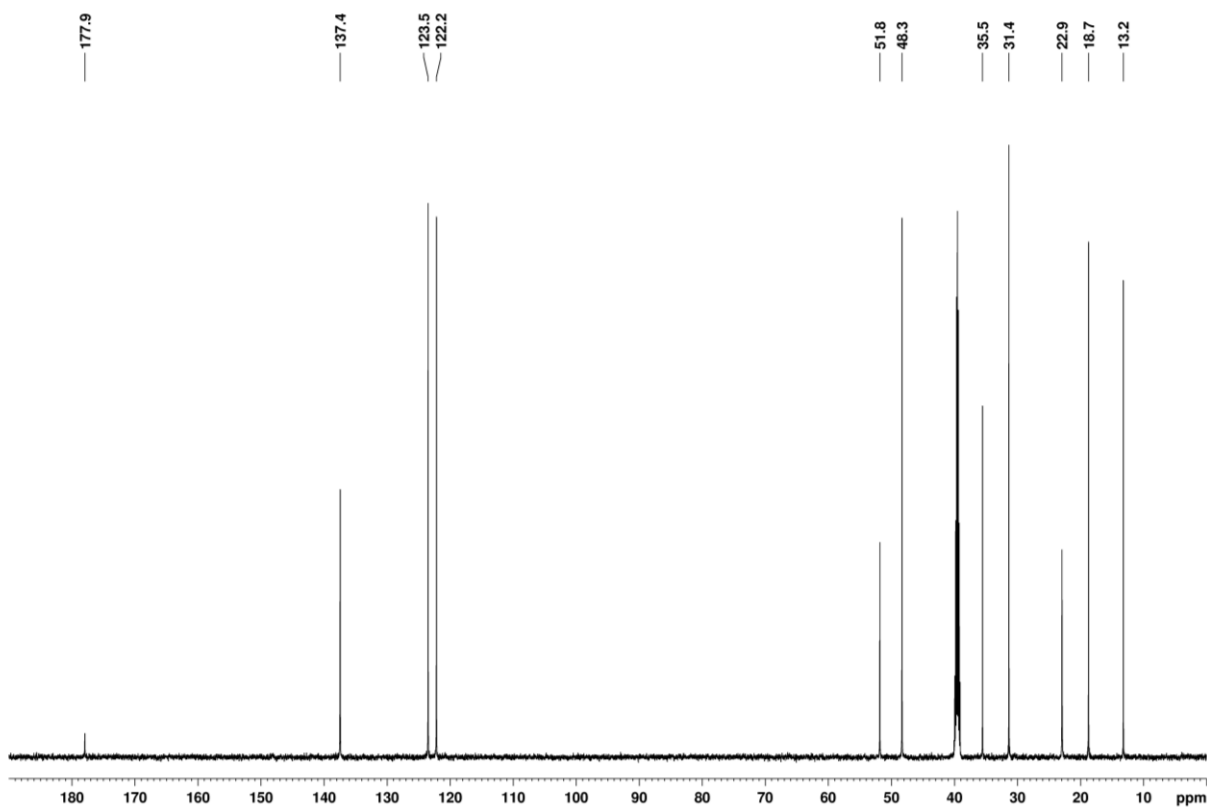


Figure S4. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum [BMIM][Ala].

1-Butyl-3-methylimidazolium valinate ([BMIM][Val])

Obtained 0.184 g (72%) light yellow oil.

^1H NMR (500 MHz, $\text{dms}\text{-d}_6$) δ , ppm: 0.68 (d, $J=6.7$ Hz, 3H), 0.8 (d, $J=6.7$ Hz, 3H), 0.89 (t, $J=7.4$ Hz, 3H), 1.24 (sext, $J=7.5$ Hz, 2H), 1.76 (pent, $J=7.3$ Hz, 2H), 1.89 (m, $J=6.7$ Hz, 3.9 Hz, 1H), 2.64 (d, $J=3.9$ Hz, 1H), 3.88 (s, 3H), 4.19 (t, $J=7.2$ Hz, 2H), 7.78 (s, 1H), 7.85 (s, 1H), 9.93 (s, 1H); NH_2 may appear as very broad signal around 1.5 ppm.

$^{13}\text{C}\{^1\text{H}\}$ NMR (150.9 MHz, $\text{dms}\text{-d}_6$) δ , ppm: 13.2; 17.0; 18.7; 20.6; 31.3; 31.6; 35.6; 48.3; 61.5; 122.2; 123.5; 137.2; 176.5.

HRMS (ESI): found m/z 139.1232; calculated for cation $\text{C}_8\text{H}_{15}\text{N}_2$ m/z 139.1230 ($\Delta=1.4$ ppm).

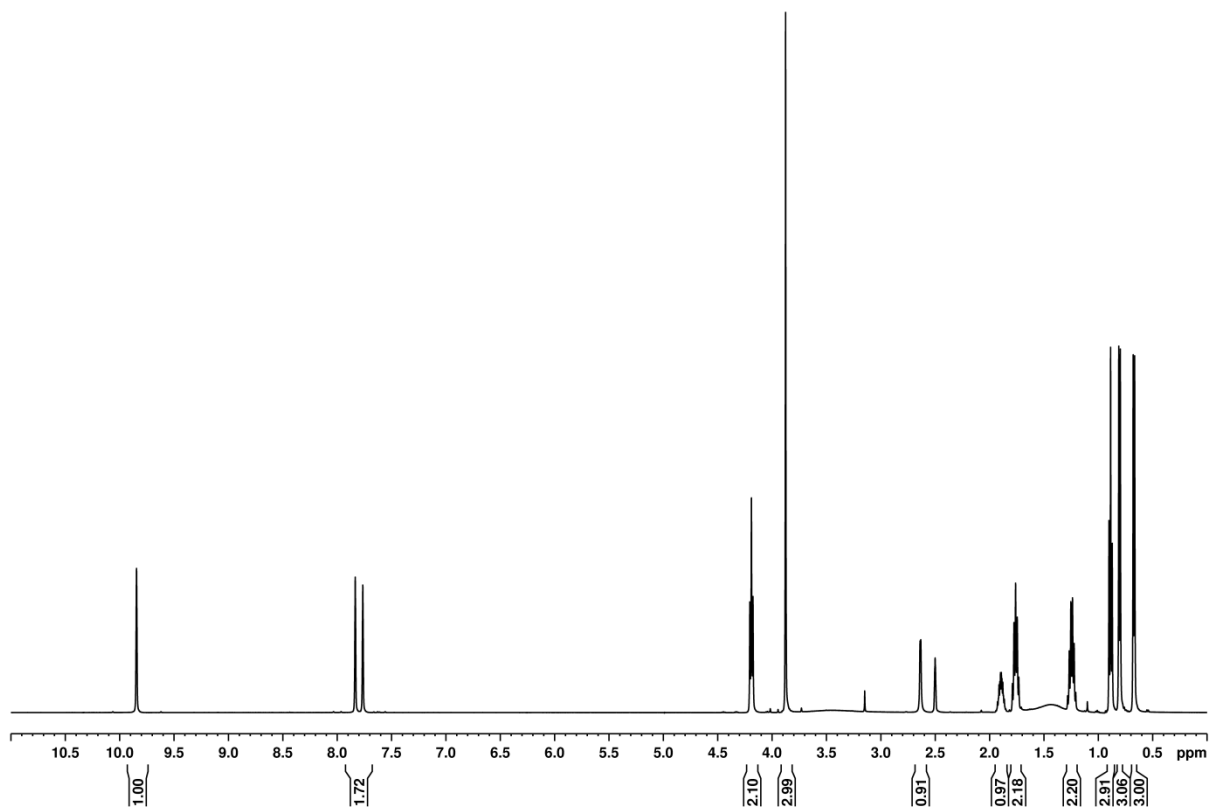


Figure S5. ^1H NMR spectrum of [BMIM][Val].

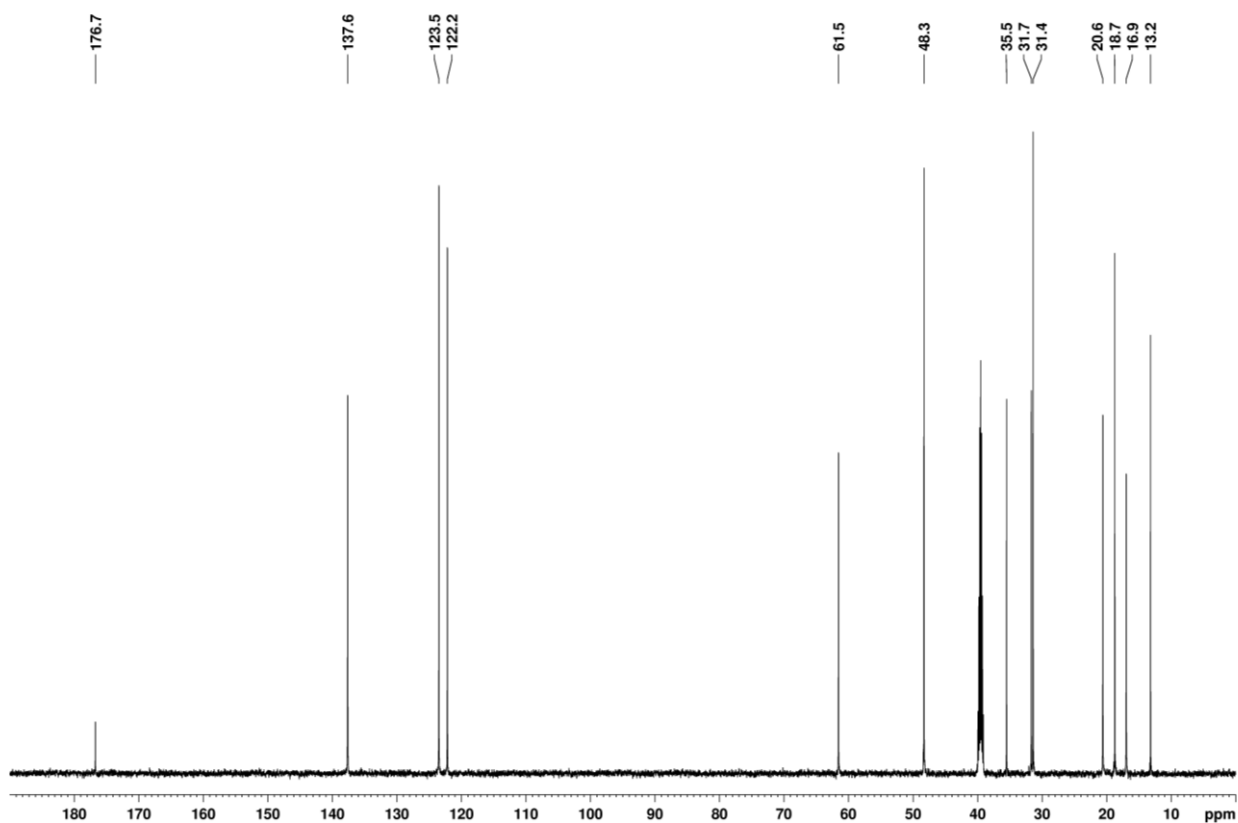


Figure S6. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of [BMIM][Val].

O-Methylglycinate tetrafluoroborate ([Gly-OMe][BF₄])

Obtained 0.167 g (94%) light yellow oil, which crystallizes instantly.

^1H NMR (500 MHz, dms o -d $_6$) δ , ppm: 3.74 (s, 3H), 3.82 (s, 2H), 8.08 (br. s, NH $_3$)

$^{13}\text{C}\{^1\text{H}\}$ NMR (150.9 MHz, dms o -d $_6$) δ , ppm: 39.7, 52.6, 168.3

HRMS (ESI): found m/z 90.0556; calculated for cation C $_3$ H $_8$ NO $_2$ m/z 90.0550 (Δ =6.7 ppm).

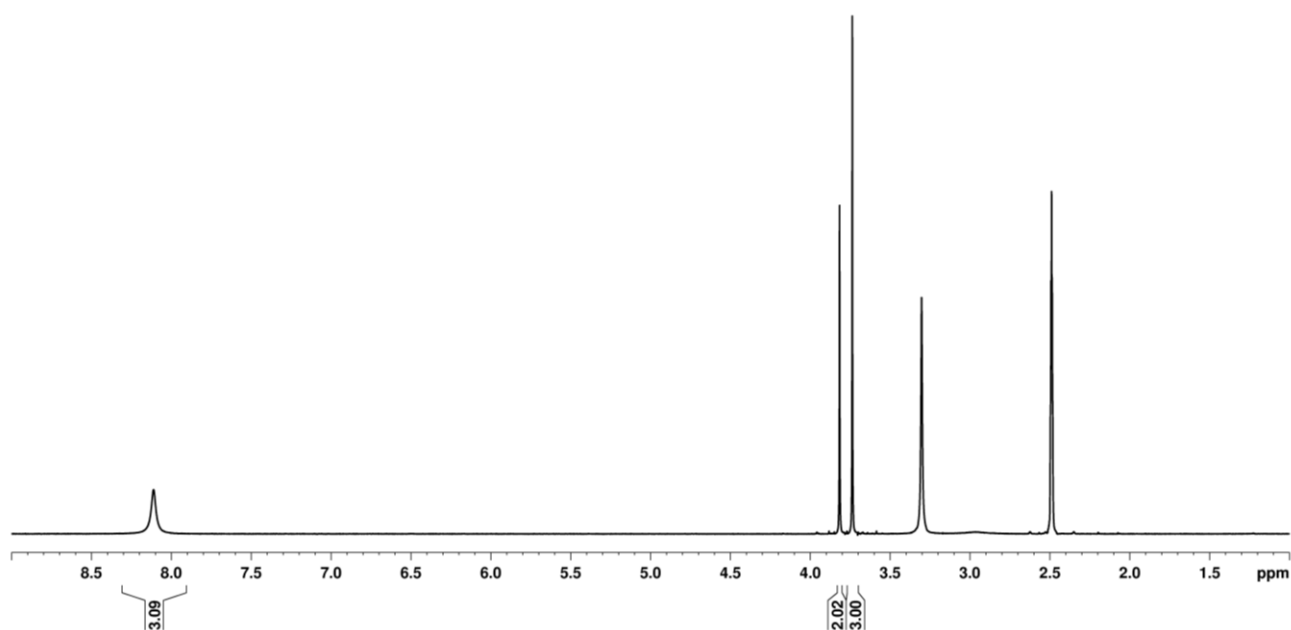


Figure S7. ^1H NMR spectrum of [Gly-OMe][BF₄].

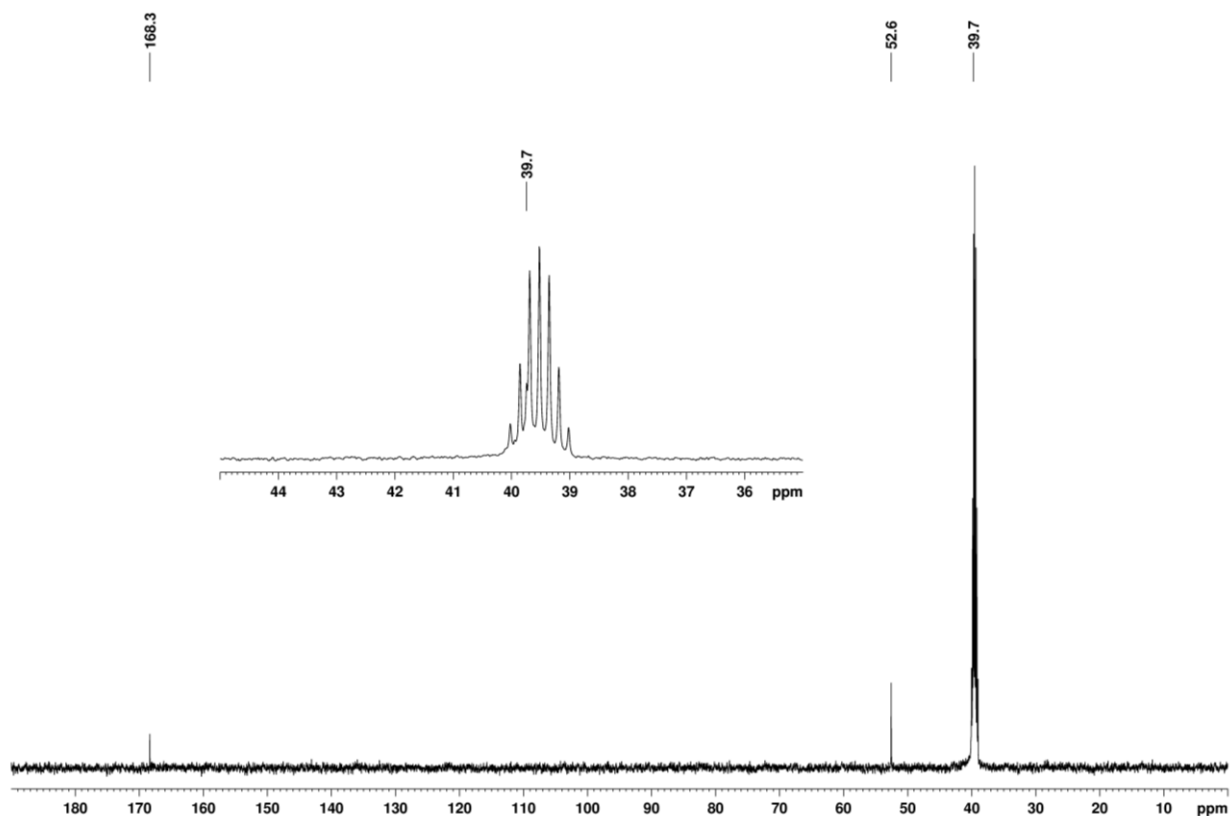


Figure S8. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of [Gly-OMe][BF₄].

O-Methylalaninate tetrafluoroborate ([Ala-OMe][BF₄])

Obtained 0.176 g. (92%) light yellow oil, which crystallizes instantly.

^1H NMR (500 MHz, dms o -d $_6$) δ , ppm: 1.38 (d, J =7.2 Hz, 3H), 3.75 (s, 3H), 4.10 (q, J =7.2 Hz, 1H), 8.20 (br. s, 3H).

$^{13}\text{C}\{^1\text{H}\}$ NMR (128.5 MHz, dms o -d $_6$) δ , ppm 15.8, 48.0, 53.0, 170.5.

HRMS (ESI): found m/z 104.0709; calculated for cation C₄H₁₀NO₂ m/z 104.0706 (Δ =2.9 ppm).

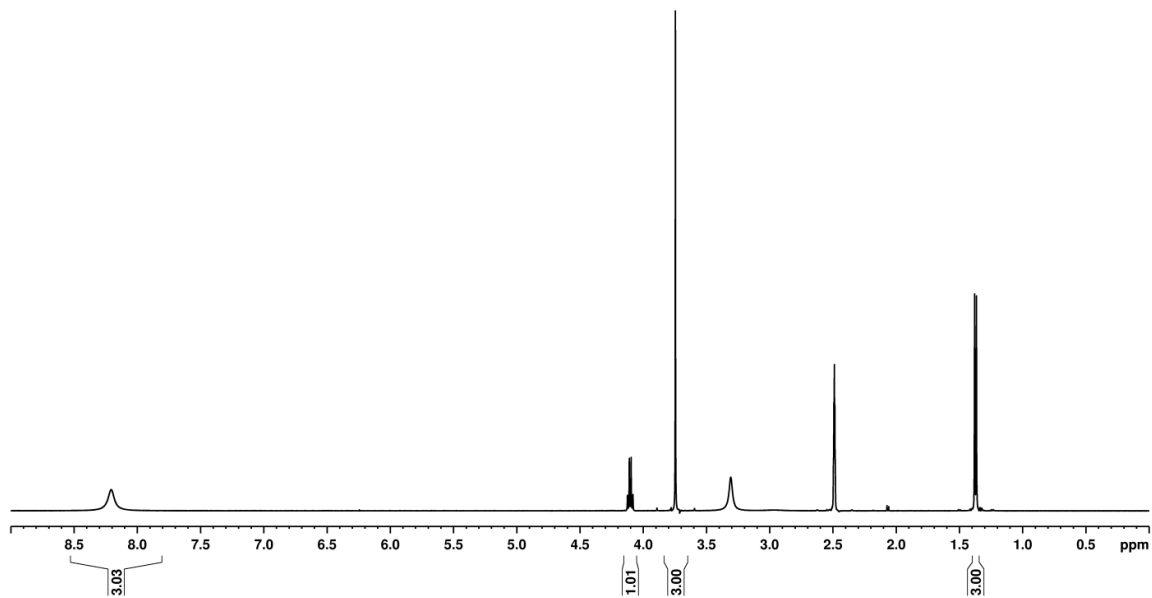


Figure S9. ^1H NMR spectrum of [Ala-OMe][BF₄].

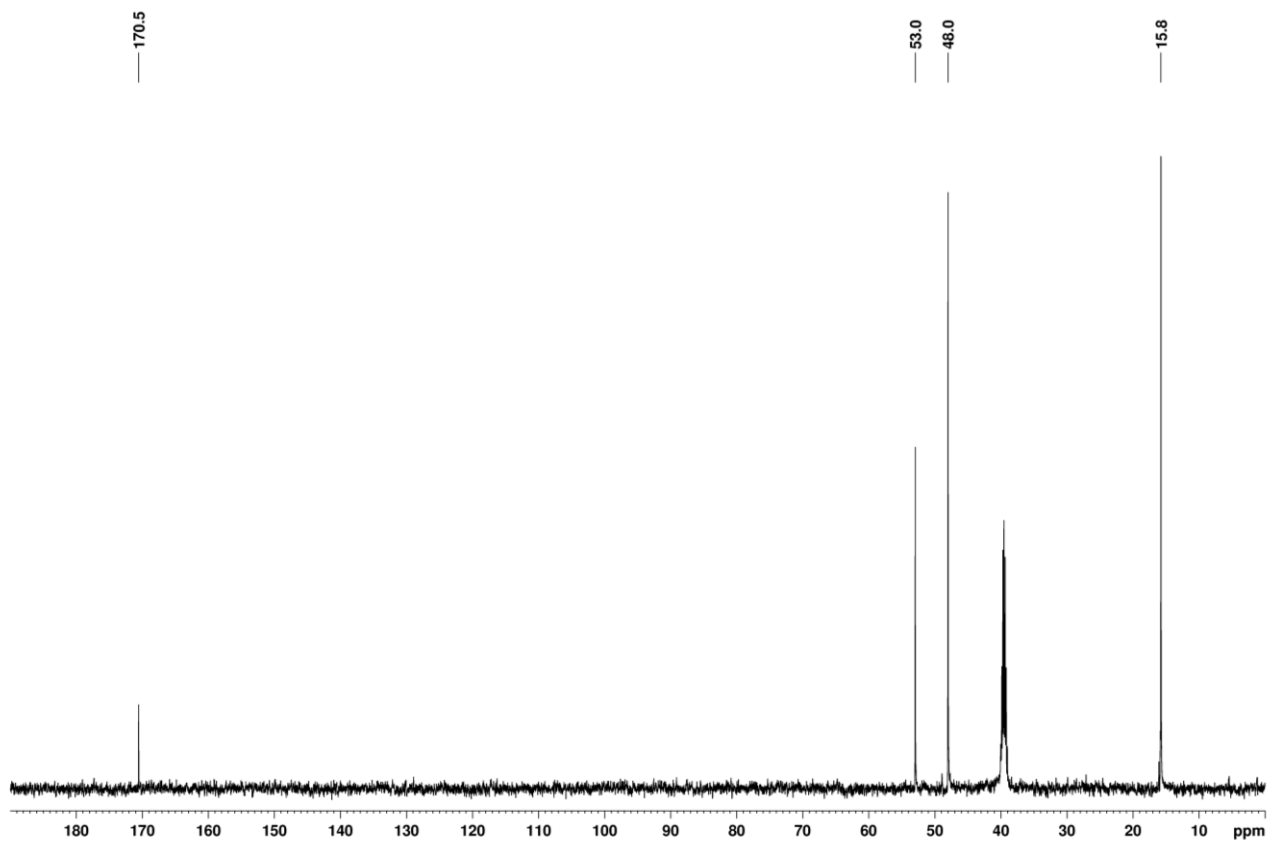


Figure S10. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of [Ala-OMe][BF₄].

O-Methylvalinate tetrafluoroborate ([Val-OMe][BF₄])

Obtained 0.199 g (91 %) light oil, which crystallizes instantly.

¹H NMR (500 MHz, dms_o-d₆) δ, ppm: 0.94 (d, J=6.7 Hz, 3H), 0.97 (d, J=6.7 Hz, 3H), 2.14 (dq, J=6.7 Hz, 4.8 Hz, 1H), 3.77 (s, 3H), 3.93 (d, J=4.8, 1H), 8.22 (br. s, 3H).

¹³C{¹H} NMR (128.5 MHz, dms_o-d₆) δ, ppm: 17.6, 18.1, 29.4, 52.7, 57.4, 169.5.

HRMS (ESI): found m/z 132.1021; calculated for cation C₆H₁₄NO₂ m/z 132.1019 (Δ=1.5 ppm).

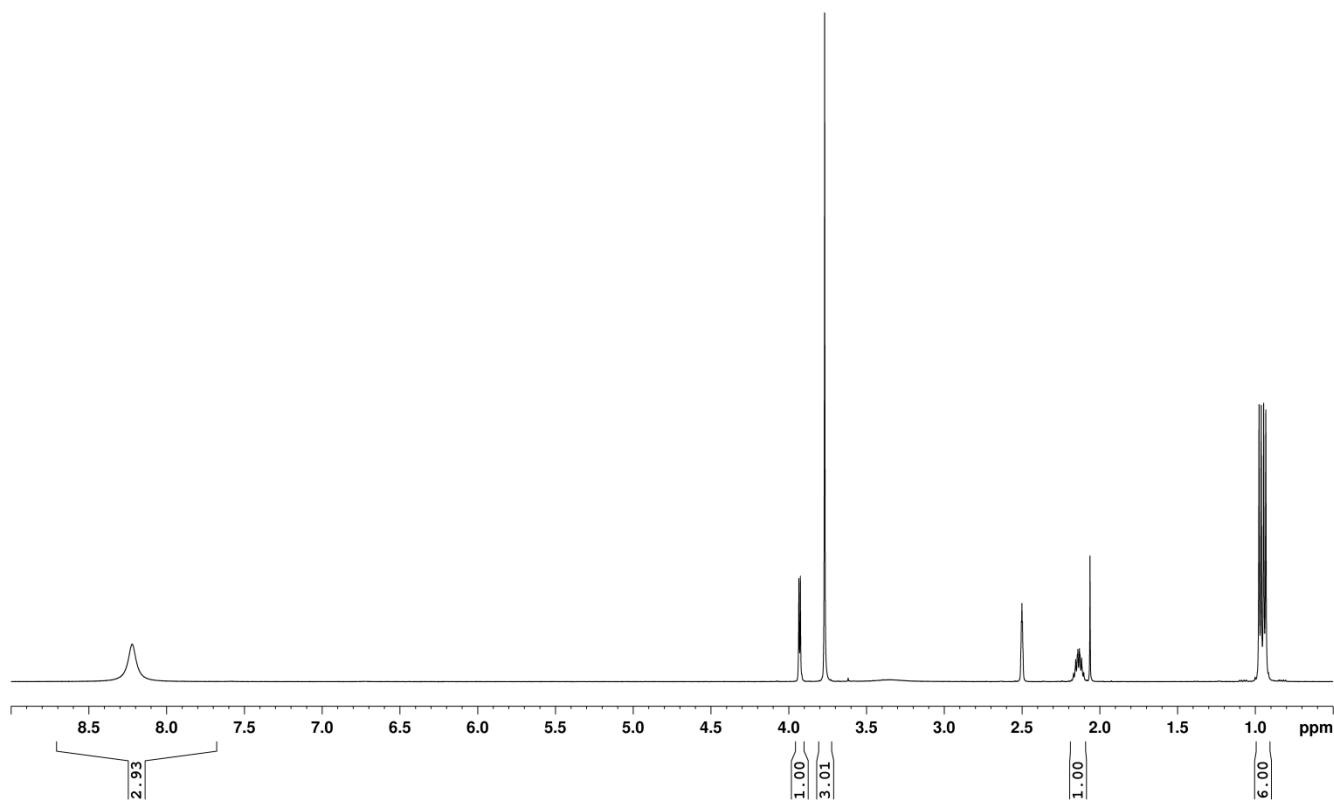


Figure S11. ¹H NMR spectrum of [Val-OMe][BF₄]. (residual organic solvent appears as a singlet at 2 ppm)

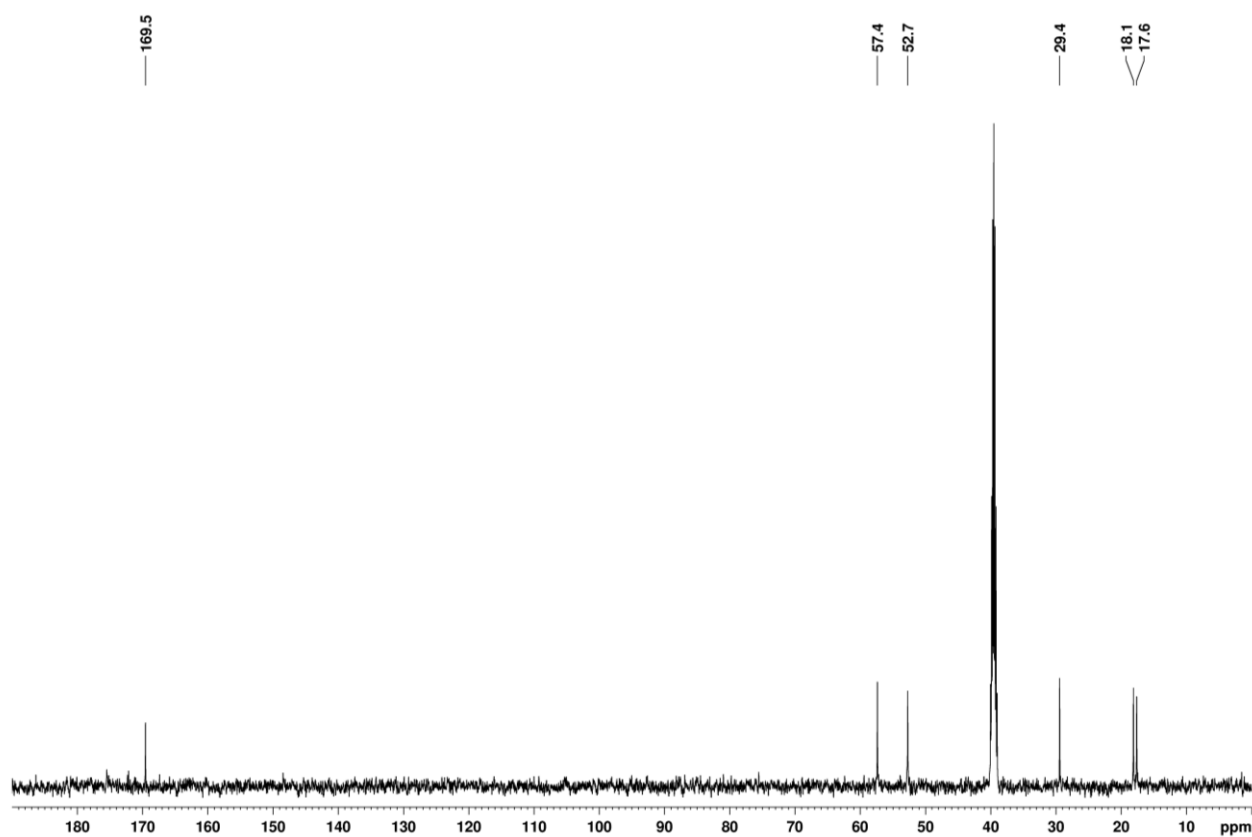


Figure S12. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of [Val-OMe][BF₄].

1-(2-(Glycyloxy)ethyl)-3-methylimidazolium tetrafluoroborate ([GlyO-EMIM][BF₄])

Obtained 0.493 g. (91%) beige powder

^1H NMR (500 MHz, dms o -d₆) δ , ppm: 3.83 (s, 2H), 3.88 (s, 3H), 4.50 (s, 4H), 7.73 (s, 1H), 7.83 (s, 1H), 8.48 (br. s, 2H), 9.28 (s, 1H).

$^{13}\text{C}\{^1\text{H}\}$ NMR (125.8 MHz, dms o -d₆) δ , ppm: 35.9; 39.7; 47.8; 63.5; 122.7; 123.6; 137.3; 167.3.

Confirmed according to the literature data.³

HRMS (ESI): found m/z 184.1082; calculated for cation C₈H₁₄N₃O₂ m/z 184.1081 ($\Delta=0.5$ ppm).

³ Debdab, M.; Mongin, F.; Bazureau, J.P. *Synthesis*, 2006, **23**, 4046.

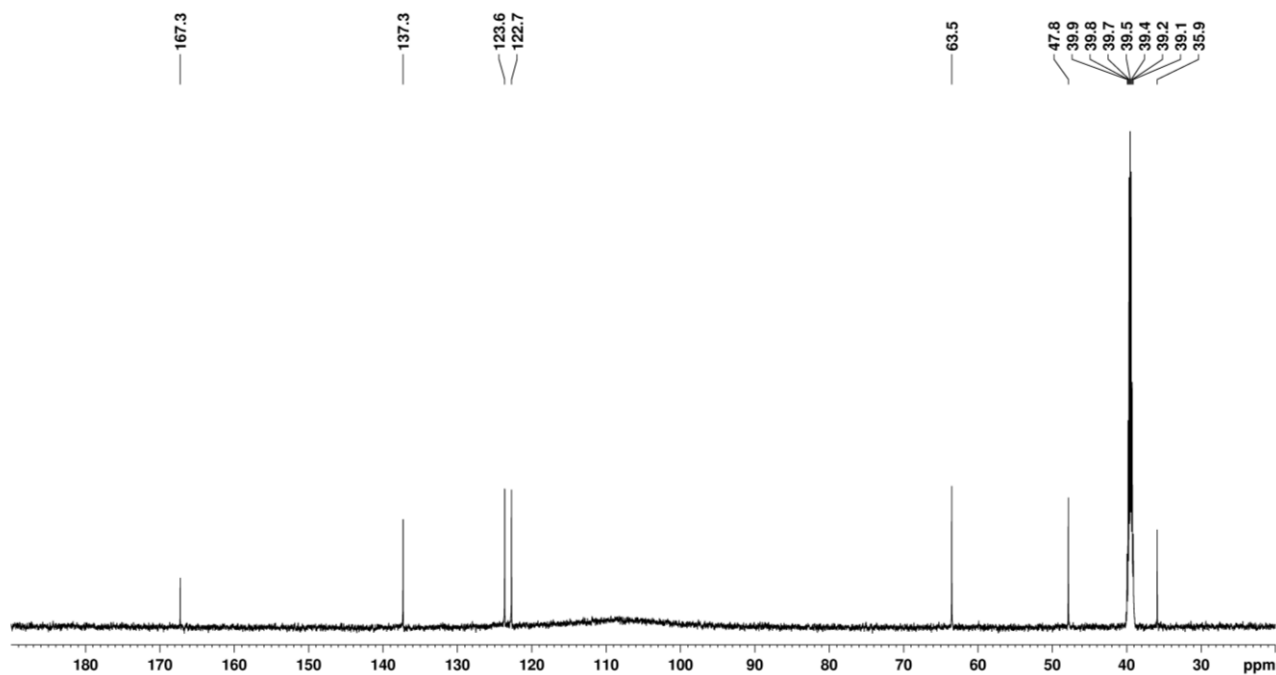


Figure S13. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of [GlyO-EMIM][BF₄].

1-(2-Hydroxyethyl)-3-methylimidazolium tetrafluoroborate ([EMIMOH][BF₄])

Obtained 0.393 g (92%) yellow oil.

^1H NMR (500 MHz, dms o -d $_6$) δ , ppm: 3.72 (q, J=5 Hz, 2H), 3.86 (s, 3H), 4.21 (t, J=5 Hz, 2H), 5.13 (t, J=5 Hz, 1H), 7.67 (s, 1H), 7.70 (s, 1H), 9.06 (s, 1H).

$^{13}\text{C}\{^1\text{H}\}$ NMR (150.9 MHz, dms o -d $_6$) δ , ppm: 35.7, 51.7, 59.3, 122.7, 123.3, 136.8.

Confirmed according to the literature data.⁴

HRMS (ESI): found m/z 127.0870; calculated for cation C₆H₁₁N₂O m/z 127.0866 (Δ =3.2 ppm).

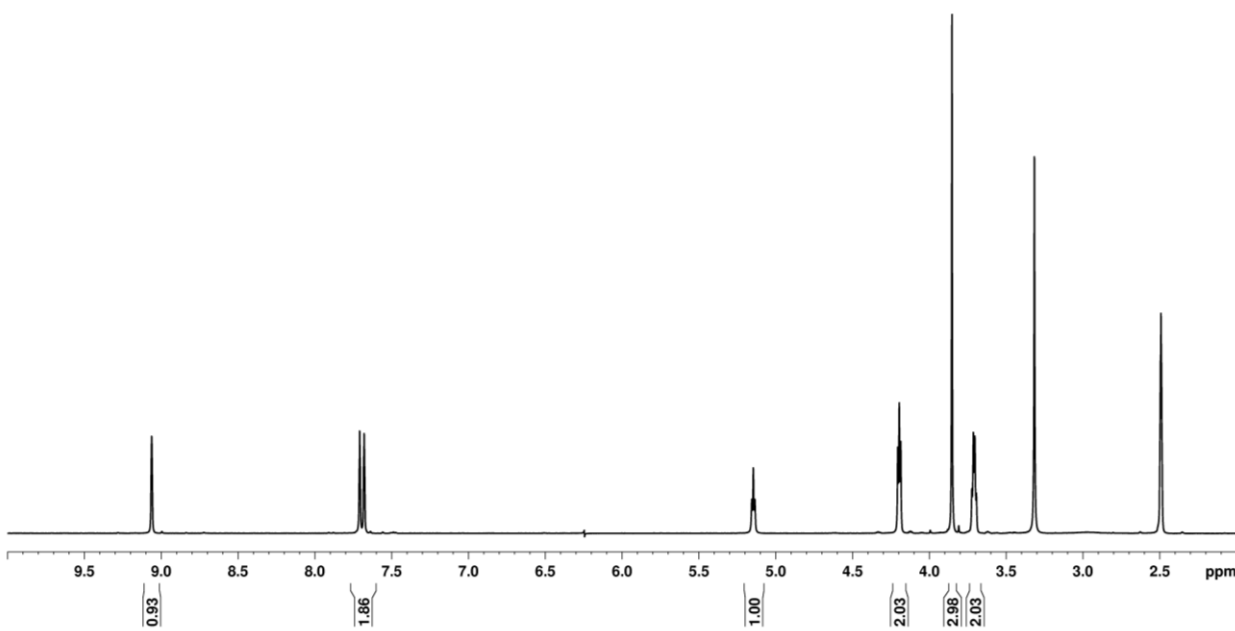


Figure S14. ^1H NMR spectrum of [EMIMOH][BF₄].

⁴ L. C. Branco, J. N. Rosa, J. J. Moura Ramos and C. A. Afonso, *Chem.-Eur. J.*, 2002, **8**, 3671.

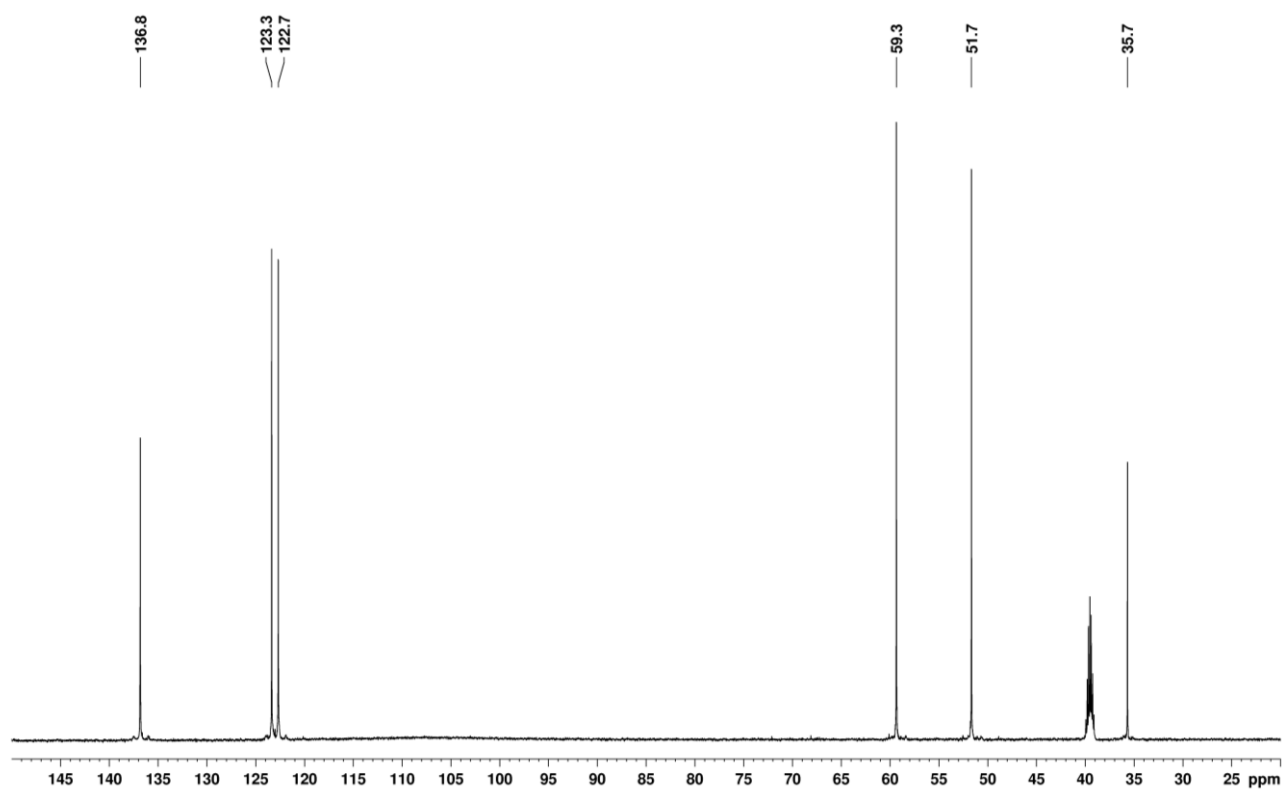


Figure S15. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of [EMIMOH][BF₄].